Modeling Petroleum Hydrocarbon Degradation. Part I, Gasoline. Statistics and Pattern Assemblage.

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Once released to the environment, petroleum hydrocarbons are not stable. That is, they undergo chemical changes consistent with their surroundings and their chemical structure. The changes that take place can be described in a variety of ways.

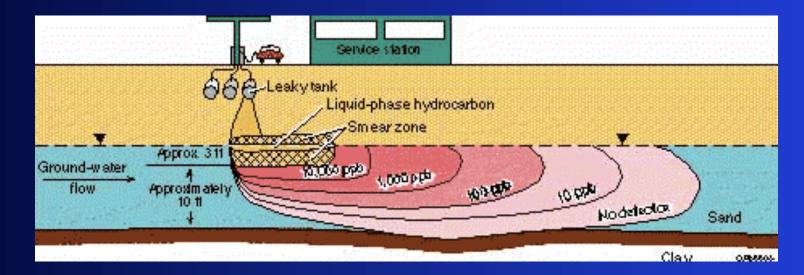
The overall effects of individual reactions that change individual hydrocarbons have come to be known collectively as "petroleum weathering."

## Here is the problem:





A leak from an underground gasoline storage tank releases a single product which first saturates the adjacent soils and then migrates into groundwater. Following the groundwater gradient . . .



We can collect samples of original fuels, and impacted soils and/or groundwater, ship them to a high level forensic analytical laboratory and collect a large amount of data.

For gasoline releases, we can follow the five principal hydrocarbon classes of gasoline: paraffins, isoparaffins, aromatics, naphthenics and olefins or the PIANO (or PONA) analyses. Unfortunately, not enough target analyte uniformity yet among labs.

With such data, we can examine each of the five major categories of gasoline HCs and begin to ask questions.



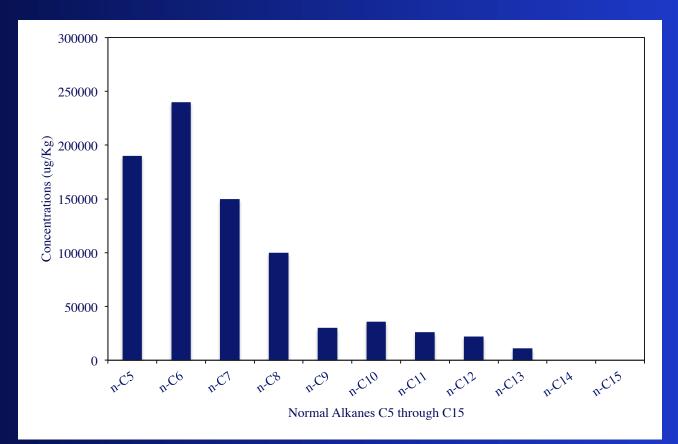
Paraffins
Isoparaffins
Aromatics
Naphthenics
Olefins
Other

Fuel Type	Alkane Range	BP Range (°C)
Gasolines (GRO)		
Automotive	C4 - C12	50-200
Aviation	C4 - C12	25-170
Distillate Fuel Oils	(DRO)	
Fuel Oil No. 1	C9 - C16	150-300
Fuel Oil No. 2	C11 - C20	160-360
Residual Fuel Oils	(RRO)	
Fuel Oil No. 4,	5,6 > C20	350-650

