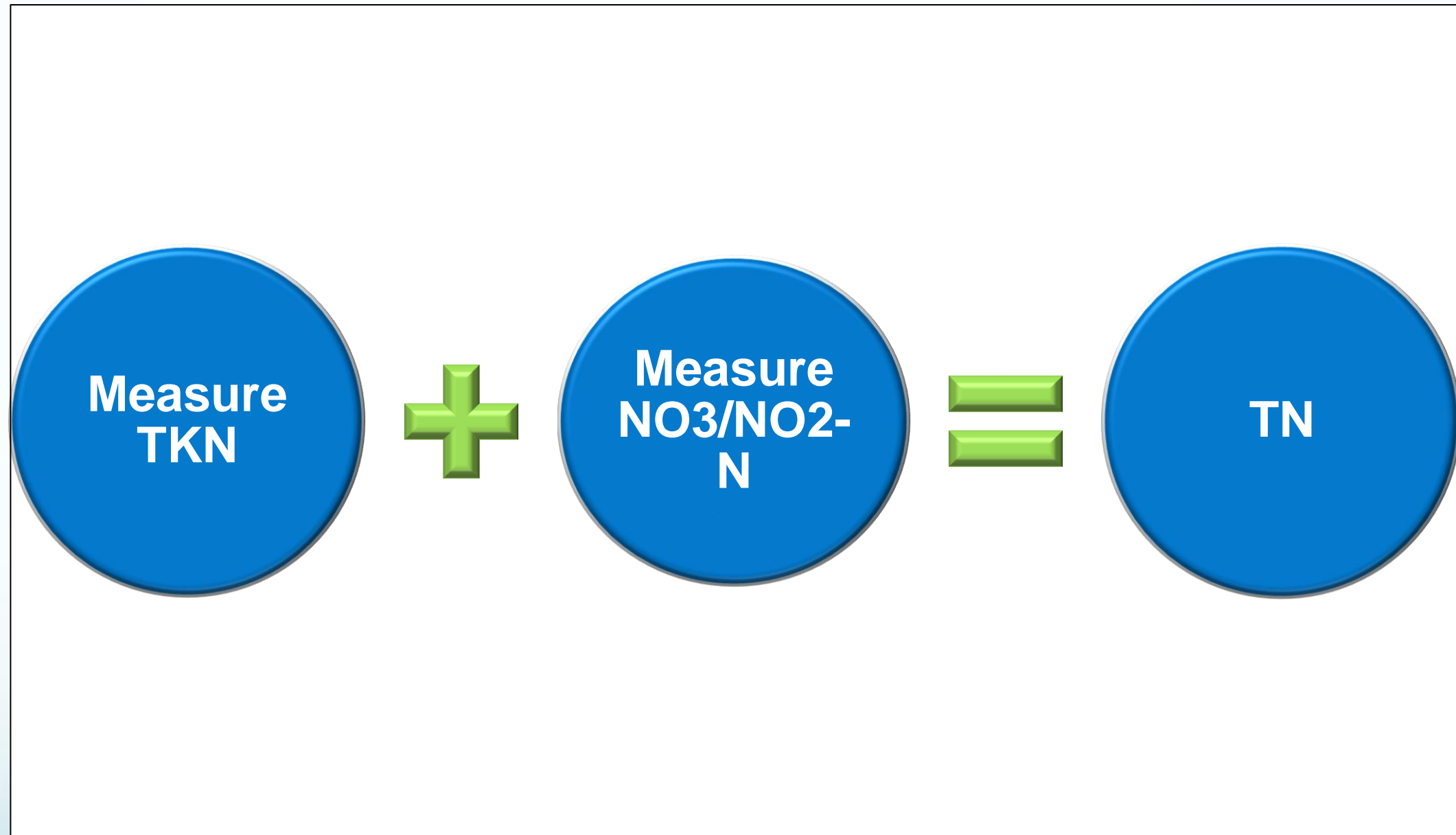


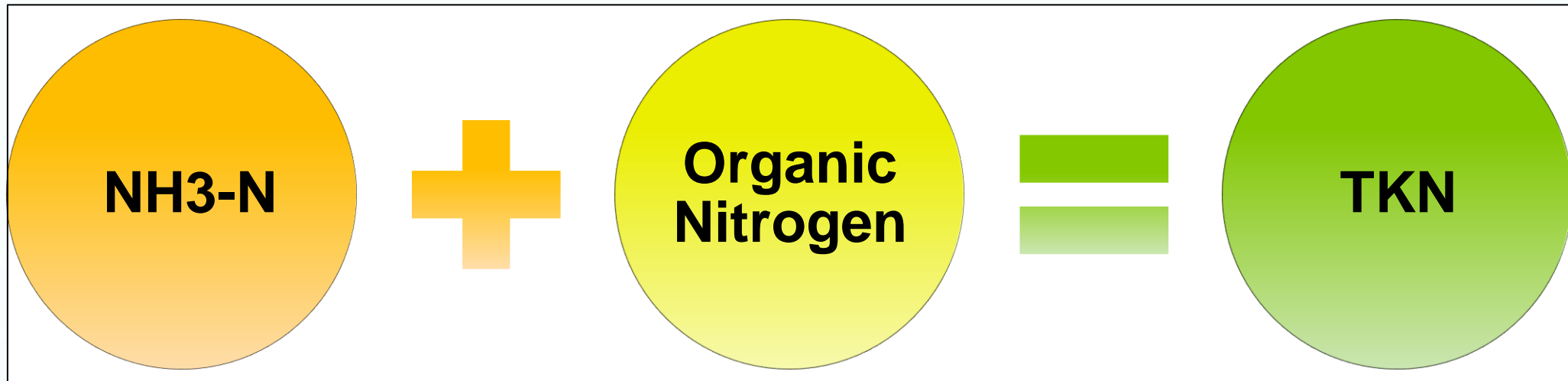
# **Automated Digestion of Total Nitrogen and Total Phosphorus in Environmental Samples**

**William Lipps  
OI Analytical**

# EPA definition of total nitrogen



# Measuring organic nitrogen requires a TKN and ammonia analysis



# TKN is a “classical” analysis of total nitrogen

- **TKN includes**
  - **Organic Nitrogen**
  - **Ammonia Nitrogen ( $\text{NH}_3\text{-N}$ )**
- **TKN does not include**
  - **Nitrate Nitrogen ( $\text{NO}_3\text{-N}$ )**
  - **Nitrite Nitrogen ( $\text{NO}_2\text{-N}$ )**

# Must use 40 CFR Part 136 methods

## TKN Methods

- Manual digestion
- Manual distillation
- Nessler, IC, Titration, ISE, phenate

## TKN Methods

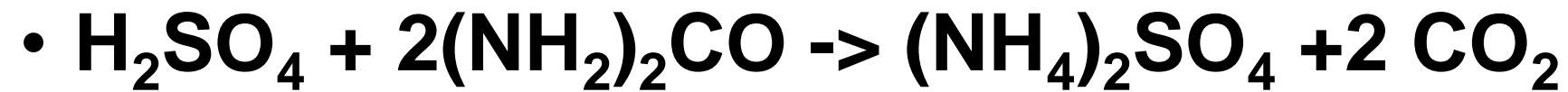
- Manual digestion
- Automated distillation/diffusion
- Direct Colorimetry

## NO<sub>3</sub>/NO<sub>2</sub> –N Methods

- IC, Reduction colorimetry

# The “classical” Kjeldahl Reactions distilled and titrated ammonia

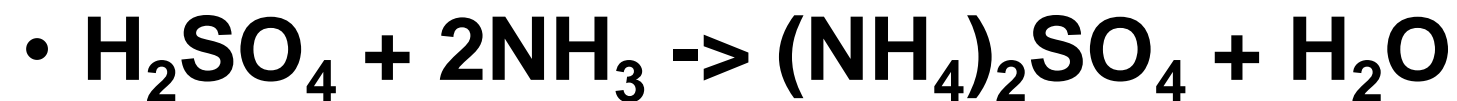
## Digestion



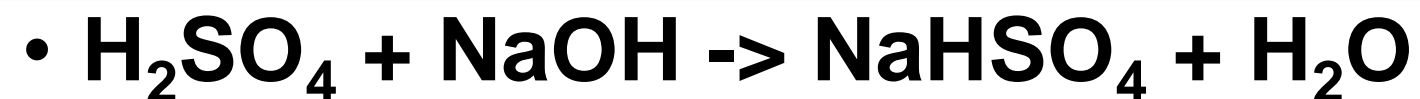
## Distillation



## Capture



## Titration



**Present Day TKN methods can either distill (diffuse) or measure  $\text{NH}_4^+$  in the matrix**

## **Manual Digestion**

- $\text{H}_2\text{SO}_4 + (\text{NH}_2)_2\text{CO} \rightarrow (\text{NH}_4)_2\text{SO}_4 + 2\text{CO}_2$
- **Must use catalyst**

## **Distillation (required for some methods)**

- $(\text{NH}_4)_2\text{SO}_4 + \text{NaOH} \rightarrow \text{NaHSO}_4 + \text{H}_2\text{O} + 2\text{NH}_3$

## **Detection**

- **Titration, ISE, manual colorimetry, automated colorimetry**

# **The block digester methods for TKN do not require distillation**

- **Sodium Citrate is a better complexing agent**
- **Salicylic acid method**
- **Blue color measured at 640 – 660 nm**
- **Continuous Flow or Discrete Analyzer methods.**



# **Digest multiple samples on a block and analyze them automatically**

- **Faster than titration**
- **Results in minutes**
- **Easy repeat of questionable results**
  - **Dilute off scale samples**
    - **No distillation**
  - **Smaller volumes (25 ml)**

