## FUGITIVE METHANE FORENSICS

Alan Jeffrey, Ph.D. NEMC Meeting Chicago, July 14, 2015





## Ross Store, Fairfax District

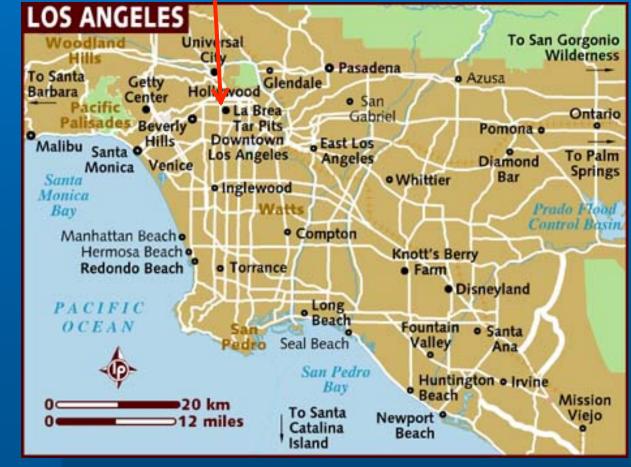




BLAST: 23 Injured; Burnoff of Gas Could Take Months, Fire Dept. Says



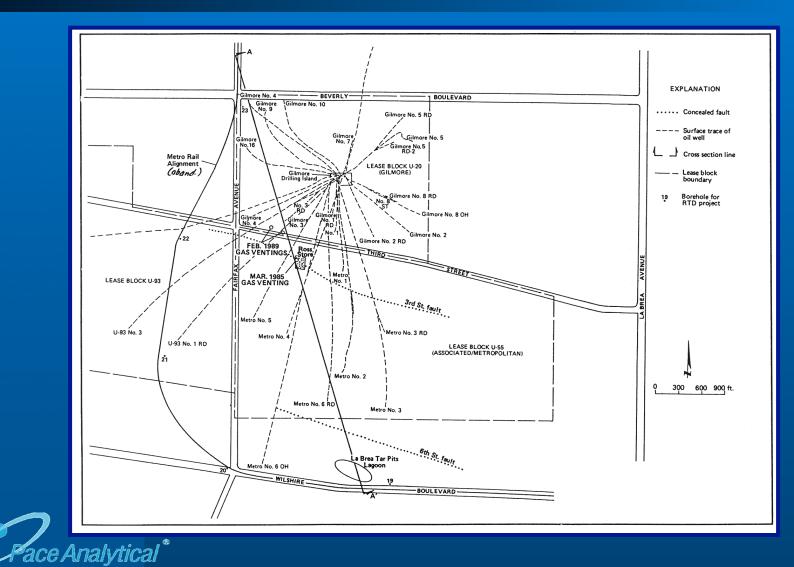
## Fairfax District, Los Angeles



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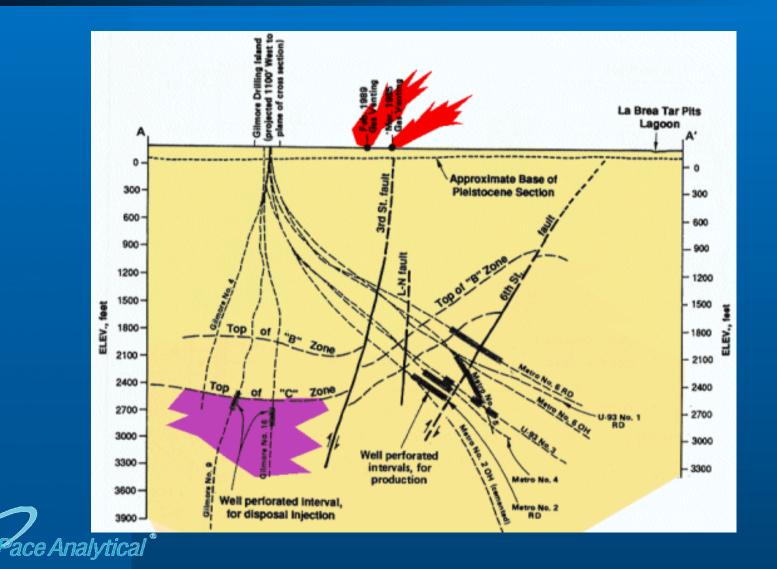


## Ross Store, Fairfax District





## Ross Store, Fairfax District





## Sources of Methane

#### BIOGENIC METHANE

- Marsh gas, methane hydrates
- Landfill gas
- Degradation of petroleum products
- THERMOGENIC METHANE
  - Pipeline gas
  - Migrated petroleum gas





