Environmental Forensics: Employing 2-Dimensional Gas Chromatography/Mass Spectrometry (GCxGC/MS) to Predict Environmental Weathering of Oil and Tar

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I. Introduction	II. Technology	III. Weather Maps
Pollution from petroleum sources and coal tar is the result of	Comprehensive GC/MS, GCxGC/MS, is a leading-edge separation	Understanding the impact that weathering processes have on
naturally occurring and unintended seepages; collection, transport,	technique. All of the effluent from the first-dimension column is	hydrocarbon contamination is essential in risk assessment, forensic
storage and manufacturing activities; and from incomplete	focused in many narrow fractions at regular intervals and	analyses and liability studies. GCxGC/MS offers the ability to

combustion sources. When these products are released into the environment, they are subject to a number of weathering factors including physical (evaporation, adsorption, dissolution, and emulsification), biological (microbial degradation), and chemical (photo- and oxidative degradation), all of which can significantly change the chemical composition of the source material over time. Understanding how local environments impact weathering is critical to determining if natural attenuation is sufficient to reduce pollutant impact on the environment or if active remediation is required. To make this determination, the compositional effects of dissolution, organic phase partitioning, and evaporation must be known; each of which can be examined by studying the aqueous solubility (S_w), octanol-water partition coefficient (K_{OW}), and vapor pressure (V_P) of sample components, respectively.

A novel approach for estimating these physical properties from comprehensive 2-dimensional gas chromatography/mass spectrometry (GC×GC/MS) retention data is presented. A polyparameter linear free energy relationship (LFER) model was developed to correlate each compound's retention index with the aforementioned physical properties. The model employs isovolatility curves to generate accurate and precise retention indices for families of homologous compounds, leading to better estimates of their physical properties. Findings indicate that the physical property estimates produced by this approach have the same error on a logarithmic-linear scale as previous researchers' log-log estimates, yielding marked improvement in their property estimates. A new software program was developed that automates the determination of these properties for each compound in the sample from a single GC×GC analysis of that sample. This process produces component maps that we use to discern the mechanism and progression of how a particular site weathers due to dissolution, organic phase partitioning, and evaporation resulting from local conditions. Outcomes support hazardous waste site investigation and cleanup projects by providing quantitative data of pollutants by GC×GC/MS. This approach can be used by researchers in a wide range of disciplines, including toxicology (bioaccumulation and toxicity studies) and restoration (ecological and urban planning studies).

subsequently injected onto a second capillary column.



Ion Analytics[™] software visualizes extracted ion abundances as a histogram and normalizes them to the expected ion ratio of the target compound. Compound identity confirmation is automatically achieved according to spectral deconvoltion algorithms that subtract background noise and match sample with library spectra.



visualize dissolution and evaporation processes based on retention characteristics on compounds on the 2D chromatogram.



are components volatizing?

- has dissolution stopped?
- what is <u>bioaccessible</u>/available?
- is remediation necessary?



Weathering effects were estimated via LFER based on compound's aqueous solubility (S_w) , octanol-water partition coefficient (K_{OW}) , and vapor pressure (V_P) . These physical properties were linked to a compound's retention index, allowing component maps to be drawn that predict the progression of weathering based on site conditions (e.g. arid climate vs. high soil water content).