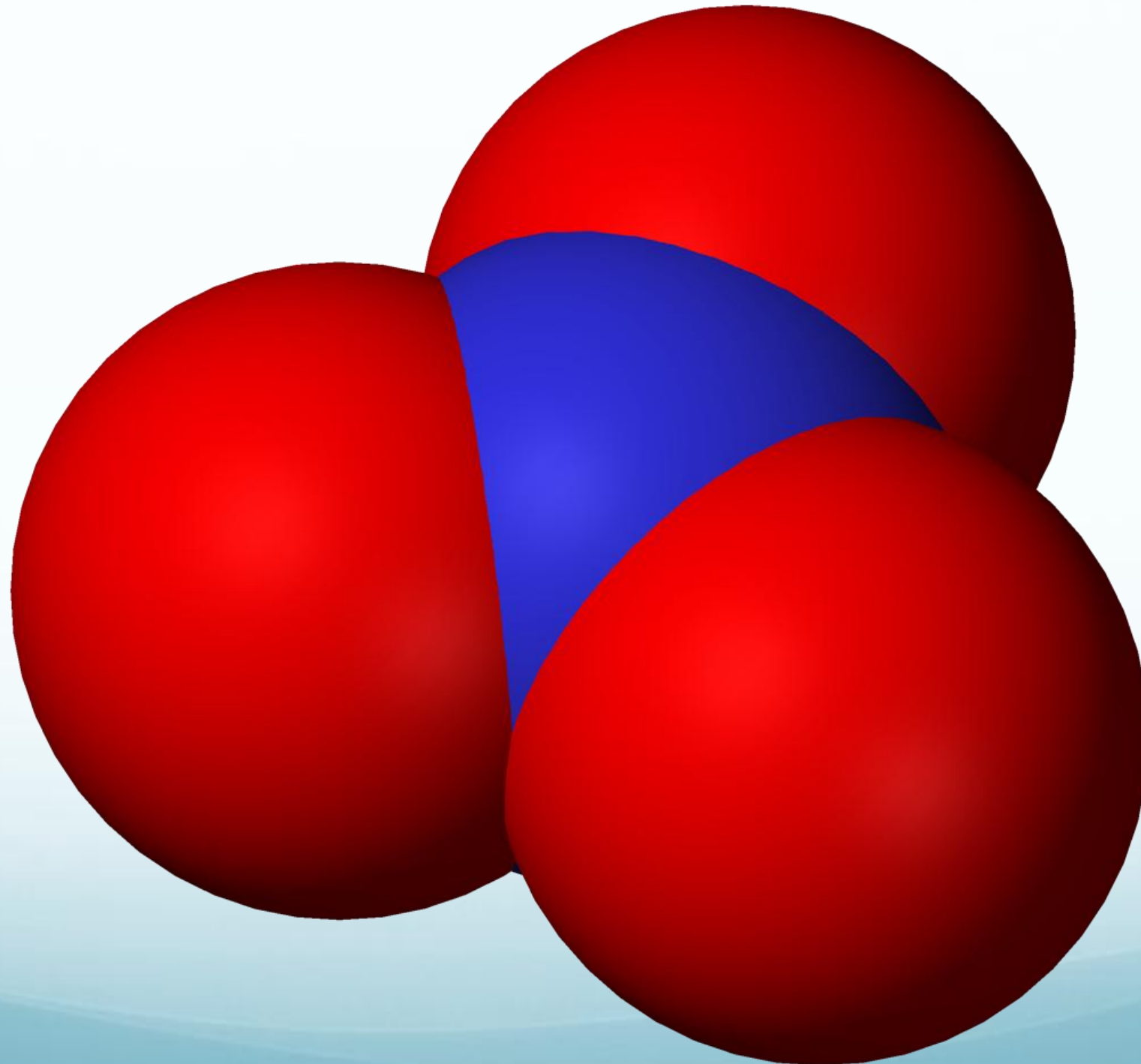
# Operational Comparison of TKN methods

<b>Manual Digestion &amp; Distillation Followed By:</b>	<b>Require final digest volume</b>	<b>Require final distillate volume</b>
<b>Titration</b>	<b>no</b>	<b>no</b>
<b>Ion Selective Electrode</b>	<b>no</b>	<b>yes</b>
<b>Manual Phenate</b>	<b>no</b>	<b>yes</b>
<b>Automated Phenate</b>	<b>no</b>	<b>yes</b>
<b>Block Digestion</b>		
<b>Automated Salicylate</b>	<b>yes</b>	<b>N/A</b>

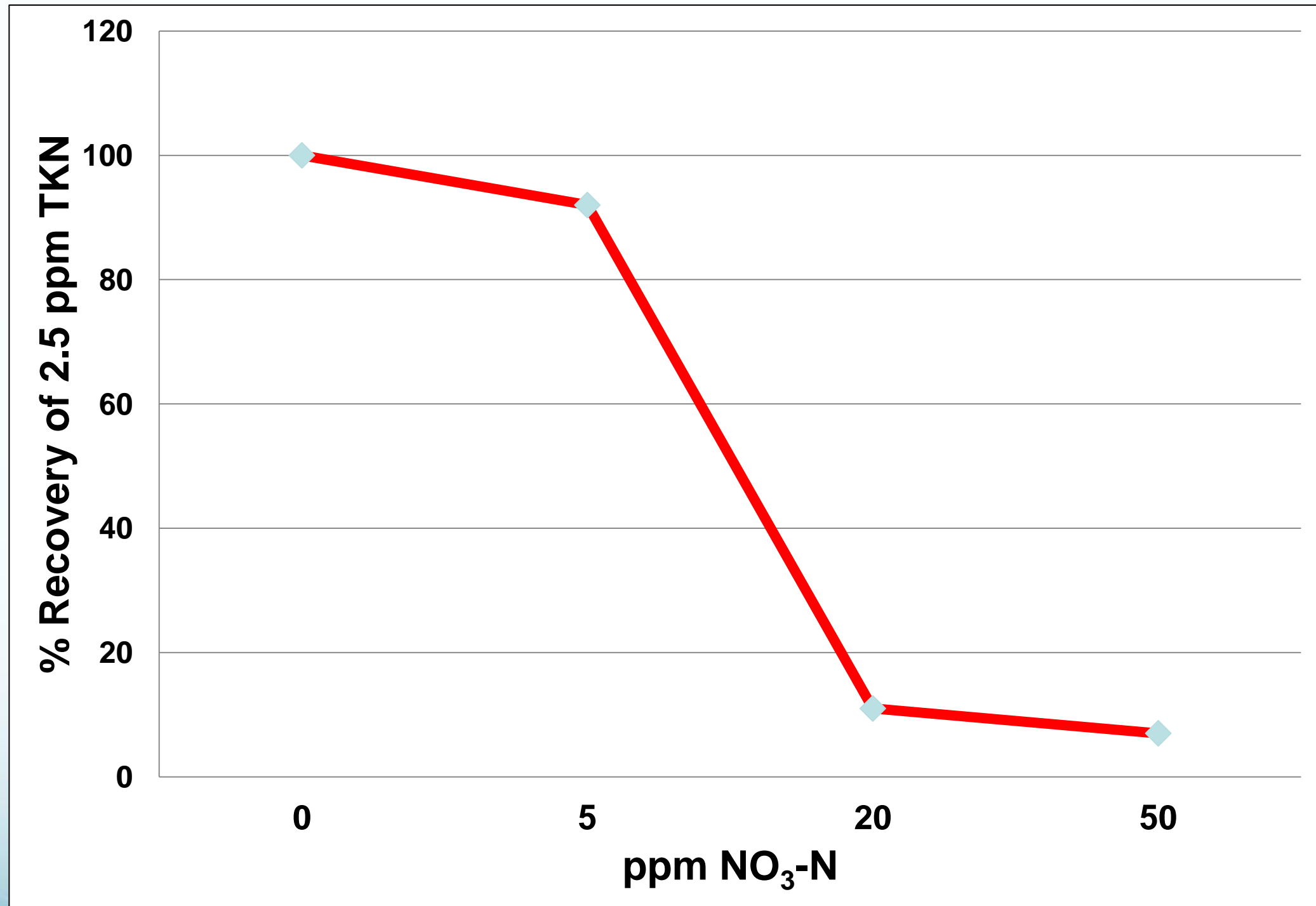
# Techniques for the analysis of Total Nitrogen (EPA only approves TKN)

<b>Manual Digestion &amp; Distillation Followed By:</b>	<b>MDL (mg/L)</b>
<b>Titration</b>	<b>1.0</b>
<b>Manual Phenate</b>	<b>0.05 ?</b>
<b>Ion Selective Electrode</b>	<b>0.03</b>
<b>Automated Phenate</b>	<b>0.01</b>
<b>Block Digestion</b>	
<b>Automated Salicylate</b>	<b>0.1</b>
<b>Auto-GD/pH colorimetry</b>	<b>0.2</b>
<b>Auto-GD/Salicylate</b>	<b>0.02</b>

**Problem -TKN does not measure nitrate, and total nitrogen includes nitrate plus nitrite**



# Low Recovery of TKN in presence of $\text{NO}_3\text{-N}$



Schlueter 1977

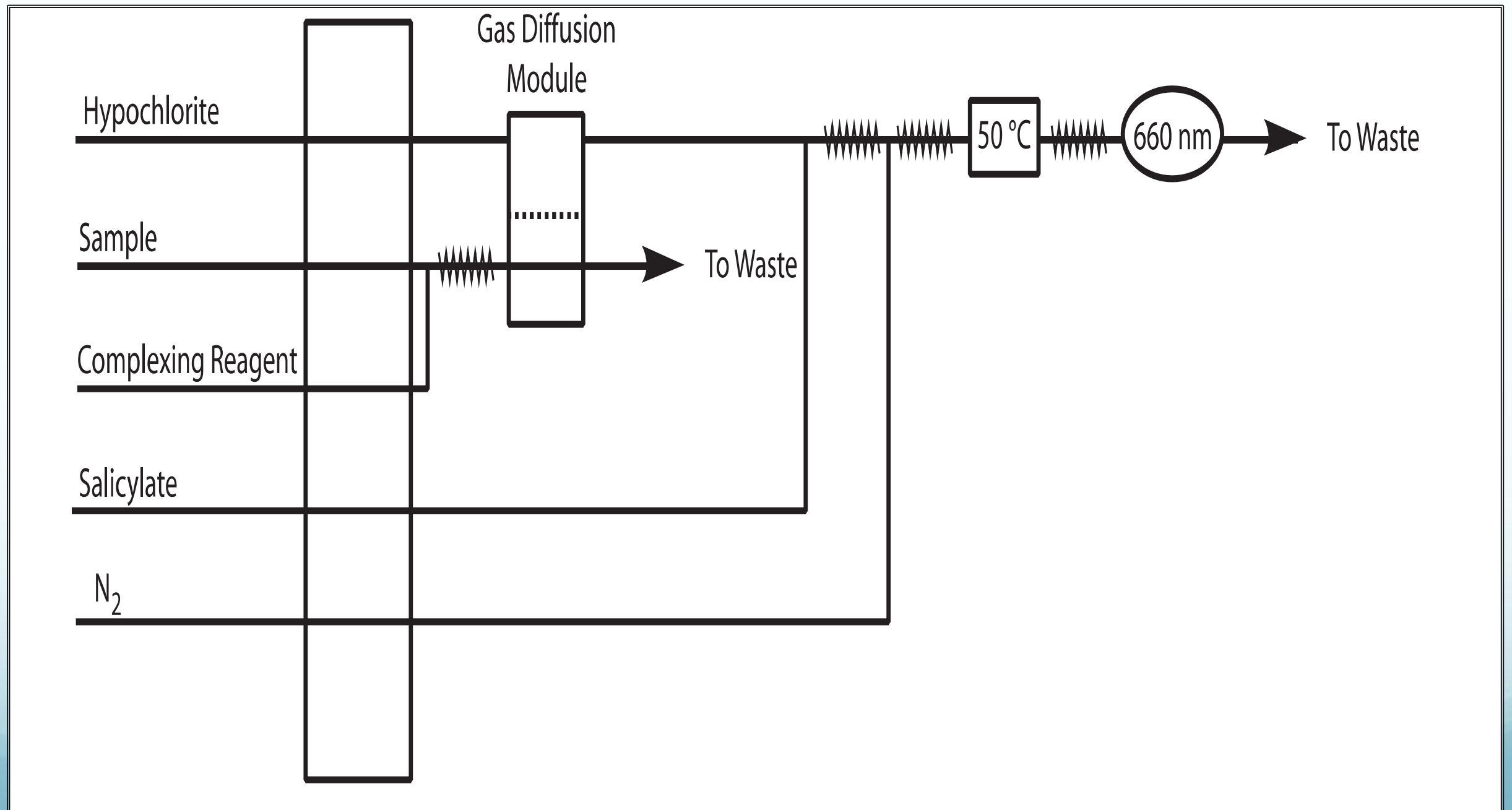
# Typical N in influent and effluent

	Influent	Effluent
TKN (mg/L)	30	2
NO <sub>3</sub> -N (mg/L)	0	17
TN (mg/L)	30	19

# **Minimize digestion time to maximize recovery when nitrate is present**

- 1. 25 mL sample**
- 2. 10 mL digestion reagent**
- 3. 5 stones**
- 4. 160° C for 1 hour**
- 5. 380° C for 30 minutes**
- 6. Cool, rinse sides, cool**
- 7. Dilute to 25 mL (or 50 mL)**