**Biogenic Methane Formation** 









**Thermogenic Methane Formation** 

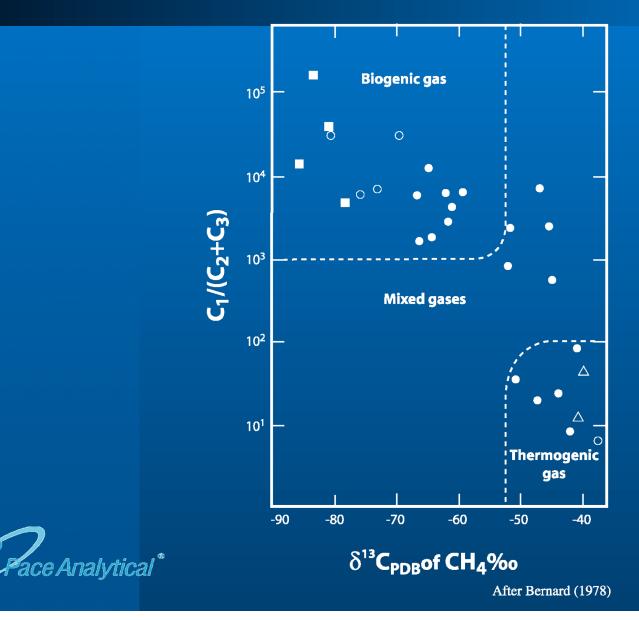
**Mature Source Rock** 

Kerogen 
$$\longrightarrow$$
  $CH_4 + C_2H_6 + C_3H_8$ , etc.

### **Overmature Source Rock**

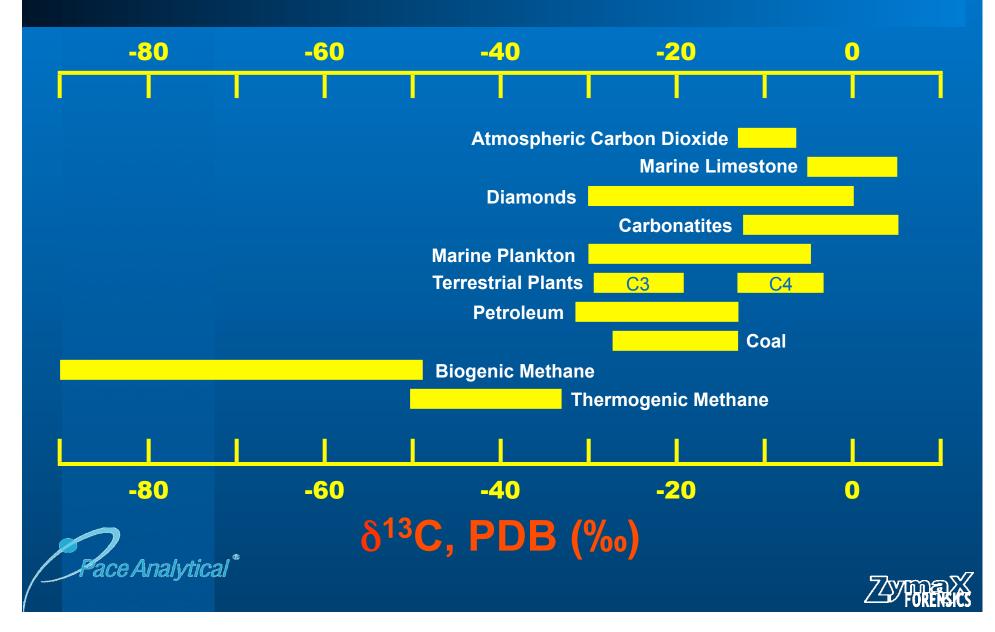
Petroleum Oil Petroleum Gas Petroleum Gas Petroleum Gas

