

Analysis of Dissolved Hydrocarbon Gases in Water – Pitfalls and Improvements

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Why measure hydrocarbon gases in groundwater?

- Oil & gas application
 - Predrill site characterization
 - [°] Establish baseline for comparison
 - [°] Find existing issues
 - Post drill site characterization
 - [°] Assess potential changes
- Remediation monitoring
 - Reducing conditions indicator
 - [°] Active reductive dechlorination
 - Monitored natural attenuation
- Other applications?





How to measure hydrocarbon gases in groundwater?

- Headspace GC / FID
 - ~ RSK-175 Rev 0, 1994
 - ~ J Chrom Sci Kampbell, Vandergrift, 1998
 - ~ M E E Analysis Guidance, US EPA Region 1, 2002
 - ~ RSK-175 Rev 2, 2004
 - ~ RSK-175 Rev 3, 2006
 - ~ RSK-175 Rev 5, 2010
 - ~ PA DEP 3686 Rev 1, 2012
- Purge and Trap
 - ~ PA DEP 9243 Rev 0, 2012

		RSKSOP-175	
		Revision No.2	
		May 2004	
		Page 1 of 14	
		Felisa Hudson	
	STANDARD OPERATING	PROCEDURE	
	Sample Preparation and Calculations in Water Samples Using a GC Headspa		
	1. Disclaimer:		
	This standard operating procedure has been prepared for	or the use of the Ground Water and	
	Ecosystems Restoration Division of the U.S. Environm		
	specifically applicable to the activities of other organiz		
	EPA APPROVED METHOD. This document has no	ot been through the Agency's peer review	
	process or ORD clearance process.		
	2. <u>Purpose</u> (Scope and Application):		
	This method is applicable to the preparation of water s	amples for determination of dissolved	
	gases. After quantitation of gas equilibrated into the prepared headspace, this method permits		
	calculation of the concentration of the dissolved gas in		
	concentrations are expressed as mg/L and µg/L of diss		
	used for determining dissolved hydrogen, methane, eth	wiene, einane, propane, butane, aceiviene,	

nitrogen, nitrous oxide and oxygen. The number of analyses that can be performed in an eight



- Lab to lab implementation variability of RSK-175 *method*
- No commercially available proficiency test samples
- Calibration standards are gas phase
 - Samples are water
 - Thus standards and samples are not handled identically
- Analyte loss during sample preparation
- Maintaining analyte representativeness during sampling

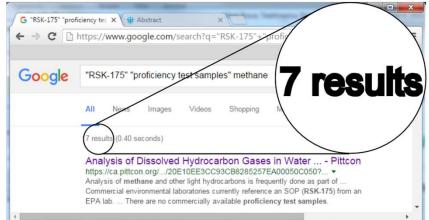


- Lab to lab implementation variability of RSK-175 method
 - Critical process elements
 - [°] Sample transfer between containers
 - Headspace development
 - [°] Equilibration
 - time
 - temperature
 - mixing
 - ~ Process options
 - ° Manual
 - [°] Semi-automated
 - ^o Fully automated



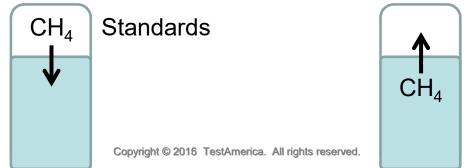


- No commercially available proficiency test samples
 - No catalog items
 - No custom PTs available either
 - ~ Limited water solubility
 - Limited organic solvent solubility
 - Limited stability in water
 - Complex preparation process
 - = expensive to produce, short shelf life & lower quality
 - >> No independent check on accuracy





- Calibration standards are gas phase
 - ~ Samples are water
 - Thus standards and samples are not handled identically
 - Equilibrium goes in one direction for samples and the opposite for standards
 - Non-equilibrium conditions
 - High biased standard response
 - Low biased sample response
 - = low biased calculated sample concentration



Samples



- Analyte loss during sample preparation
 - ~ Volatilization during open transfer
 - ° Too slow or turbulent
 - PA DEP study demonstrated this can be done right





- Analyte loss during sample preparation
 - ~ Degassing using syringe transfer
 - ° Too fast



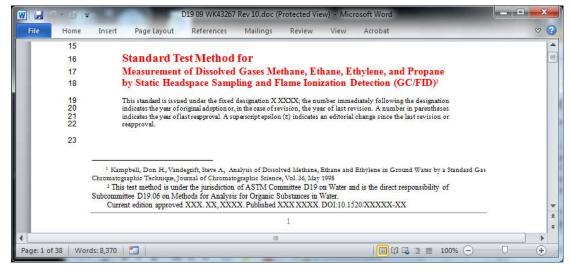


- Analyte bias during sample preparation
 - Losses or increases during headspace development
 - [°] Transfer water too slow, increases air exposure low bias
 - ° Transfer HS gas too slow, purges water high bias
 - ~ Pierced septum leaks
 - ^o Headspace / analyte loss while waiting in autosampler