# With EPA allowed method modifications, both ammonia and TKN can be determined with the same CFA method

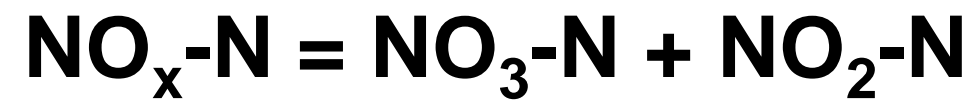
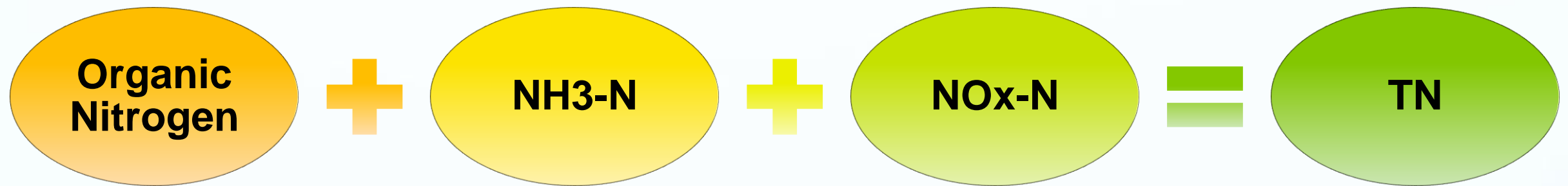




# EPA 351.2 TKN by Gas Diffusion Colorimetry

<b>EPA</b>	<b>Modified</b>
Direct Colorimetry	Automated Diffusion
Salicylate	Salicylate
Tartrate	Citrate
0.01 – 2.0 ppm	0.01 – 20 ppm

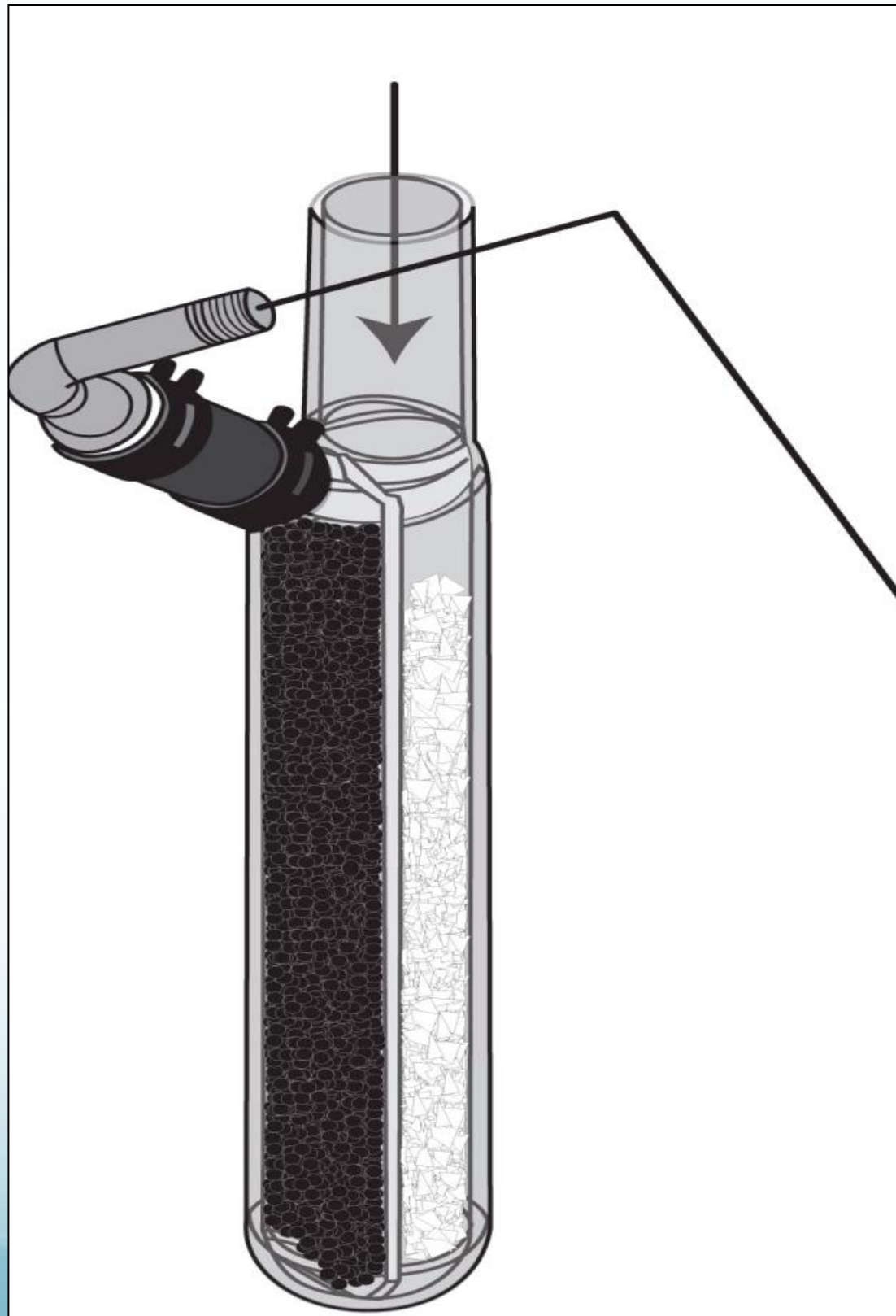
# Low Level Organic Nitrogen Requires a TN analysis, not TKN



# Analysis of Total Nitrogen Compounds

- **Filter preserved sample for DIN**
- **Do Not filter preserved sample for TN**
- **Analyze TN**
- **Analyze DIN (DIN=TIN)**
  - **NO<sub>3</sub> + NO<sub>2</sub>**
  - **NH<sub>4</sub>**
- **TON = TN - DIN**