## Gas Composition and Isotope Ratio

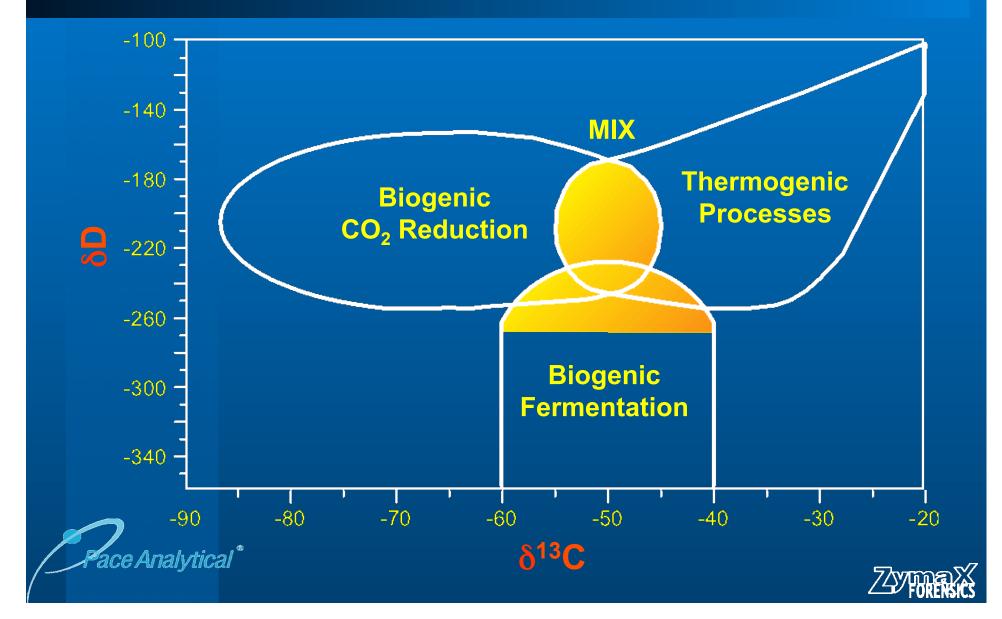




# δ<sup>13</sup>C Values of Geochemical Materials



## Methane Isotope Ratios



## **Complications: Methane Oxidation**



### • OFTEN OCCURS IN THE NEAR SURFACE

### • CAN ALTER ISOTOPIC SIGNATURE





### Methane Oxidation

ce Analvtical "

<sup>12</sup>C – X Bonds are weaker than <sup>13</sup>C – X Bonds

In a chemical reaction, <sup>12</sup>C – X Bonds break faster than <sup>13</sup>C – X Bonds

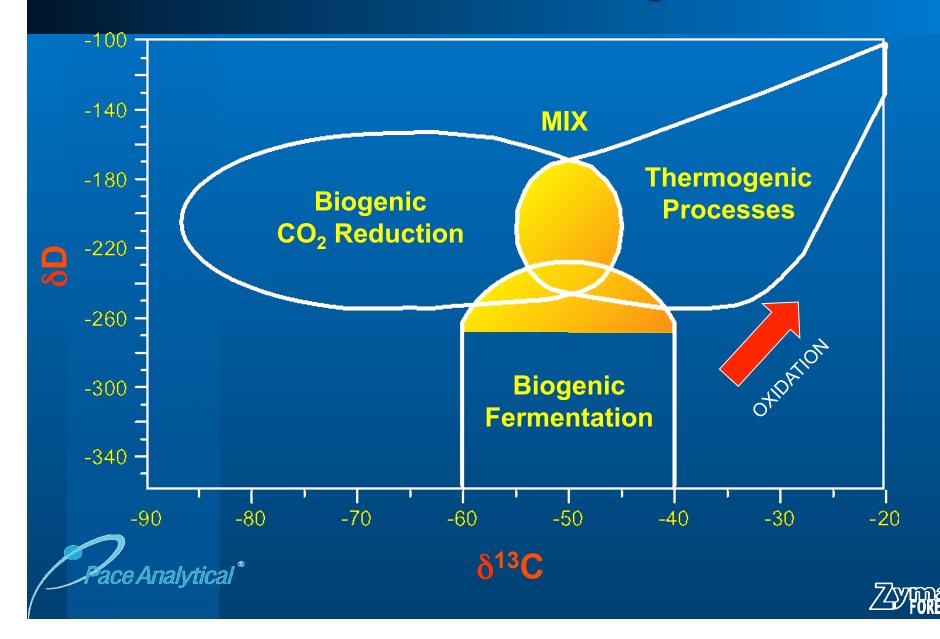
If reaction proceeds to completion – all CH<sub>4</sub> goes to CO<sub>2</sub> isotope ratio in product is the same as in starting material

If reaction is partially completed, <sup>12</sup>C is concentrated in product, and <sup>13</sup>C is concentrated in starting material