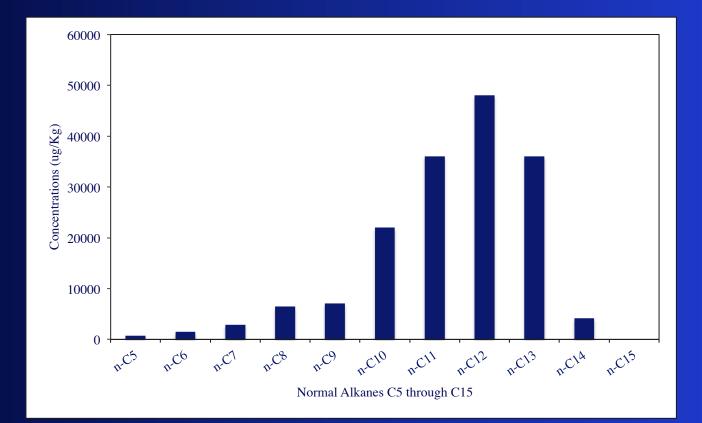
Component	Formula	BP (°C)
Pentane	$n-C_5H_{12}$	36
Benzene	$C_6H_6$	80
Toluene	$C_6H_5CH_3$	110
Isooctane	$C_{8}H_{18}$	116
Octane	$n-C_8H_{18}$	126
o-, m-, p-Xylene	$(CH_3)2C_6H_4$	144
Nonane	$n-C_9H_{20}$	150
Decane	$n-C_{10}H_{22}$	174
Undecane	$n-C_{11}H_{24}$	196
1,2,4-Trimethylbenzene	$(CH_3)3C_6H_3$	176
Pentadecane	$n-C_{15}H_{32}$	271



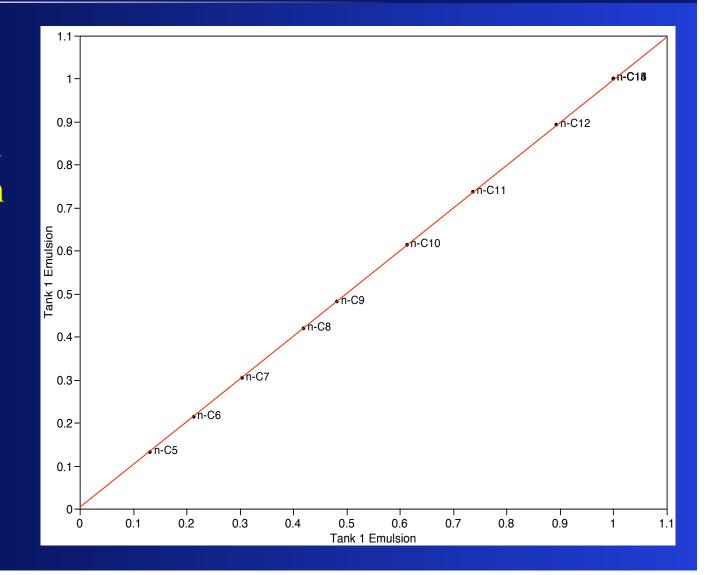
#### Case 1: UST sludge – leaded gasoline



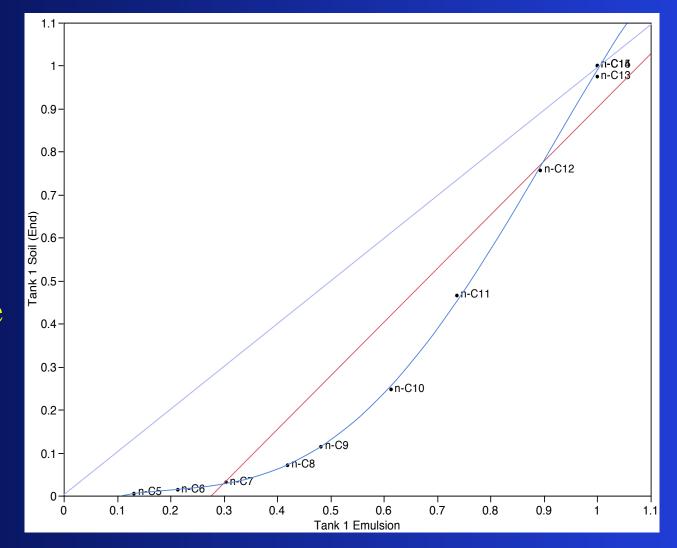
Case 1: soils, down gradient from UST.



If we mix one gasoline with only itself, and do a regression for n-alkanes against itself where no HC changes are occurring, we must obtain a straight line.