# HTCO Total Nitrogen Bound (TNb)



Analyzer

**720° C reactor**

**TOC & TNb**

# Total Nitrogen Bound, or TNb is measured on a TOC analyzer

## Oxidation



## Reduction

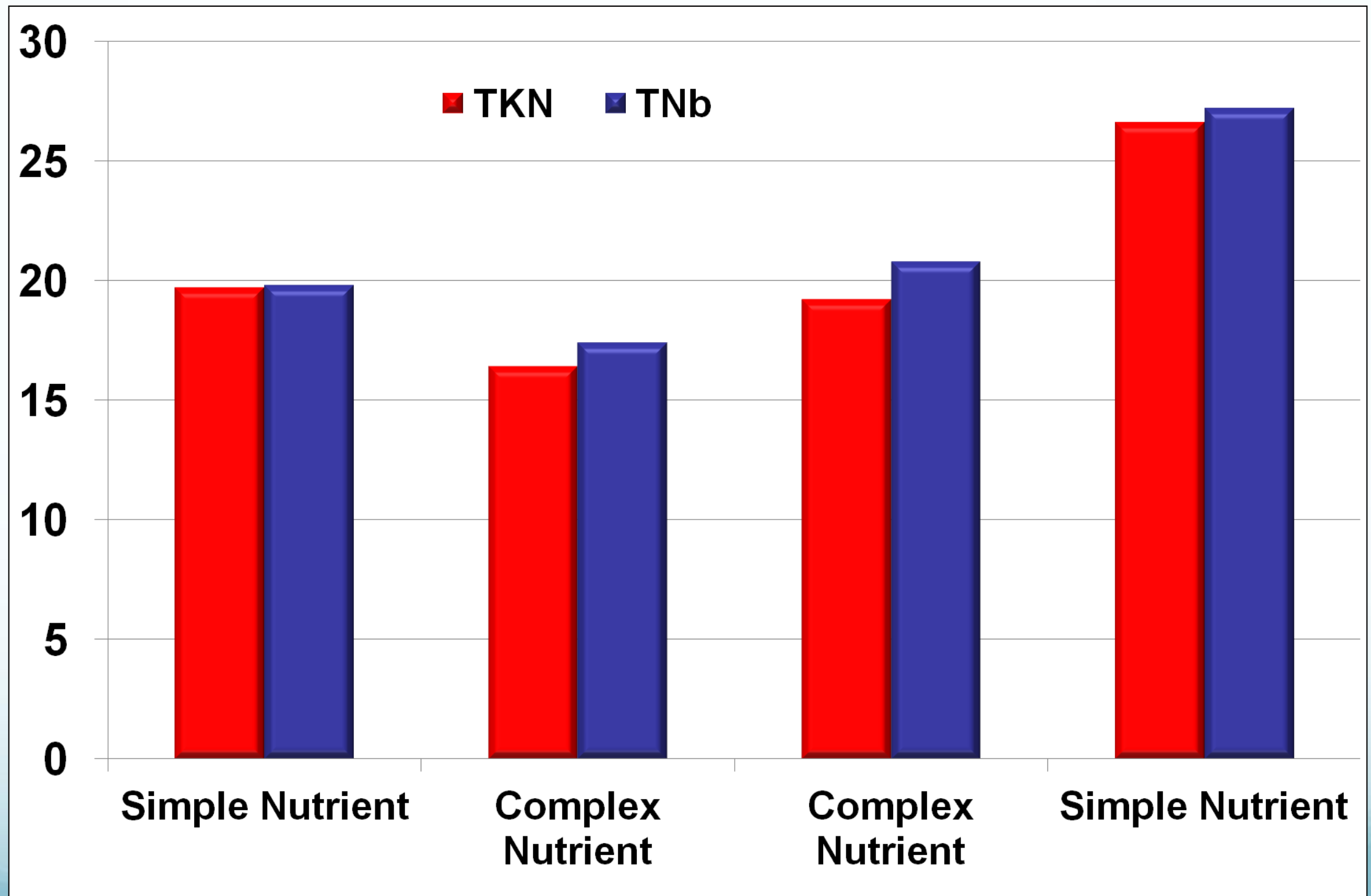


## Detection

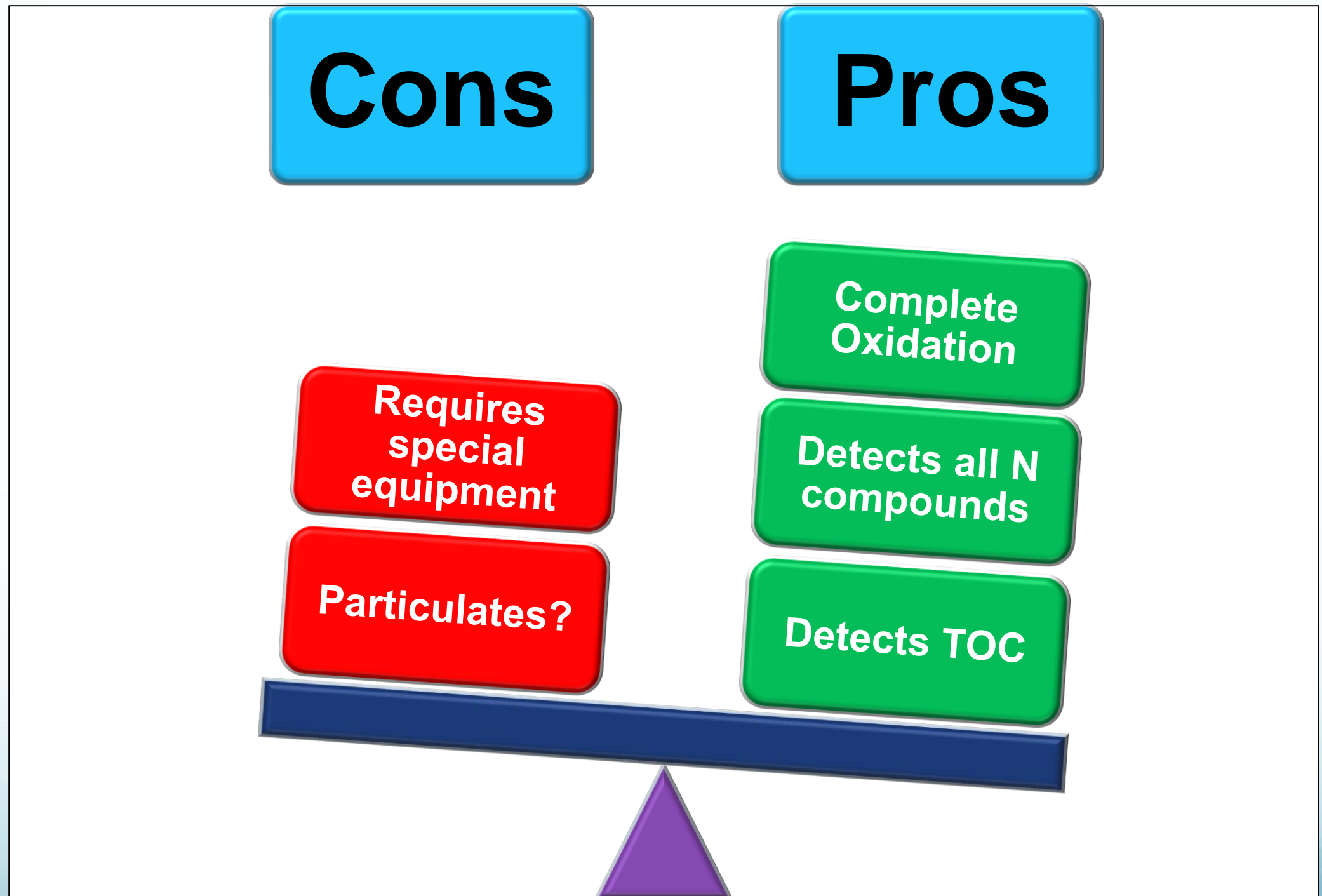
Chemiluminescence

Electrochemical

# Comparison of TNb to TKN (QC samples)



# Advantages and disadvantages of TNb



# Alkaline Persulfate Digestions for TN do not require specialized equipment

Use TOC Analyzer



**Manually digest**

**Determine NO<sub>3</sub>-N**

**Measures all TN**



# Manual Alkaline Digestion for TN can be done in an autoclave or with test tubes

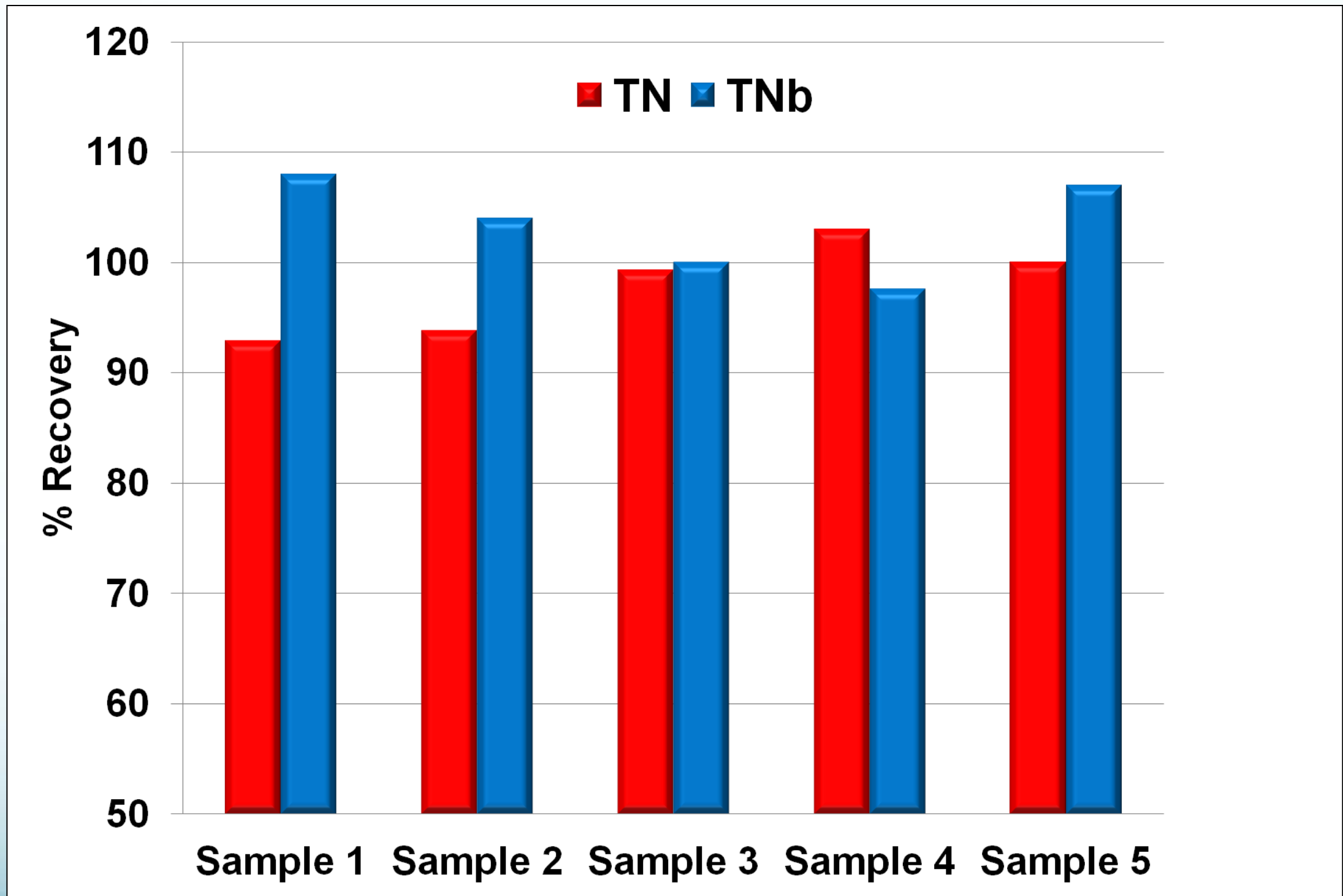
## Alkaline Digestion



## Colorimetry

- Any Nitrate or Nitrate + Nitrite method

# Comparison of TN to TNb illustrates equivalent results on samples tested



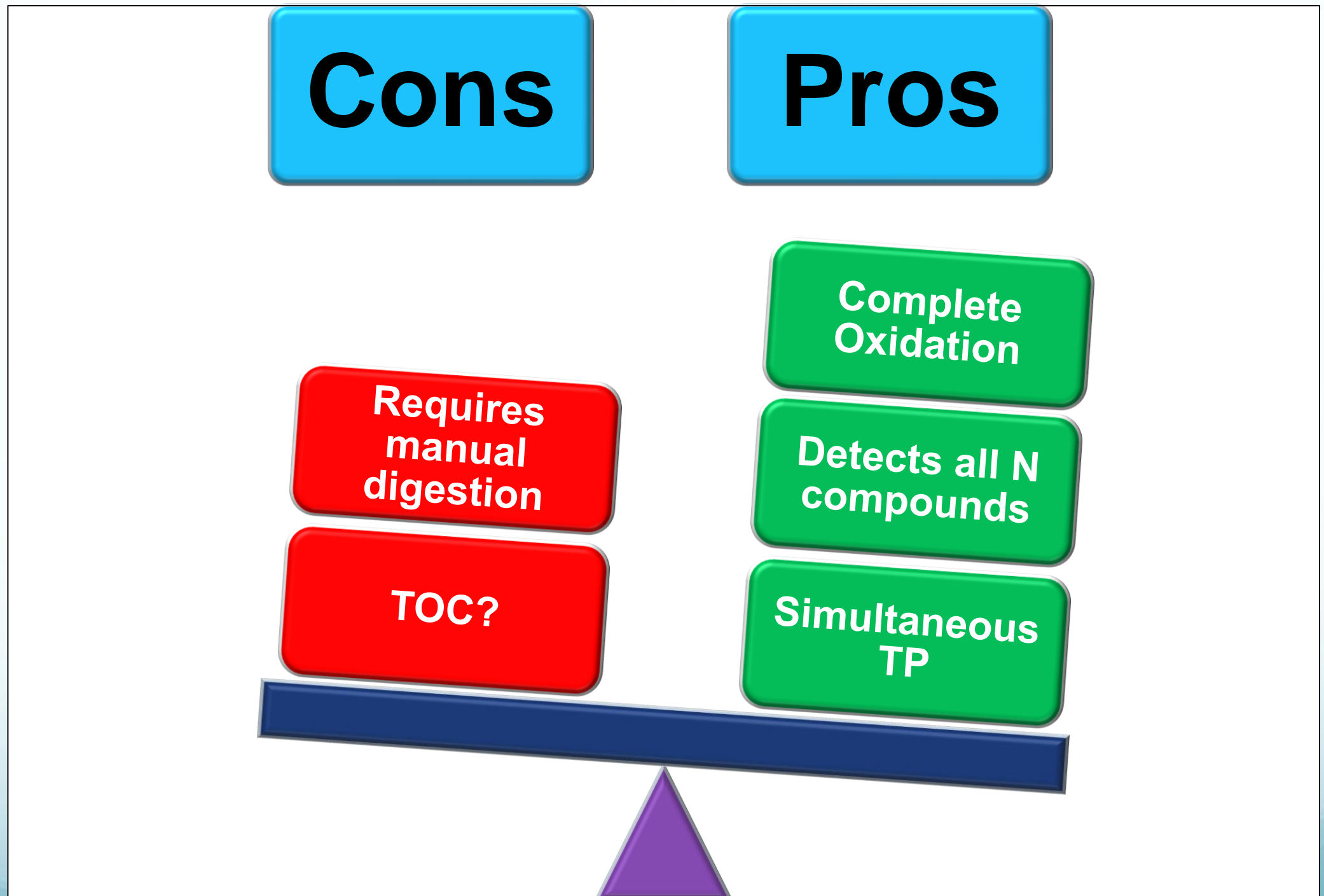
# Comparison of TN to TKN + NO<sub>3</sub>/NO<sub>2</sub> -N

	<b>TOC (mg/L)</b>	<b>TKN + NOx (mg/L)</b>	<b>TN (mg/L)</b>
<b>Cecil</b>	<b>17.5</b>	<b>2.9</b>	<b>2.9</b>
<b>Greenville 1</b>	<b>36.0</b>	<b>3.1</b>	<b>3.5</b>
<b>Hiwassee 2</b>	<b>59.0</b>	<b>8.8</b>	<b>8.9</b>
<b>Irwin</b>	<b>108</b>	<b>17.7</b>	<b>16.0</b>

# Comparison of TN to TKN with % RSD

	<b>TKN (mg/L)</b>	<b>TKN (%RSD)</b>	<b>TN (mg/L)</b>	<b>TN (%RSD)</b>
<b>Bear Creek</b>	<b>0.18</b>	<b>10.65</b>	<b>0.22</b>	<b>5.72</b>
<b>Silver Fork</b>	<b>0.36</b>	<b>19.29</b>	<b>0.41</b>	<b>6.49</b>
<b>Salt River</b>	<b>0.59</b>	<b>25.31</b>	<b>0.76</b>	<b>3.23</b>
<b>Ted Shanks</b>	<b>0.61</b>	<b>25.25</b>	<b>1.05</b>	<b>6.04</b>