Starting material (CH<sub>4</sub>) changes to lower  $\delta^{13}$ C values



## Alteration of Methane Isotope Ratios



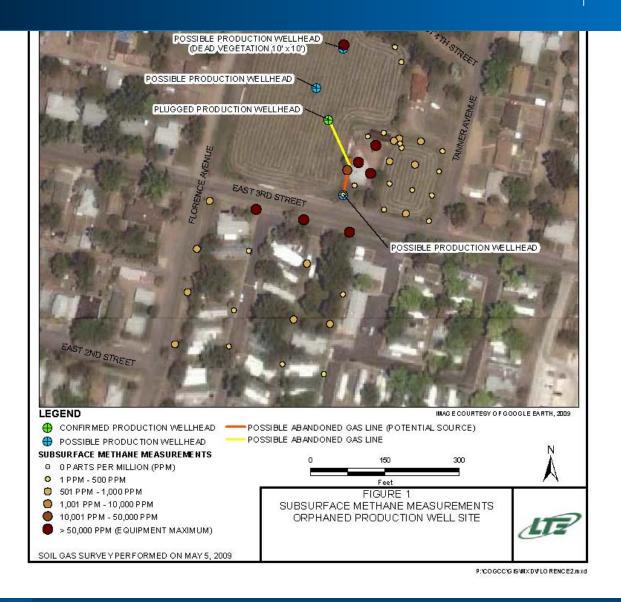
# Case Study: Origin of Gas Seeps





## Gas Seeps







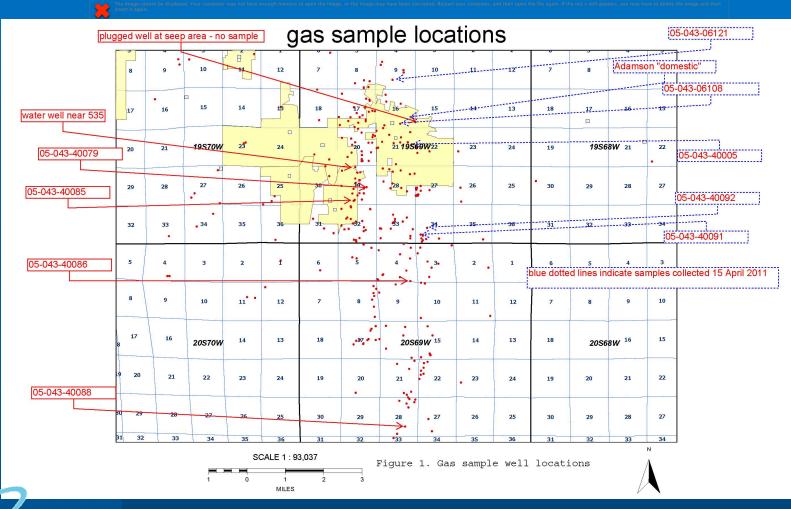
## Seep Sample locations

88





## Oil and Gas Production Wells

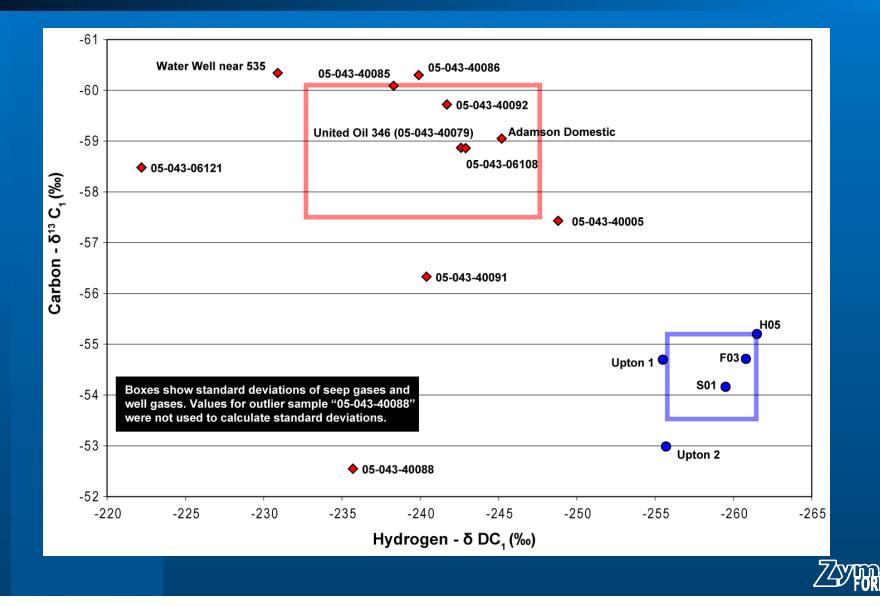




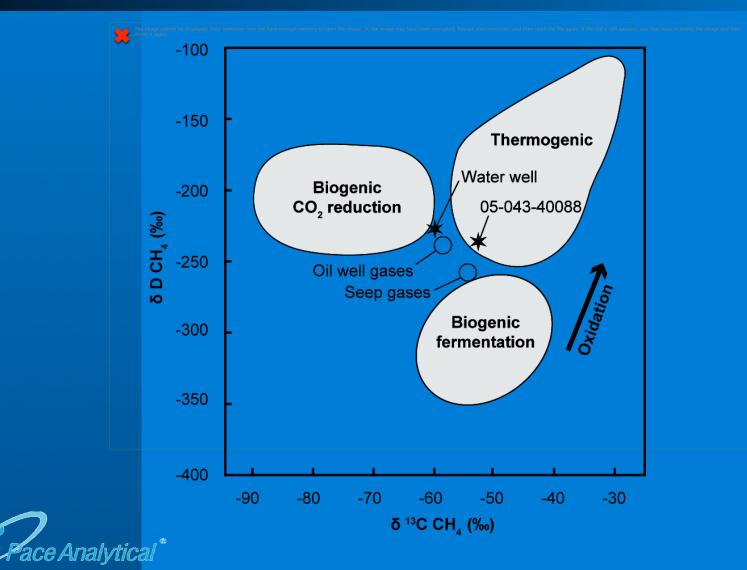
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## Methane Isotope Ratios