Case 1: if we compare UST contents (X-axis) with the outside soil HC (Y-axis), a different picture emerges - HC loss, *i.e.*, degradation of n-alkanes.



Case 1 pattern shows loss of n-alkanes in the soil sample compared with the starting material. The curvilinear relationship below the individual alkane line, a mass loss, *i.e.* degradation.

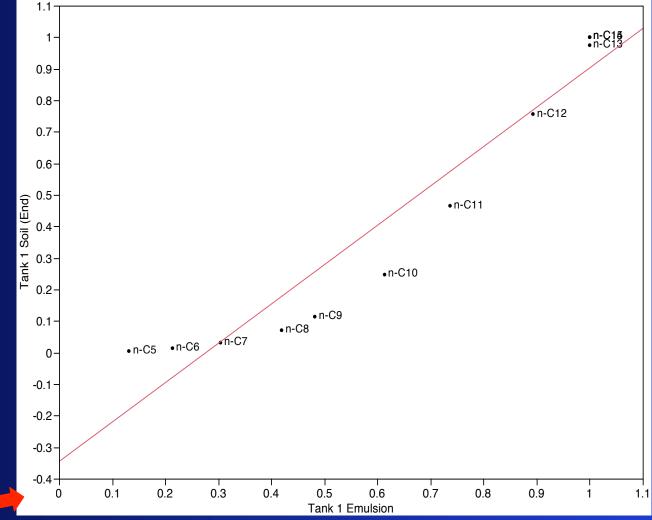
If there were to be a curvilinear relationship above the line, it would mean that individual hydrocarbons have been proportionally added when the UST gasoline was released to the soils. Enrichment does not occur in degradation but may if another HC source is present

There are two ways of modeling this behavior mathematically for Case 1:

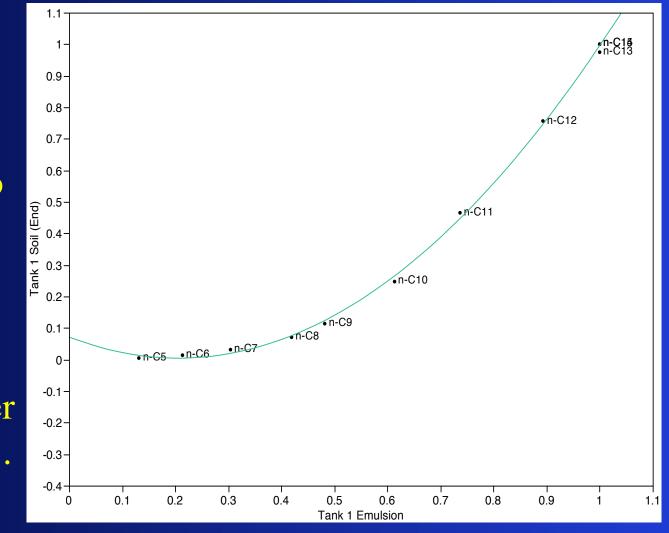
Linear modeling of loss showing how much was lost (or gained) from the Y-axis compared with the starting material.

Polynomial modeling of loss to determine how many factors are needed to define a statistically significant relationship.

Case 1: Extending the Y-axis to the negative shows the value of the b factor in linear equation Y = mX + bY-axis is negative



Case 1: But a  $(Y = mX + nX^2)$ +b) quadratic model, the two factors tells us how many gasolines we have, starting gas and another (degraded) gas.



This type of analysis can be done for each PIANO fraction in Case 1.

The most important aspect of the analysis is that the linear approach gives us a measure of the % loss of the entire fraction (we have seen only paraffin fraction), while the quadratic approach counts the number of different gasolines we have in the system.

Plus the statistics:  $r^2$  and significance (p values).