- Multi-lab validated method
 - On track ASTM D19.09 WK4367
 Rev 10 draft Feb 17, 2016
 - [°] Single lab validation at present
 - [°] Fully automated process with manual option
 - Anne Jurek @ EST Analytical
 - Marcellus Shale Coalition
 - Revied draft
 ASTM method
 - Identified inter-lab variability as a critical issue





Methane 23.2 mg/L

Ethane 62.0 mg/L

Ethene 149.0 mg/L

- Water based calibration standards and QC samples
 - ~ In-lab prep of single analyte saturated water standards
 - [°] Use literature values at known temperature and pressure
 - PA DEP 3686 Rev 1, 2012
 - ASTM D19.09 WK4367
 - Immediate dilution to working standards
 - Propane 76.7 mg/L - Store in VOC vials - no headspace, 14 days
 - ~ Commercial providers attempting to develop standards
 - Not ready yet
 - Store stock standard in gas tight syringe with valve
 - Hours of stability
 - Not weeks

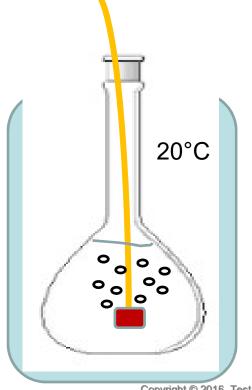




- Water based calibration standards and QC samples
 - What hasn't worked (methane)
 - Inject methane bubble into vial filled with water
 - [°] Inject methane bubble into vial filled with methanol
 - [°] Refrigerate to improve solubility
 - Apply pressure with syringe plunger



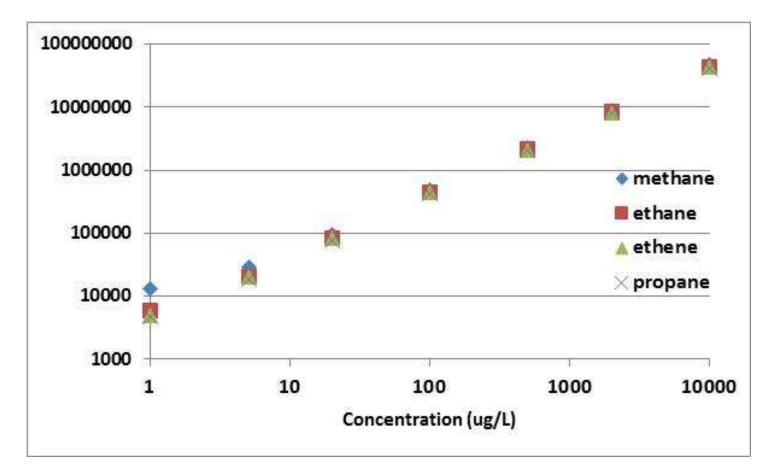
- Water based calibration standards and QC samples
 - ~ PA DEP 3686 Rev 1, 2012
 - ~ ASTM D19.09 WK4367







- Water based calibration standards
 - ~ 4 analytes in the same standard

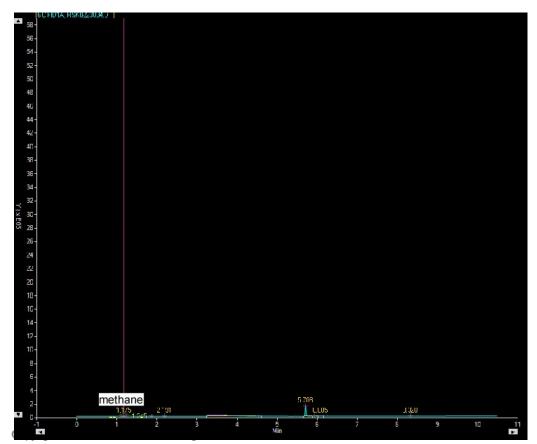




- Water based calibration standards
 - Individual analyte ICVs (second source)
 - -~100 ug/L
 - Methane 103%R
 - Ethane 101%R
 - Ethene 83%R
 - Propane 91%R