# Advantages and disadvantages of manual TN persulfate methods



# Automated Alkaline UV Persulfate Digestions increase throughput



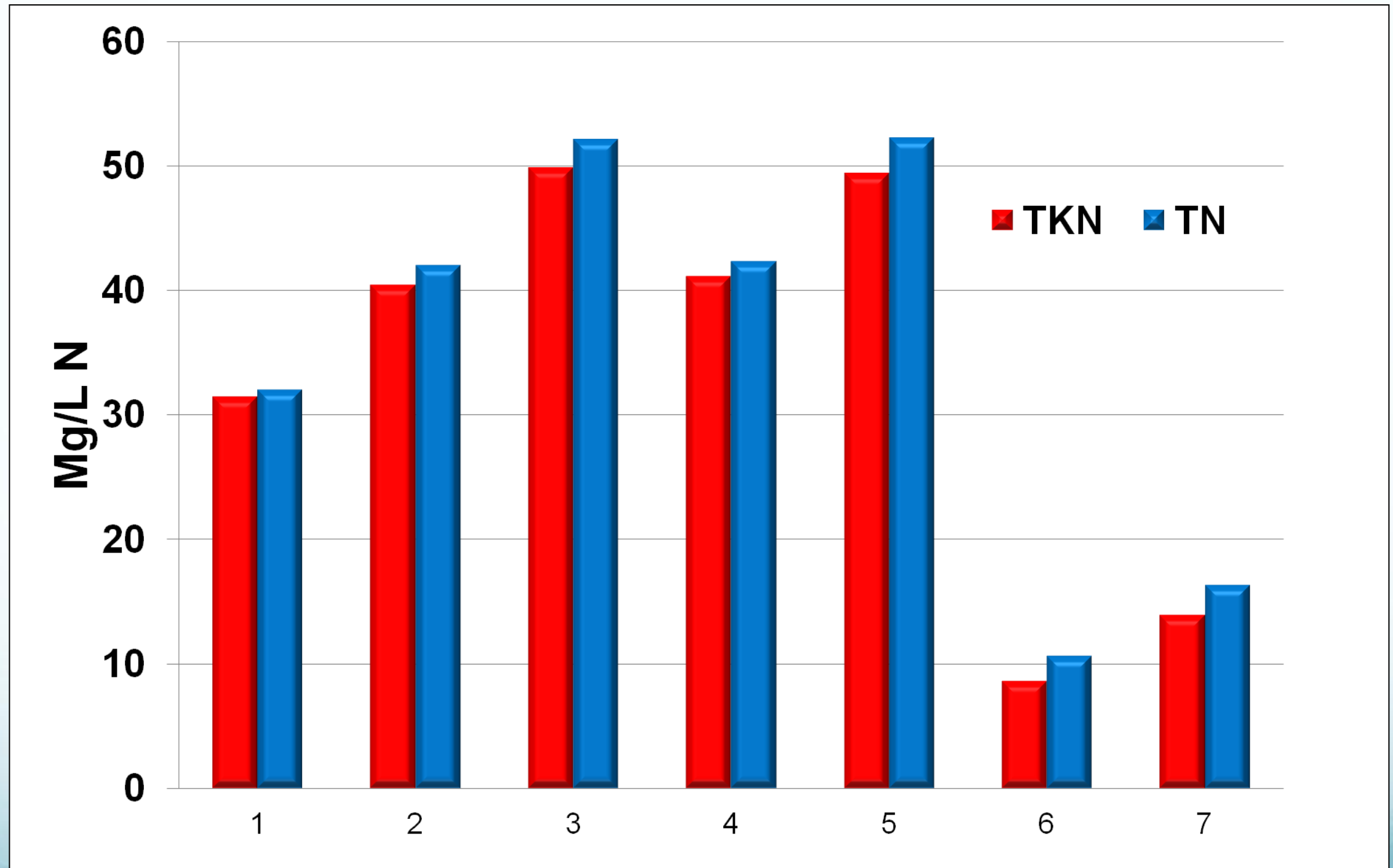
Analyzer

**Automated digest**

**Determine NO<sub>3</sub>-N**

**Measures TDN**

# Comparison of Automated TN with TKN





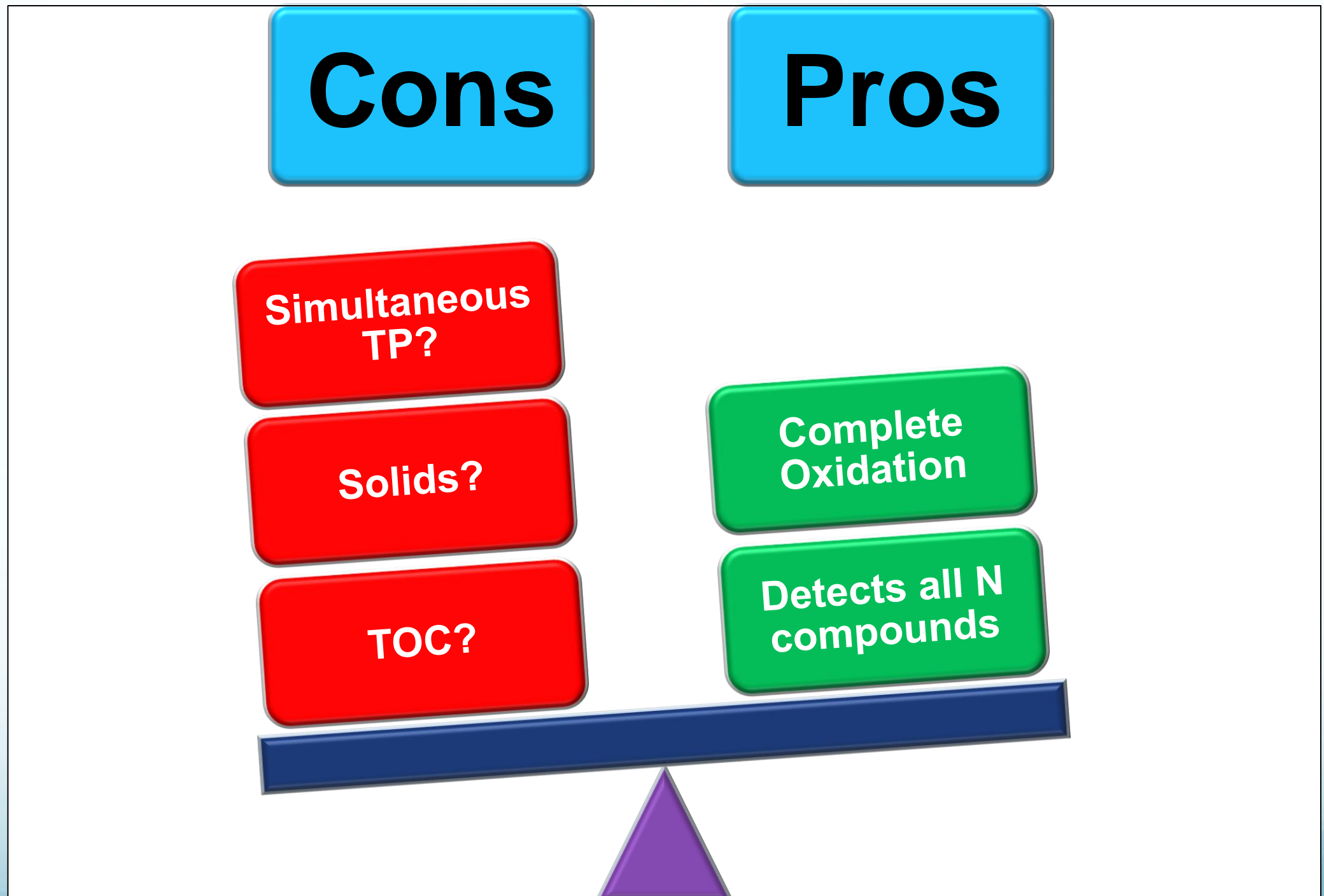
# Unknowns regarding the efficiency of Automated UV Alkaline Digestion