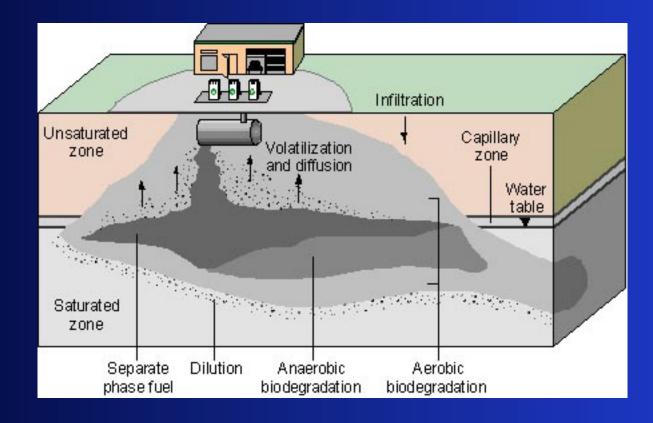
What does this analysis give us where one gasoline is not like the other, *i.e.*, not manufactured by the same producer.

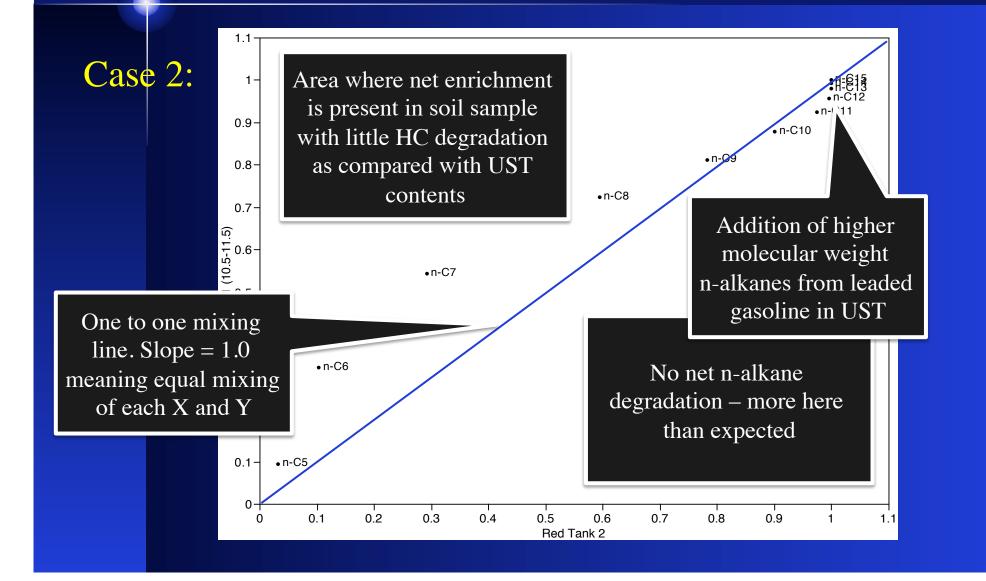
Two alternatives:

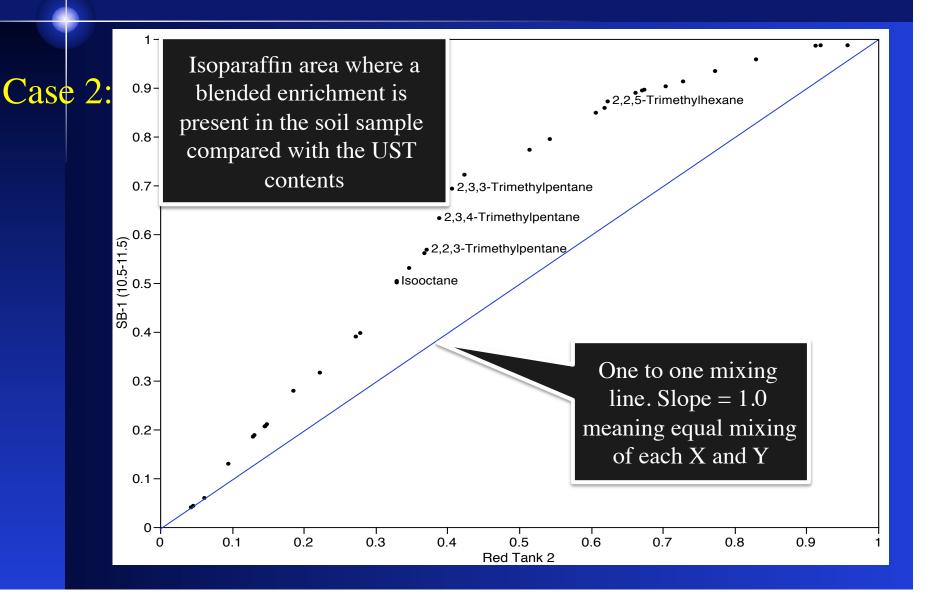
Degradation - if solely from 1 source and/or potential mixing from another degraded source.