## Methane Genetic Fields





## Gas Explosion Los Angeles

Explosion in an apartment underground parking

#### Residual gas was collected

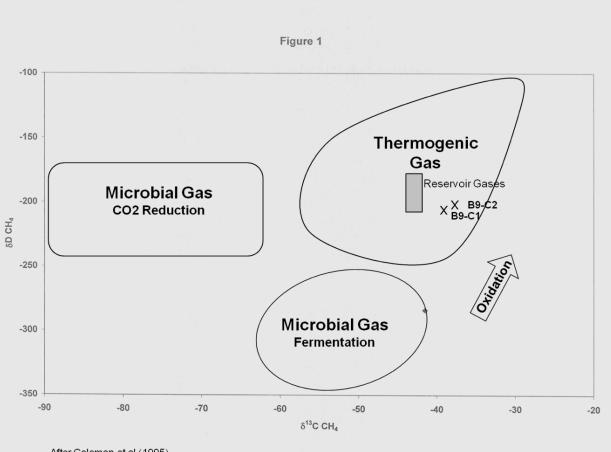
ZymaX ID	Sample ID	δ <sup>13</sup> C	δD	ppm	ppm
		CH₄	CH4	CH4	C2-C5
42122-2	#369 from B9-C2	-38.2	-199.1	8105	5.1
42122-3	#339 from B9-C1	-39.4	-203.3	2095	15.9





# Gas Explosion Los Angeles





After Coleman et al (1995)

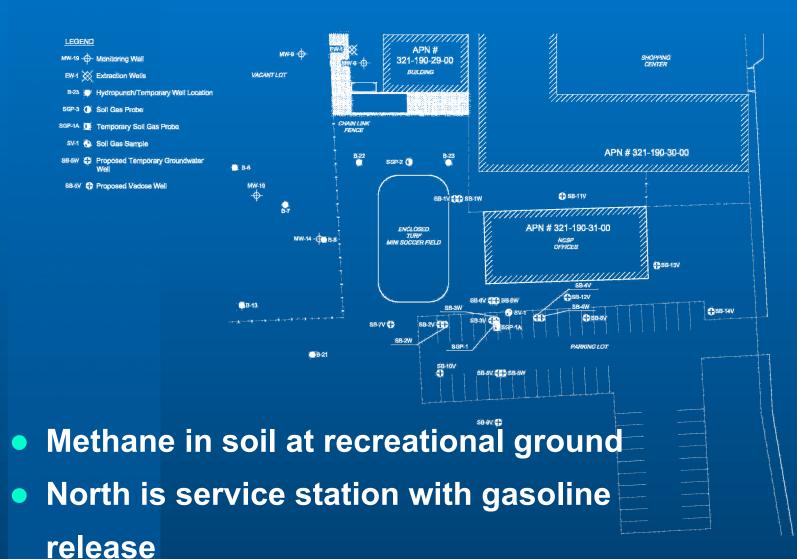


### Marsh Gas







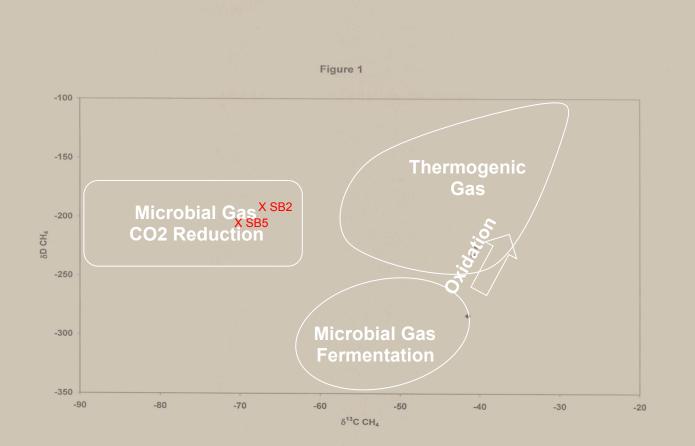




Sample ID	CH <sub>4</sub> δ <sup>13</sup> C	CO <sub>2</sub> δ <sup>13</sup> C	CH₄ δD
SB-6V-5 (8/24/11)		-25.2	
SB-5V-5 (8/24/11)		-23,9	
SB-2V-5 (8/24/11)	-66.9	<sup>°</sup> -14.8	-189.9
SB-10V-5 (8/26/11)		-21.1	
SB-2V-5 (8/26/11)	-67.2	-15.2	-191.5
SB-8V-5 (8/26/11)	-70.1	-22.5	-202.3







After Coleman et al (1995

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Sample ID	CH <sub>4</sub> δ <sup>13</sup> C	CO <sub>2</sub> δ <sup>13</sup> C	CH₄ ŏD
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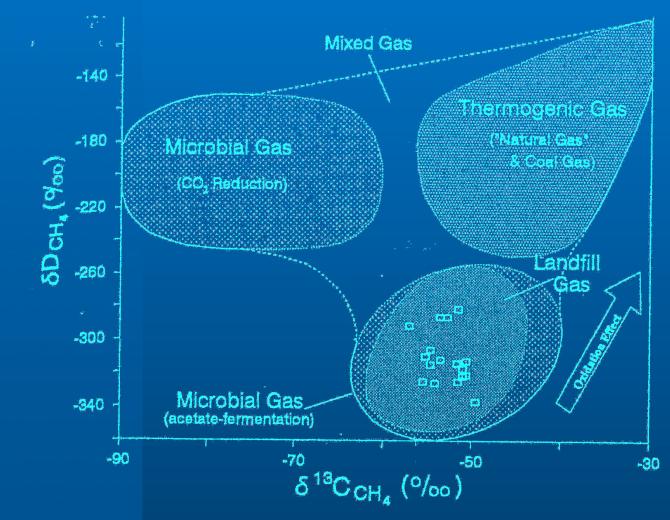
Samples with the highest CH<sub>4</sub> concentrations (SB2V) contain the heaviest CO<sub>2</sub>