**Particulates?**



# Advantages and disadvantages of automated TN persulfate methods

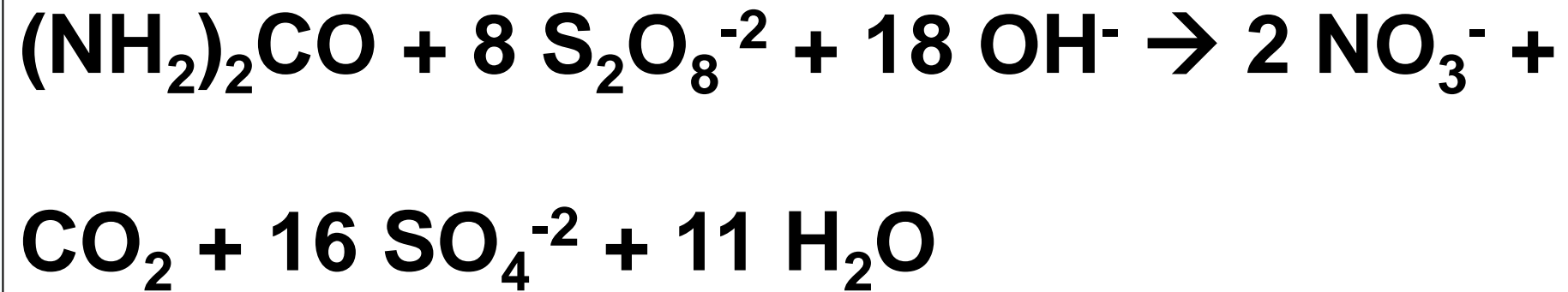


# What samples for Total Nitrogen may look like



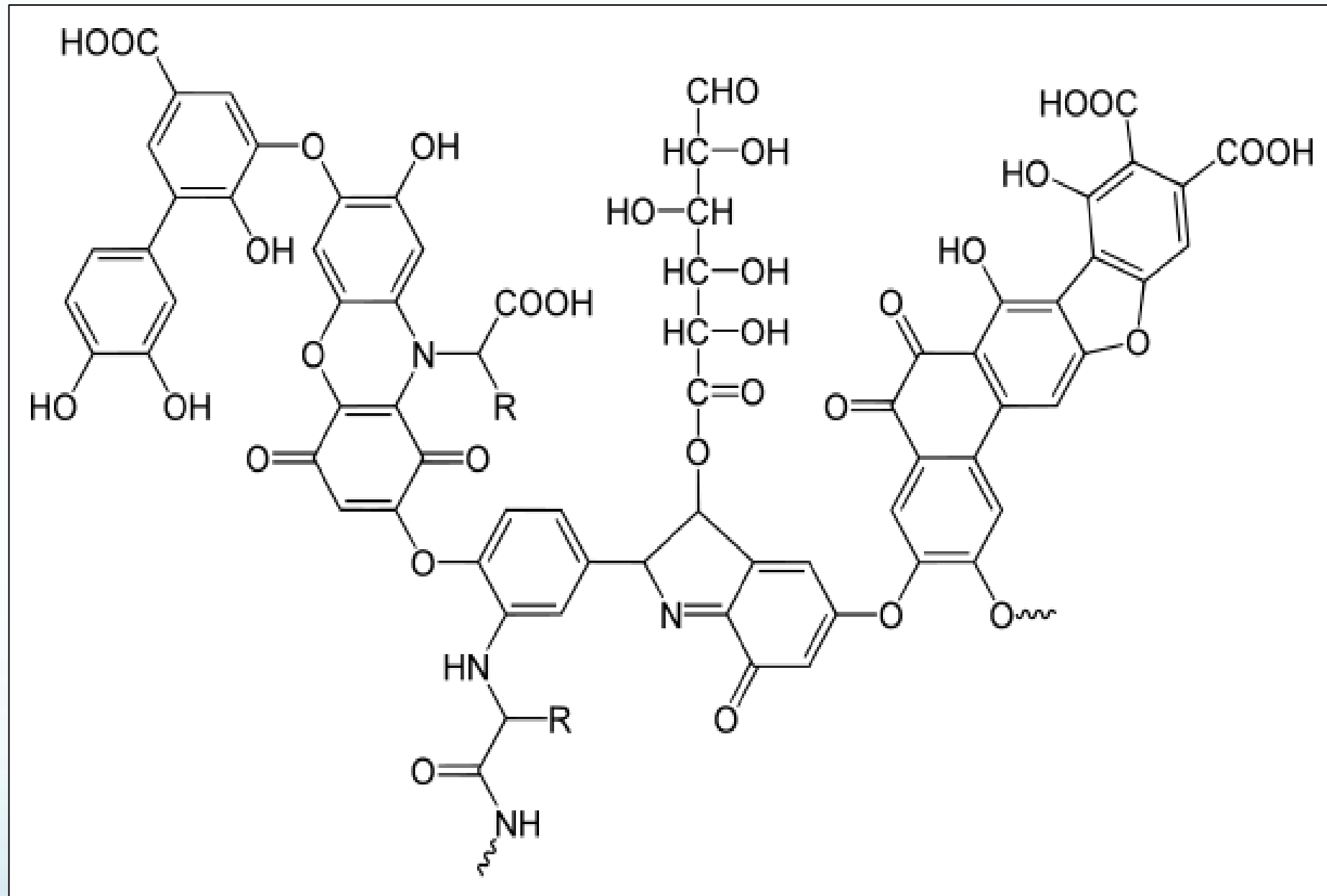


# Organic matter consumes persulfate too



**About 100 ppm C upper limit**

# A Humic acid molecule



# **Particulates consume persulfate and are hard to sample**

**~ 15 - 20% N attached to particulates**

**Particulates > ~ 30 ppm TSS N/A by CFA**

**Non quantitative transfer of particulates to HTCO**



# Total Phosphorus Digestion Methods

**Acid Persulfate (beaker or autoclave)**

**Kjeldahl (block digestion)**

**Both convert inorganic and organic phosphorus to orthophosphate**

**Acid persulfate digestion for total P can be done in a beaker, autoclave, or test tubes**

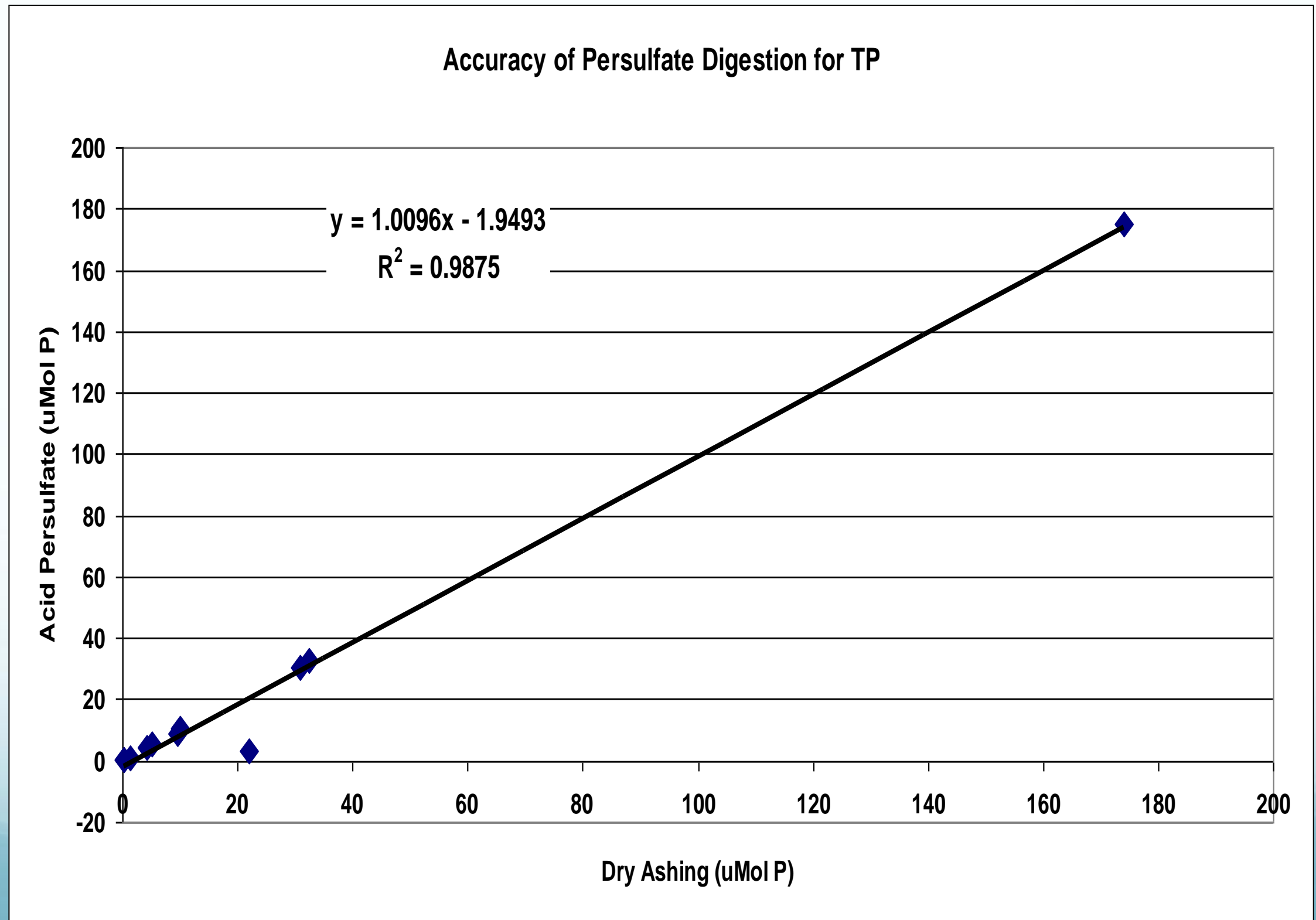
## **Acid Persulfate Digestion**



## **Colorimetry**

- **Any Phosphate method**

# Comparison of Dry Ashing and Acid Persulfate for Total P





# Outlier data on persulfate and wet oxidation comparison for TP digestion

<b>Material tested</b>	<b>Dry Ash uMol P</b>	<b>Persulfate uMol P</b>
<b>Montmorillonite</b>	<b>1.44</b>	<b>1.00</b>
<b>Kaolinite</b>	<b>22.0</b>	<b>3.04</b>