Enrichment – mixing from 2 or more sources.

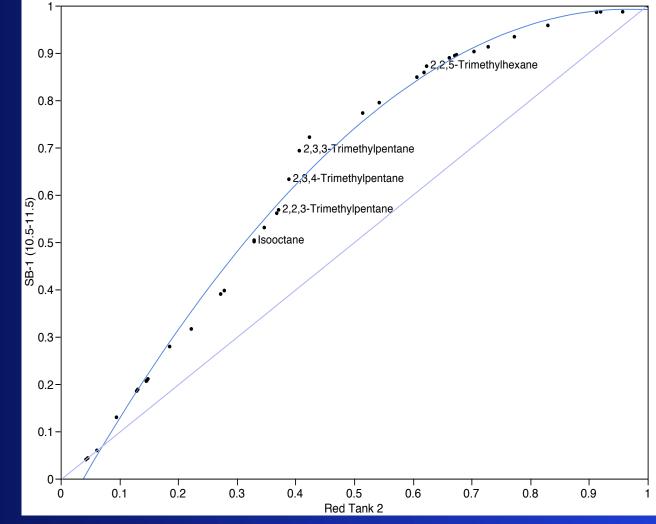
# Of course, Case 1 is relatively simple. On the more complex side is Case 2...

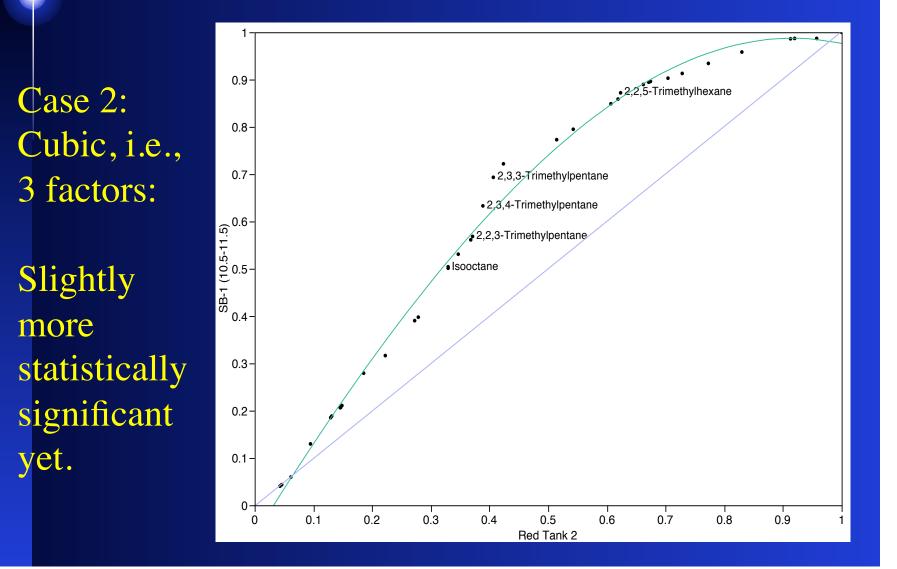




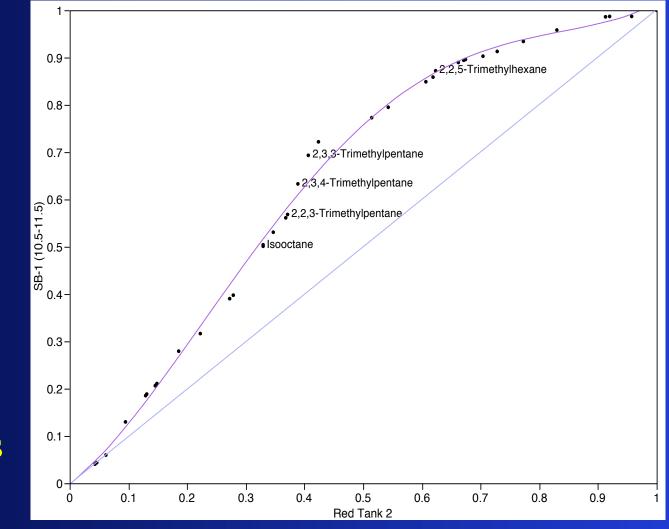


Case 2: Quadratic, *i.e.*, 2 factors: a little more statistically significant.





Case 2: Quartic, *i.e.*, 4 factors As the number of factors increases, r<sup>2</sup> increases somewhat.



Again, Case 2 data can be analyzed for each fraction, for each sample for each fraction until all five PIANO fractions have been analyzed.

What the investigator is left with at the end is an understanding of how gasolines in the soils were blended compared with the starting material in the UST being examined for loss.