Consistent with CO<sub>2</sub> reduction to CH<sub>4</sub>



## Landfill Gas







Hackley et al (1996)

### Landfill Gas Characteristics



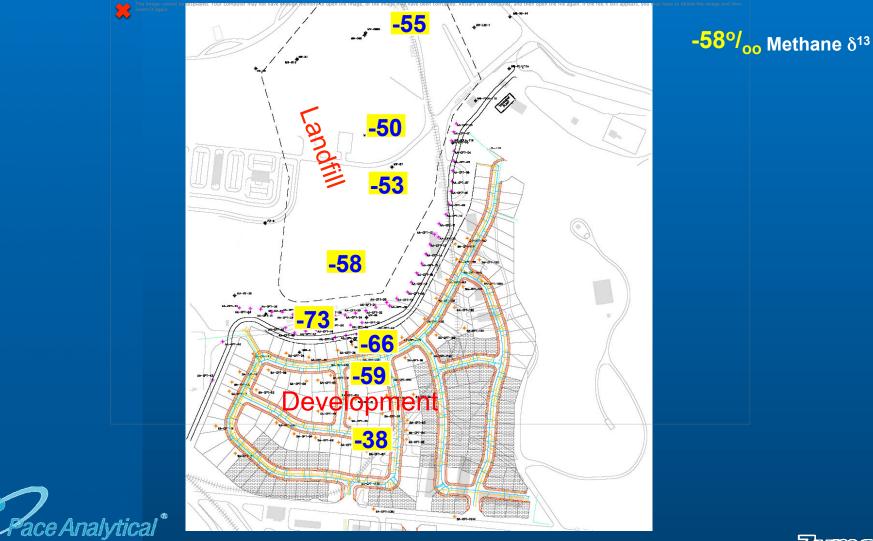
### ABUNDANT CO<sub>2</sub>

OFTEN HIGH TRITIUM CONTENT IN METHANE



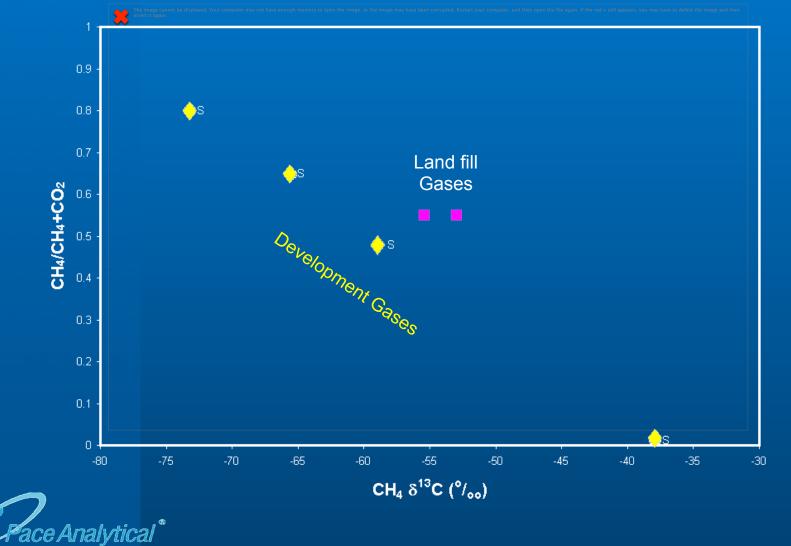


## Landfill Site



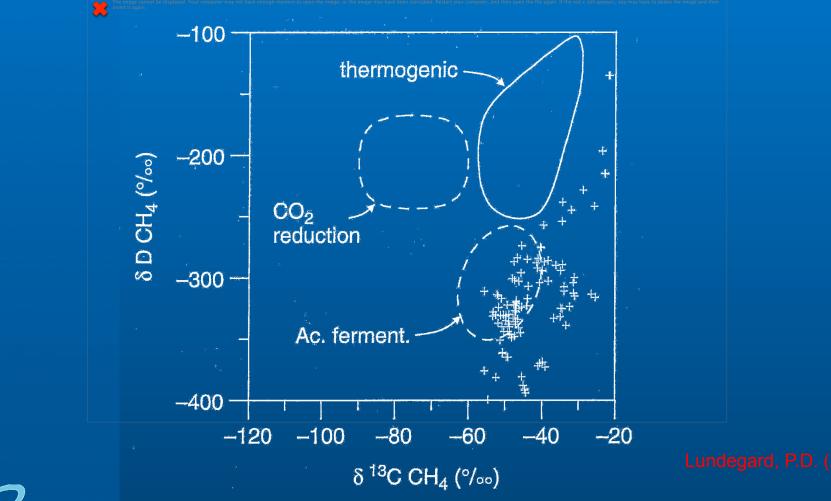


## Soil Gas Composition





## Methane from Hydrocarbon Degradation







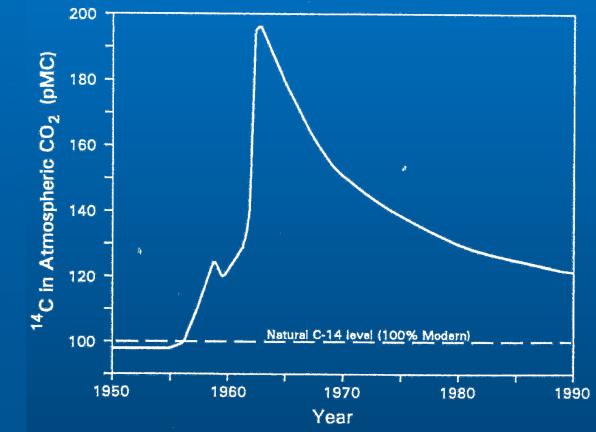