Compound/Homologue	GC×GC/MS (µg/g)	GC/MS (µg/g)	% RPD	2.5×AL (µg/g) and ¹ (RPD)	
Naphthalene	0.7	0.7	0	5 (100)	
C ₁ Naphthalenes	8.0	10	-20		
C ₂ Naphthalenes	15	19	-21		
C ₃ Naphthalenes	11	11	0		
C ₄ Naphthalenes	4.4	2.1	110		
Fluorene	25	23	9	35 (100)	
C ₁ Fluorenes	24	21	14		
C ₂ Fluorenes	13	9.7	34		
C ₃ Fluorenes	2.9	0.8	263		
C ₄ Fluorenes	1.6	ND	N/A		
Phenanthrene/Anthracene	93	84	11	355 (100)	
C ₁ Phenanthrenes	78	87	-10		
C ₂ Phenanthrenes	20	24	-17		
C ₃ Phenanthrenes	5.5	6.8	-19		
C ₄ Phenanthrenes	1.3	0.5	160		
Benzo(a)anthracene/Chrysene	29	34	-15	0.25 (60)	
C ₁ Chrysenes	27	37	-27		
C ₂ Chrysenes	3.0	3.3	-9		
C ₃ Chrysenes	0.8	ND	N/A		
C ₄ Chrysenes	ND	ND	N/A		
Pyrene/Fluoranthene	73	72	1	262.5 (100)	
C ₁ Pyrenes	61	76	-20		
C ₂ Pyrenes	17	22	-23		
C ₃ Pyrenes	4.2	1.9	121		
C ₄ Pyrenes	1.1	ND	N/A		
C ₁ 2-Ring PASH	2.0	2.3	-13		
C ₂ 2-Ring PASH	3.6	4.3	-16		
C ₃ 2-Ring PASH	4.3	4.7	-9		
C ₄ 2-Ring PASH	1.4	1.3	8		
Dibenzothiophene	22	20	10		
C ₁ 3-Ring PASH	30	29	3		
C ₂ 3-Ring PASH	22	19	16		
C ₃ 3-Ring PASH	9.7	7.3	33		
C ₄ 3-Ring PASH	1.9	0.6	217		
C ₁ 4-Ring Fused PASH	17	14	21		
C ₂ 4-Ring Fused PASH	7.3	6.6	11		
C ₃ 4-Ring Fused PASH	0.7	ND	N/A		
C ₄ 4-Ring Fused PASH	0.6	0.8	-25		
C1 4-Ring Condensed PASH	7.3	6.1	20		
C2 4-Ring Condensed PASH	3.5	3.2	9		
C. 4-Ring Condensed PASH	6.7	6.0	12		

Target Compound Identities Confirmed WhenEquations Produce Scan-to-Scan Relative Errors Less Than Analyst Established Values

The combination of spectral deconvolution software with modern fast-scanning quadrupole mass spectrometers has allowed the first quantification of pollutants by GCxGC/qMS. A minimum of three modulations-per-compound and five scans-per-modulation was found to produce results in agreement with GC/MS analysis of parent PAH within 20% RPD, however sensitivity was increased 5fold and GCxGC/MS provided much higher resolution.





GCxGC/MS was used to create the 3D component maps for a "fresh" coal tar, collected from an underground storage tank protected from the environment and from soil contaminated with coal tar outside of the tank. The figures depict changes in volatility and solubility for the sample as it undergoes weathering

References

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	RF		r ²		LOD (ng/g)	
PAH	1D	Х	1D	X	1D	Х
Naphthalene	0.89	1.04	1.00	0.99	56.7	5.7
Acenaphthylene	1.18	1.19	0.99	0.99	69.8	10.7
Acenaphthene	0.86	1.09	1.00	0.99	24.3	5.7
Fluorene	0.92	0.65	0.99	1.00	171.0	7.5
Dibenzothiophene	0.77	0.68	1.00	1.00	91	19.3
Phenanthrene/Anthracene	0.88	0.93	0.99	1.00	26.5	7.3
Fluoranthene	0.80	0.88	0.99	1.00	92.5	5.5
Pyrene	0.88	0.95	1.00	0.99	105.4	6.3
B(a)A/Chrysene	0.75	1.10	0.99	1.00	86.4	22.3
Benzo(b/k)fluoranthene	0.74	0.73	0.99	0.99	125.9	13.5
Benzo(a)pyrene	0.75	0.74	0.99	0.99	102.0	17.7
Indeno(1,2,3-c,d)pyrene	0.76	0.76	0.99	1.00	149.1	34.2
Dibenz(a,h)anthracene	0.66	0.53	0.99	1.00	161.6	28.0
Benzo(g,h,i)perylene	0.79	0.76	0.99	1.00	117.1	35.8