# Potential problems with acid persulfate digestions for total phosphorus

**Too much acid**

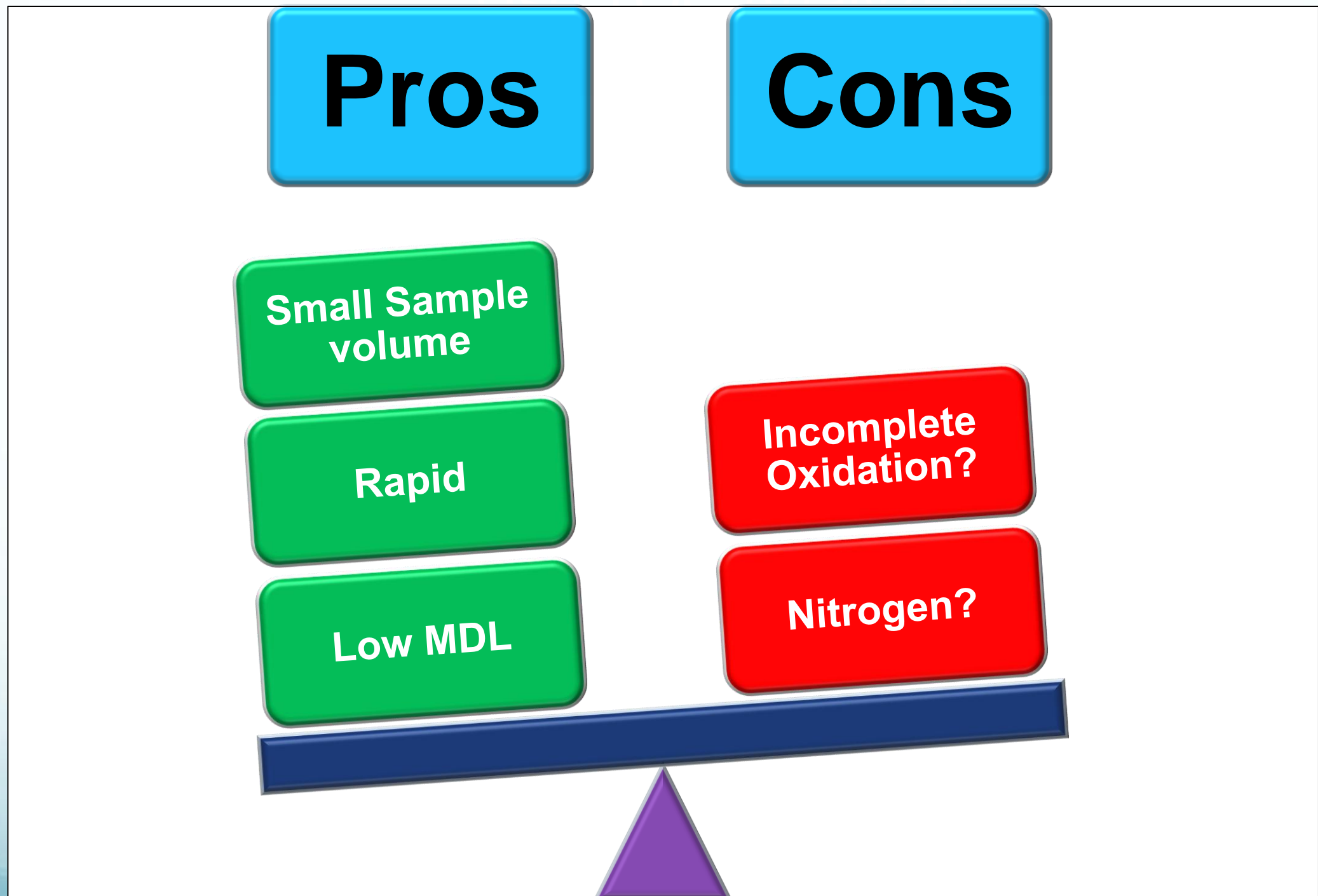
**High chloride**

**Organic Matter**

**Excess particulates**

**Contamination**

# Advantages and disadvantages of manual acid persulfate methods for TP



# “Kjeldahl” Digestion for TP

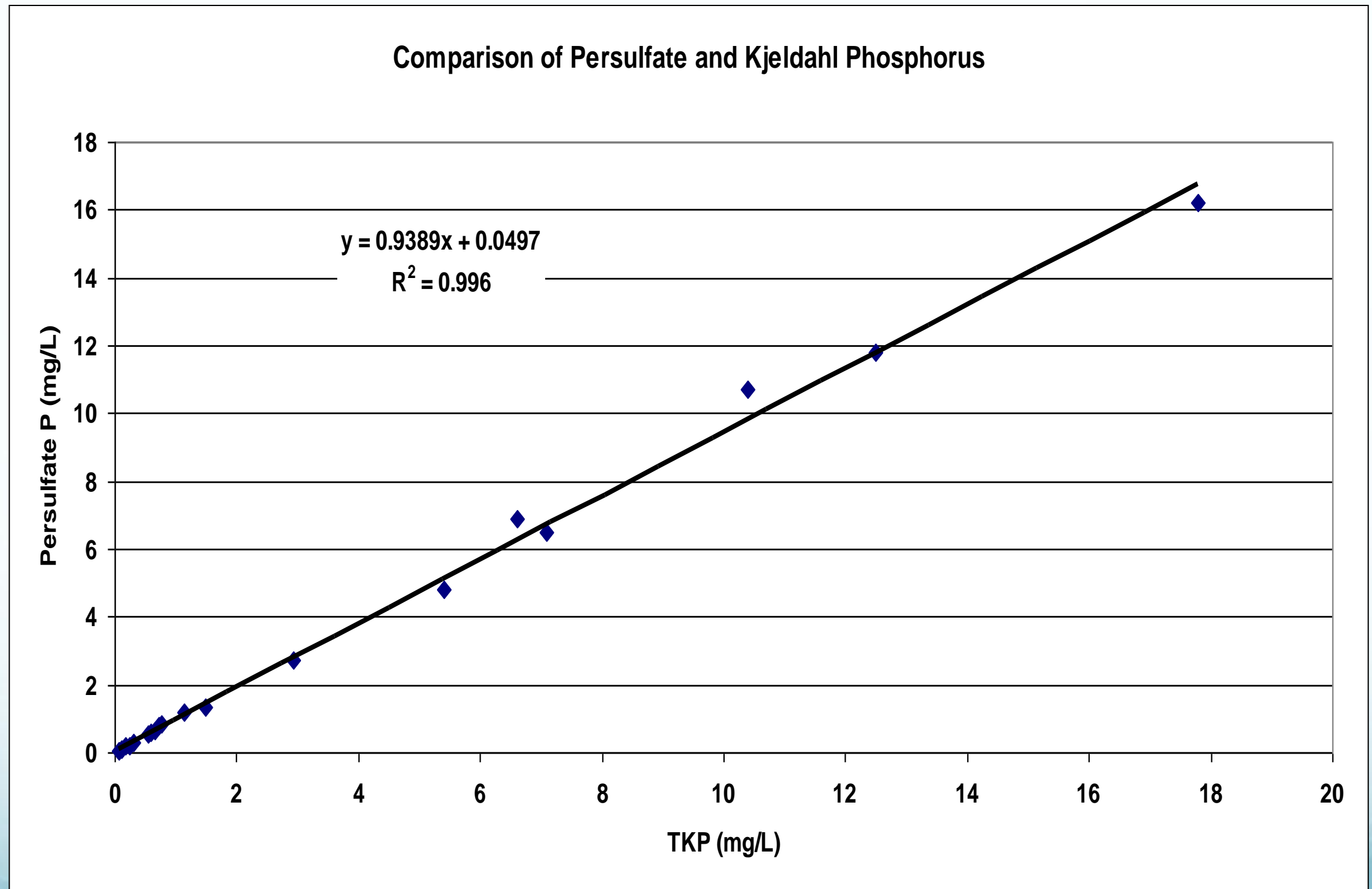
## Kjeldahl Digestion

- $R\text{-HPO}_4 + \text{H}_2\text{SO}_4 \rightarrow \text{H}_3\text{PO}_4 + n\text{CO}_2 + n\text{SO}_2 + n\text{H}_2\text{O} + \text{H}_2\text{SO}_4$
- Contains catalyst