# <sup>14</sup> C Analysis

- <sup>14</sup>C Activity
- Percent Modern Carbon
- THERMOGENIC METHANE: 0 pMC
- **BIOGENIC METHANE** 
  - Marsh gas: ~30 ~120 pMC
  - Landfill gas: ~120 ~150 pMC
  - Degradation of petroleum products: 0 pMC





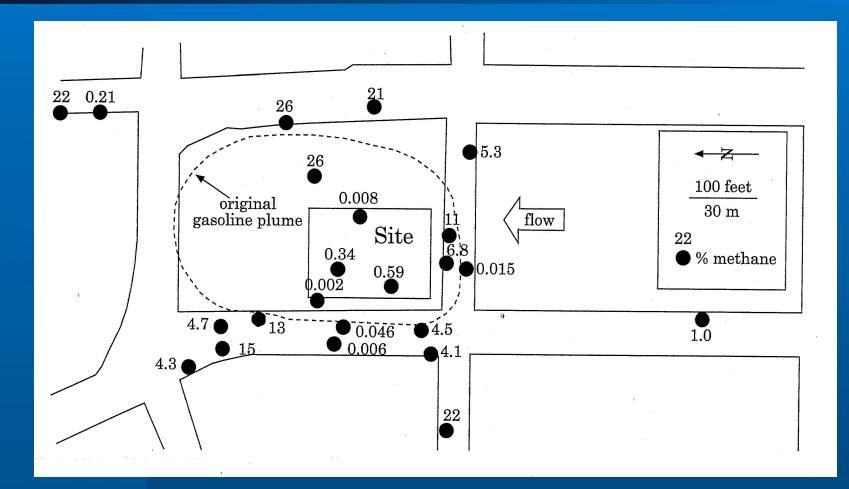
# <sup>14</sup> C Analysis



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### Methane at Service Station Site

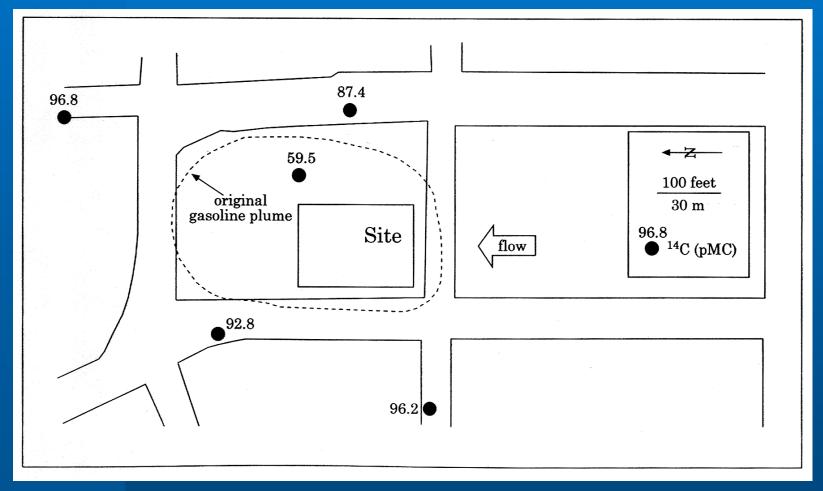


Lundegard et al, (2000)