In Case 2, no loss from the UST, gain once the HCs were in soils, *i.e.*, <u>other sources, not USTs</u>.

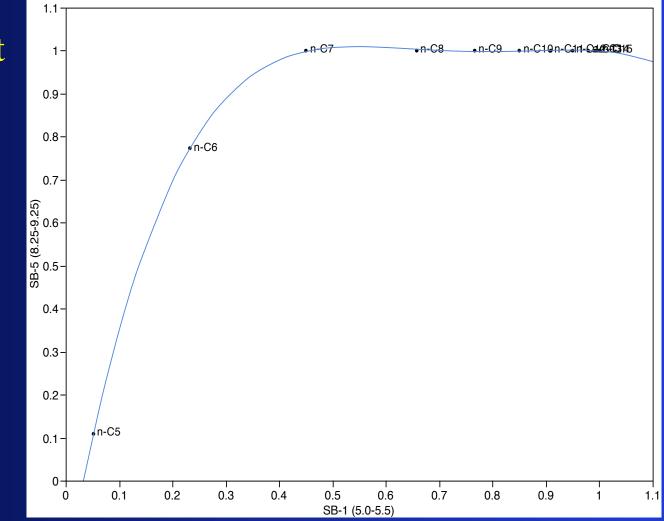
The degree of the polynomial defines how many mixing sources there are:

If there is evidence of degradation, then the one in the tank counts as one source to the soil.

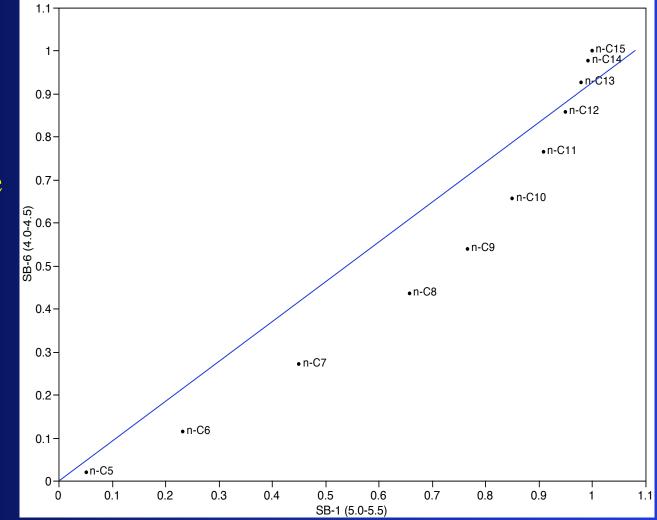
If there is no evidence for HC degradation, then the one in the tank does not factor into the count as a source to the soils/sediment because it is not involved in the final composition of the gasoline HC backbone in the soils.