## Colorimetry

- Any Phosphate method
  - High Acid concentration
  - Catalyst

# Comparison of persulfate and Kjeldahl for TP



# Potential problems with Kjeldahl digestions for total phosphorus

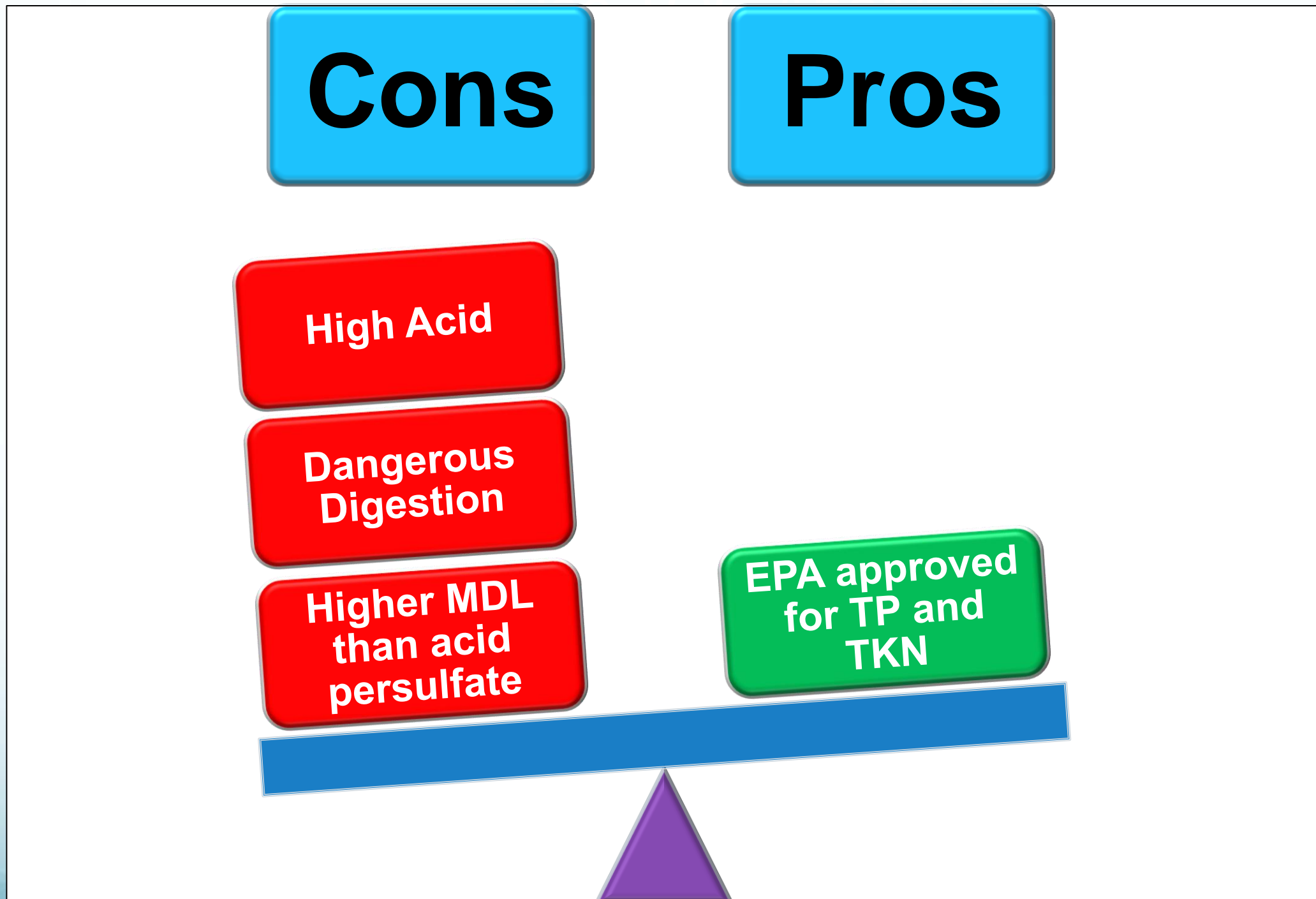
**High Acid Concentration**

**Hazardous Digestion**

**Catalyst**

**May require dilution of acid**

# Advantages and disadvantages of manual acid persulfate methods for TP



# **Using USGS I-4690-91 to overcome problems with normal Kjeldahl phosphorus methods**

**Uses Dialysis**

**includes On-line dilution**

**Includes On-line filtration**

**Controls acid concentration**

**Low Detection Limit**



# Alkaline Persulfate Digestions for Total P

Use TOC Analyzer



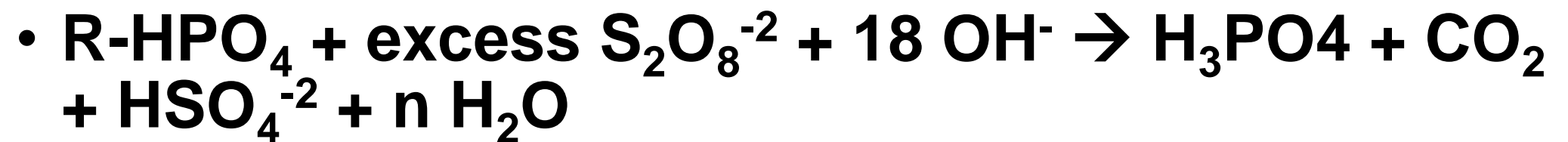
**Manually digest**

**Determine  $\text{PO}_4^{-3}$**

**Measures TP**

# Manual Alkaline Persulfate Digestion for TP

## Alkaline Digestion



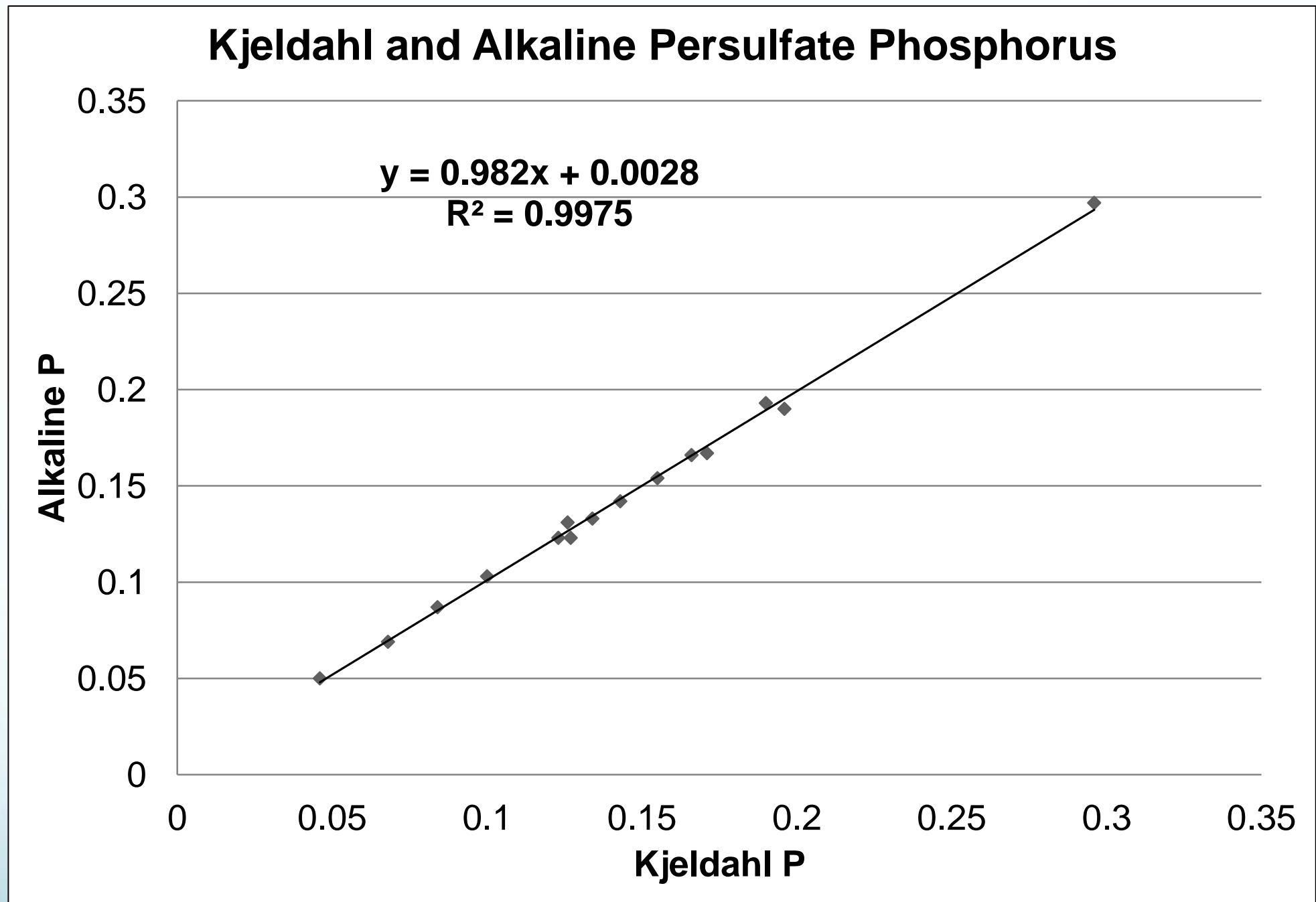
## Colorimetry

- Any phosphate method

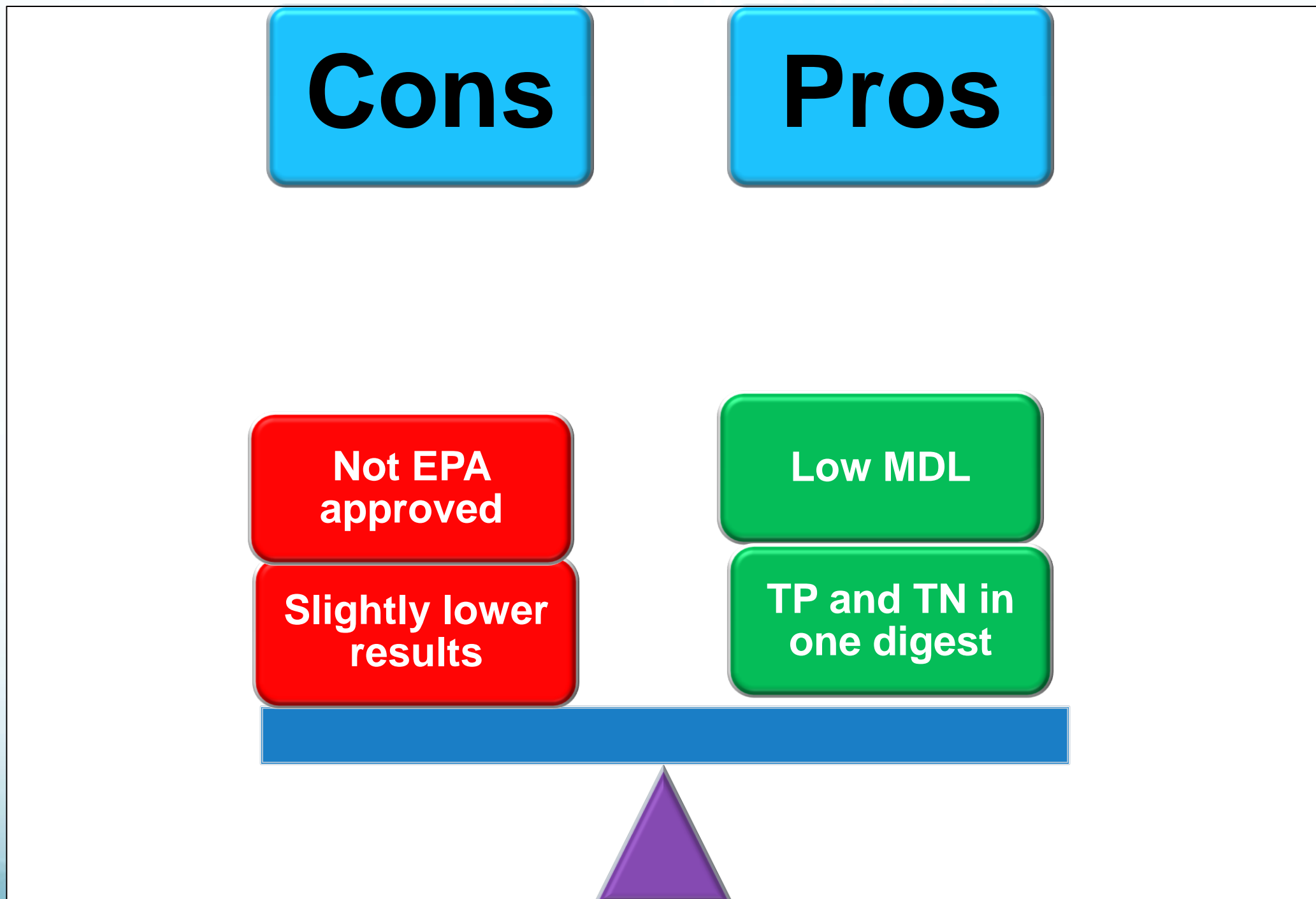
# Comparison of Alkaline Persulfate and Acid Persulfate Total Phosphorus

<b>Compound</b>	<b>Alkaline (mg/L P)</b>	<b>Acid (mg/L P)</b>
<b>Adenosine tri-phosphate (ATP)</b>	<b>0.166</b>	<b>0.176</b>
<b>Glycerophosphate</b>	<b>0.196</b>	<b>0.204</b>
<b>Phenyl Phosphate</b>	<b>0.168</b>	<b>0.179</b>
<b>Phytic Acid</b>	<b>0.177</b>	<b>0.180</b>

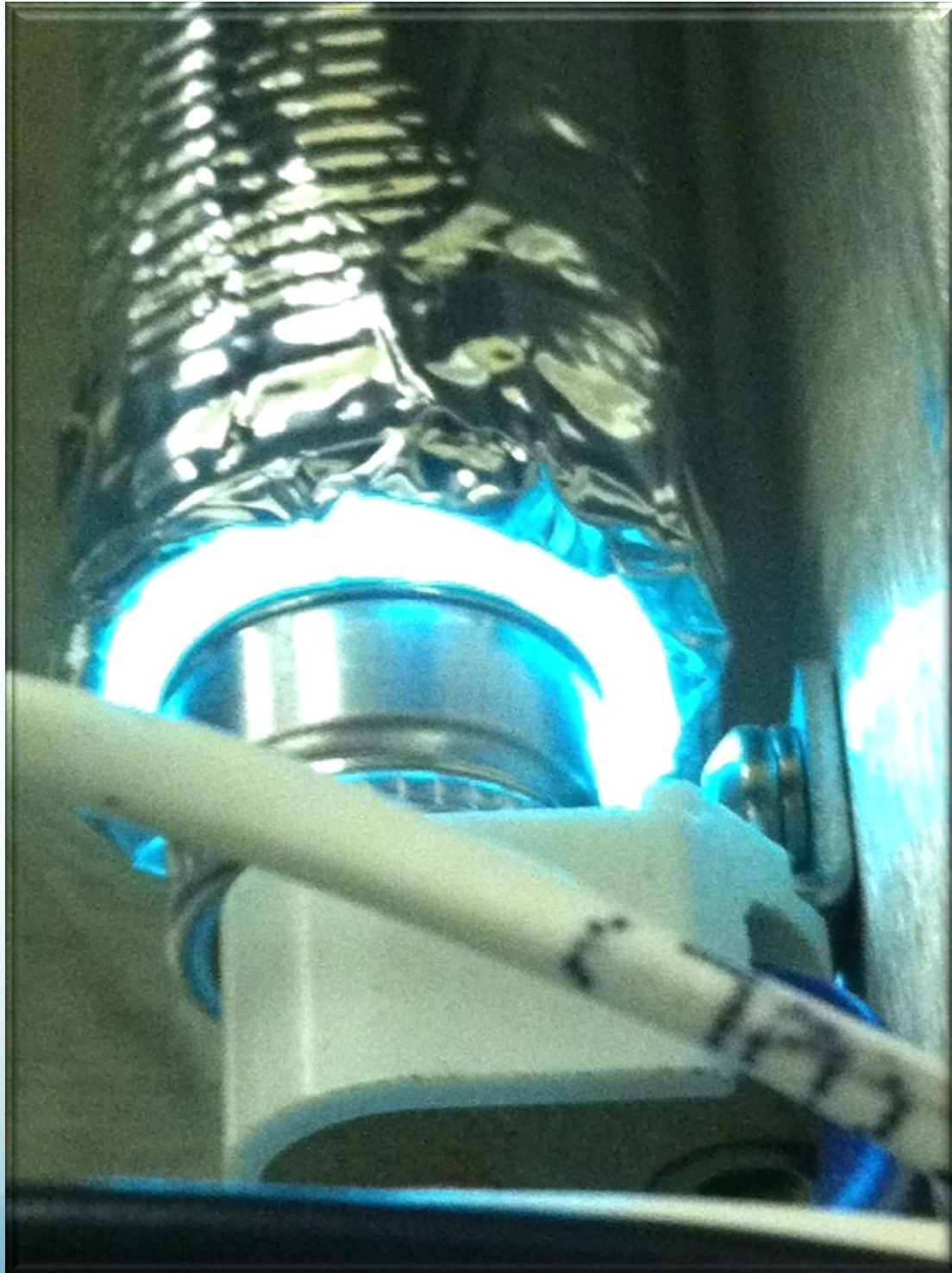
# Comparison of TKP and Alkaline Persulfate



# Advantages and disadvantages of manual Alkaline persulfate method for TP



# Automated Alkaline UV Persulfate Digestions



Analyzer

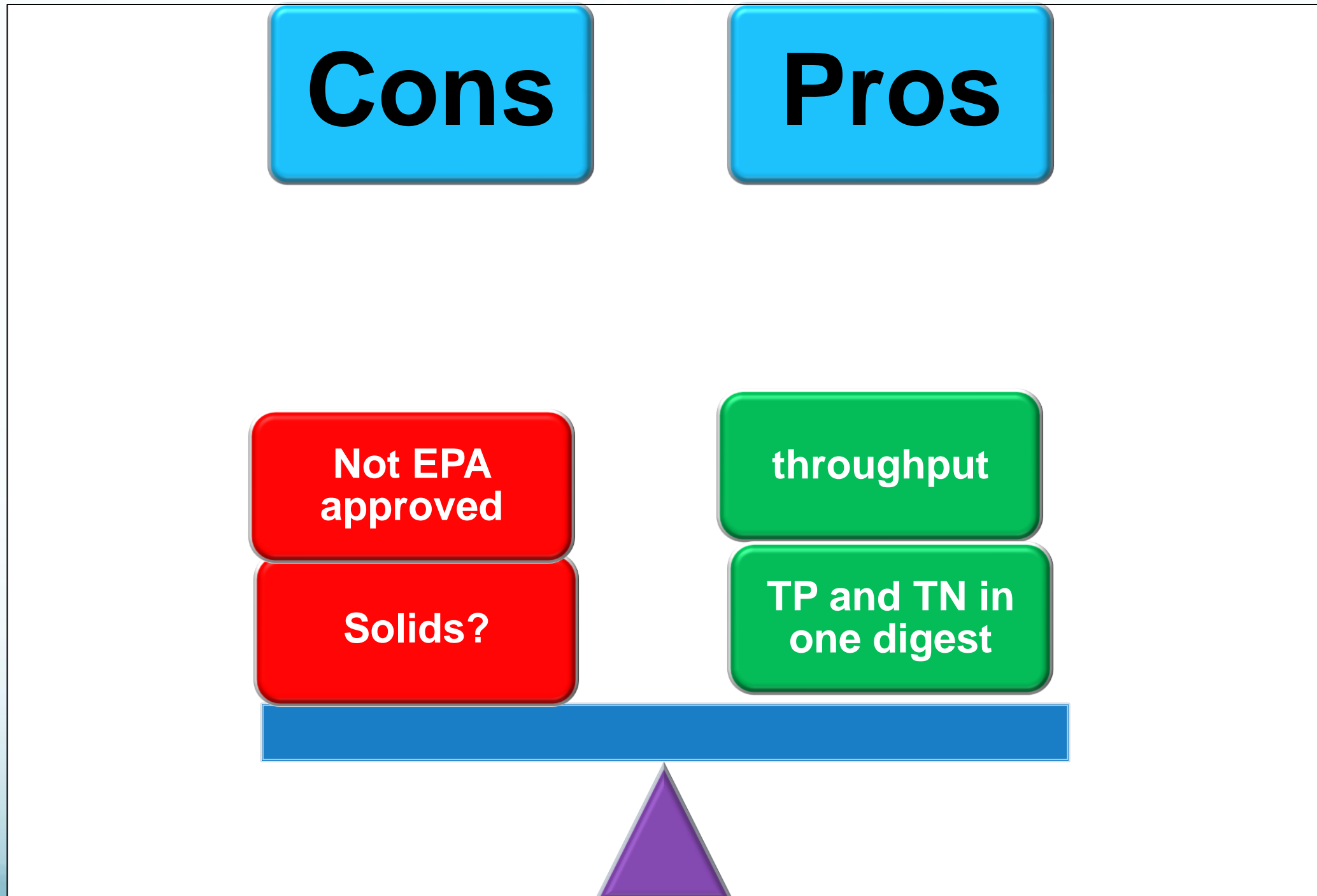
**Automated digest**

**Determine  $\text{PO}_4^{-3}$**

**Measures TDP**



# Advantages and disadvantages of automated persulfate method for TP



**Thank You!!**

**William Lipps  
OI Analytical**

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