## <sup>14</sup>C in Methane at Service Station Site

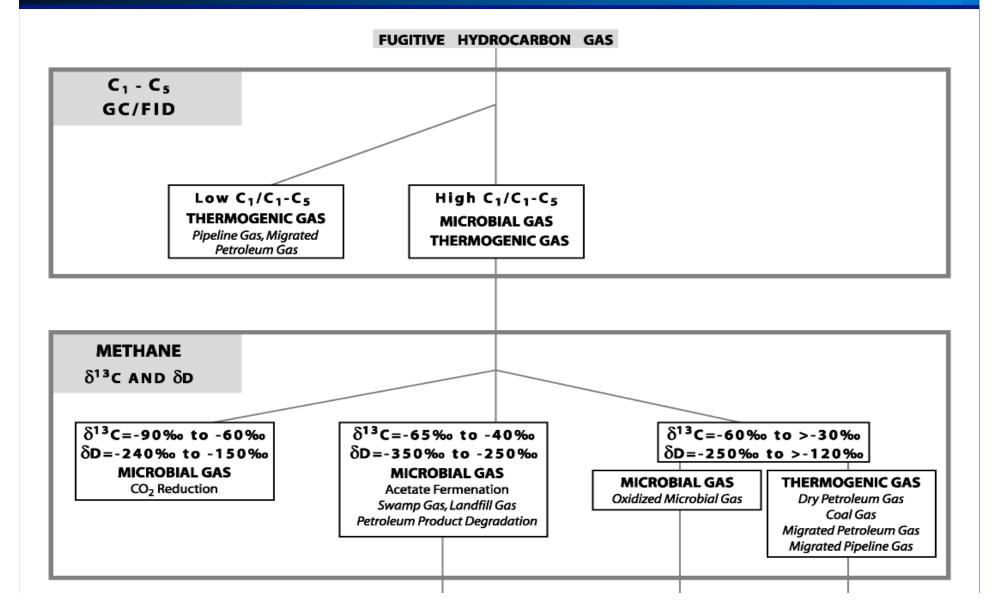


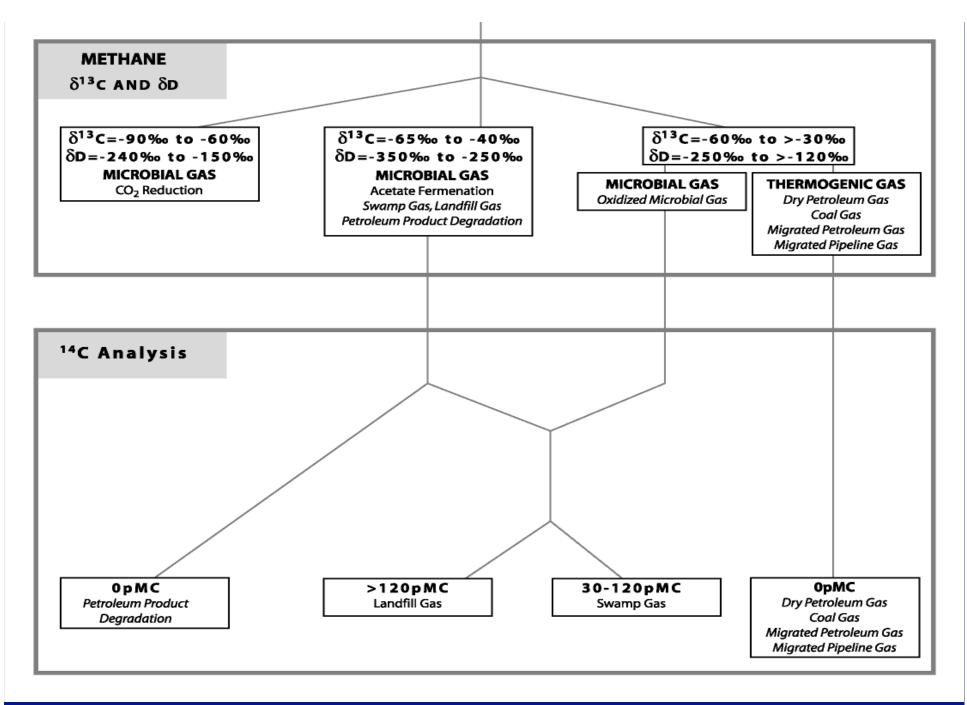


Lundegard et al, (2000)



### Source Characterization Flow Chart

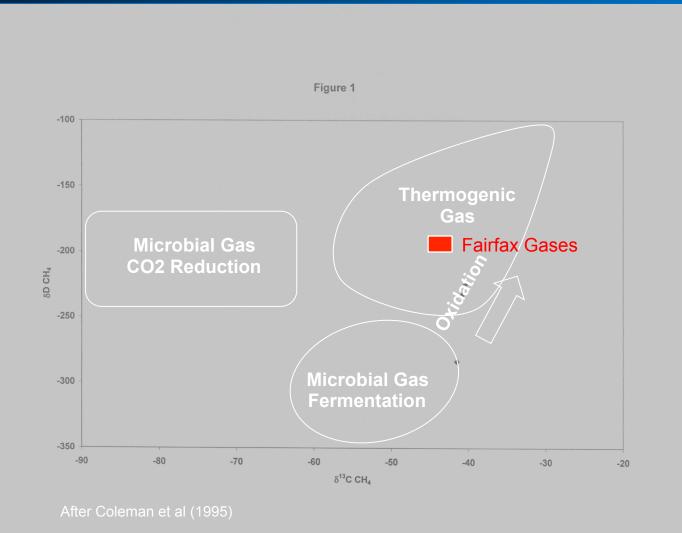






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