Case 2: What about soil v. soil? SB-5 soils in Case 2 . . . early

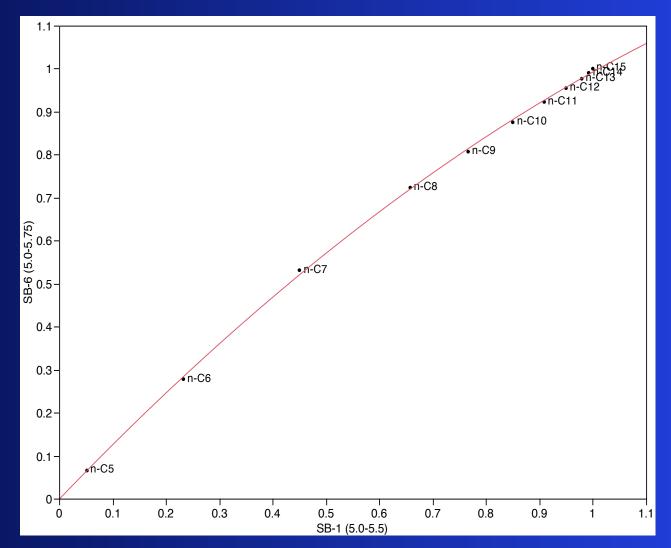
n-alkanes only v SB-1.



Case 2: SB-5 vs. SB-6 (4-4.5') in the same type of analysis, we find a different product mix in SB-6. Not from USTs but related.

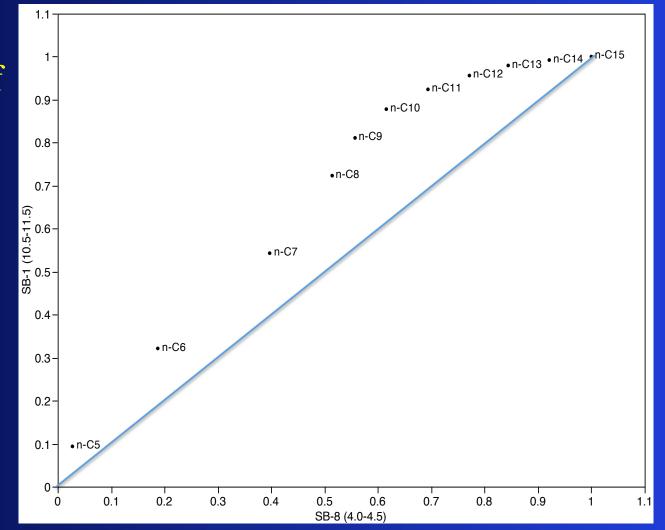


Case 2: SB-6 (5-5.5)' in the same location has a different product mix deeper into SB-6 than at shallower depths.



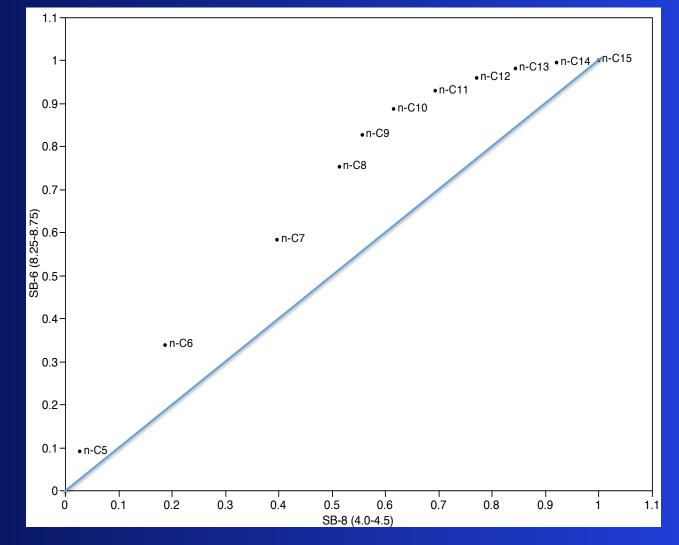
Case 2: the comparison of SB-1 (10.5-11.5') with SB-8.

Evidence of two product distributions.



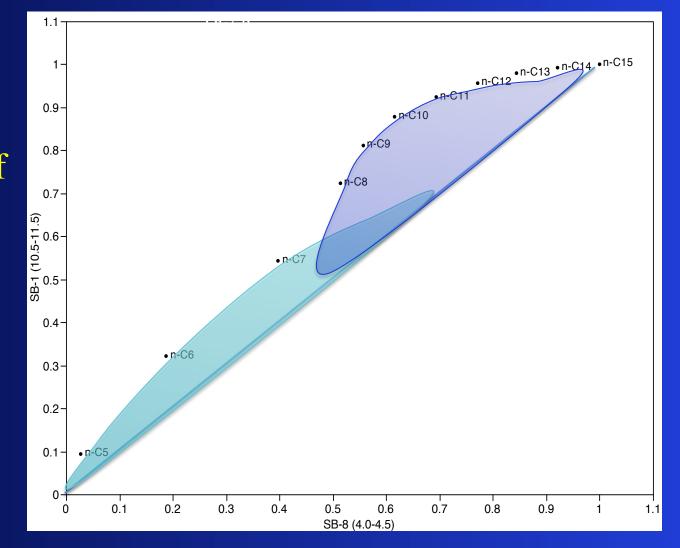
Case 2: SB-6 (8.25-8.75') with SB-8.

Evidence of two different distributions.



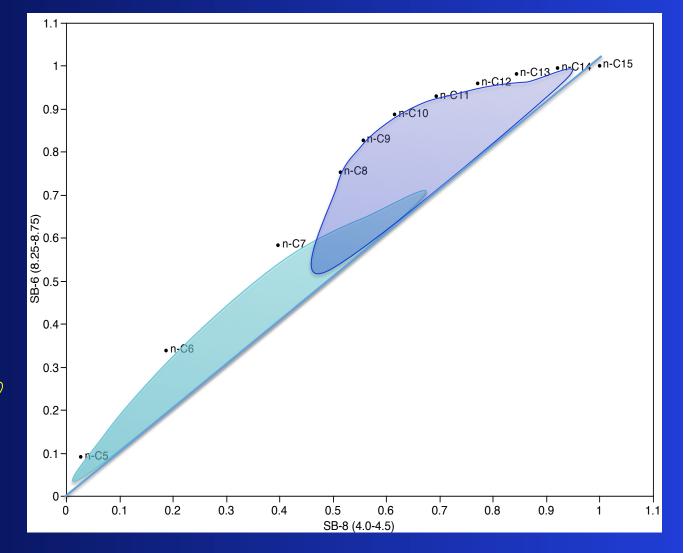
Case 2: Can we measure under the curves? Yes, of course.

Data show 45:55 and 10% overlap between products.



Case 2: For shallower depth at SB-6, the approach, appears to be similar . . .

... and data 45:56 and 9 % overlap, *i.e.* virtually identical.



Best of all, the principles remain the same regardless of scenario. Degradation or enrichment for any UST content model can be used to examine hypothesis testing of UST release(s) for a variety of products.

Case 1 & 2 hydrological analyses demonstrated that release of petroleum products from different tanks would have impacted same loci. Chemical data, thus, reinforced the hydrological transport models for each site.

Both chemical and geological/hydrological results were consistent with the types and timing of petroleum product sales as determined by depositions and/or attorney records search completed for both cases.

On site photography played critical role.

Overall synthesis approach has been written about and discussed in international technical conferences and applied in litigation.

#### Conclusion:

Pattern assemblage is the focusing of statistics and geochemical principles which produces <u>simple</u>, easily understood "jury exhibits." Allows an investigator to explain complicated science to non-scientific audiences in a <u>simple</u> manner.

Scientific Question: Why collect an array of geochemical data if you are not going to use it?

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