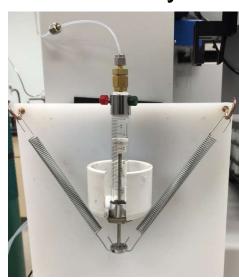


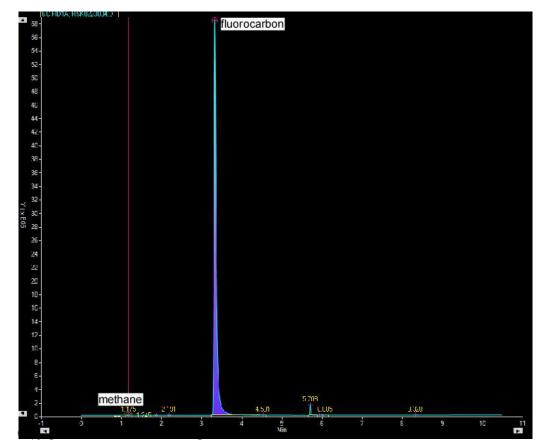
- Surrogate monitor sample prep & analysis quality
 - ~ How to know when you have a good headspace injection?
 - Int. Std.
 - ~ Surrogate
 - ~ ASTM
 - ° MTBE





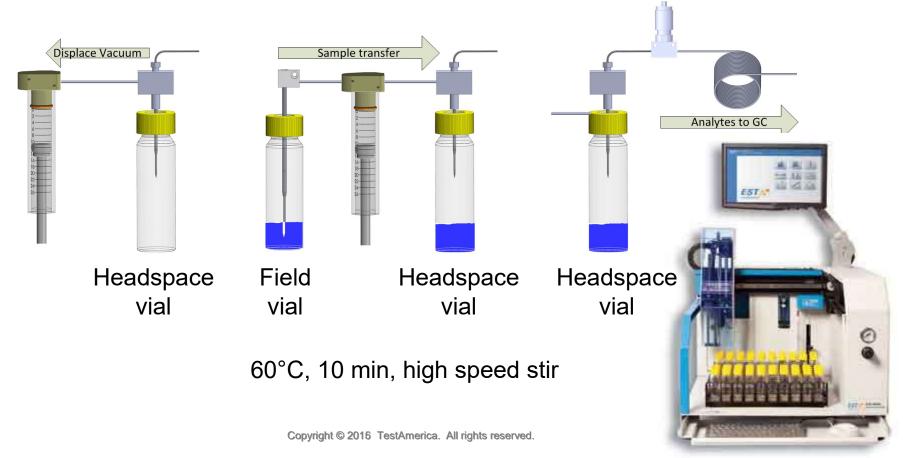
- Surrogate monitor sample prep & analysis quality
 - How to know when you have a good headspace injection?
 - Surrogate in water?
 - Automated delivery?





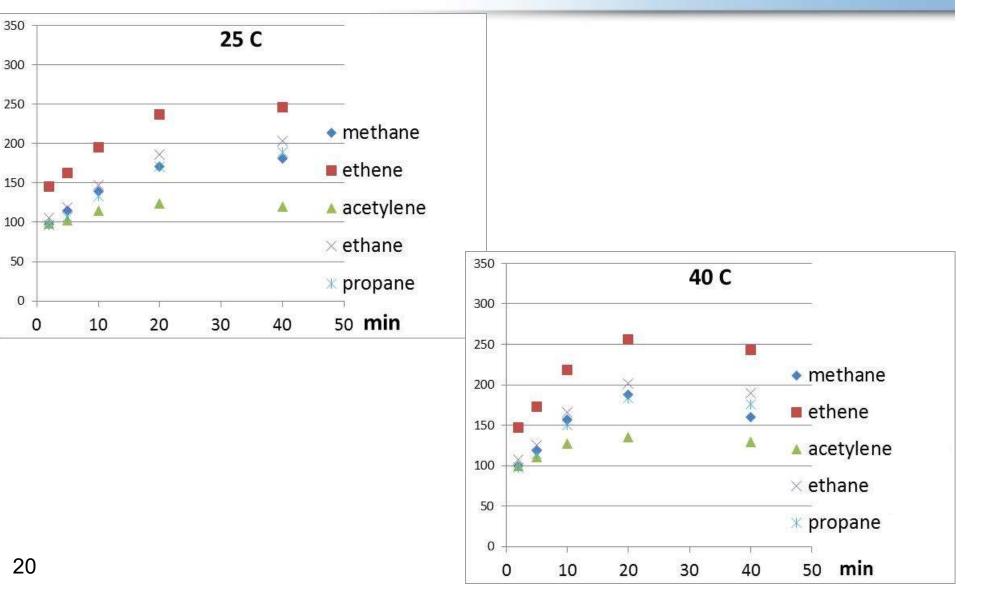


- Automated headspace preparation
 - ~ EST Analytical LGX 50



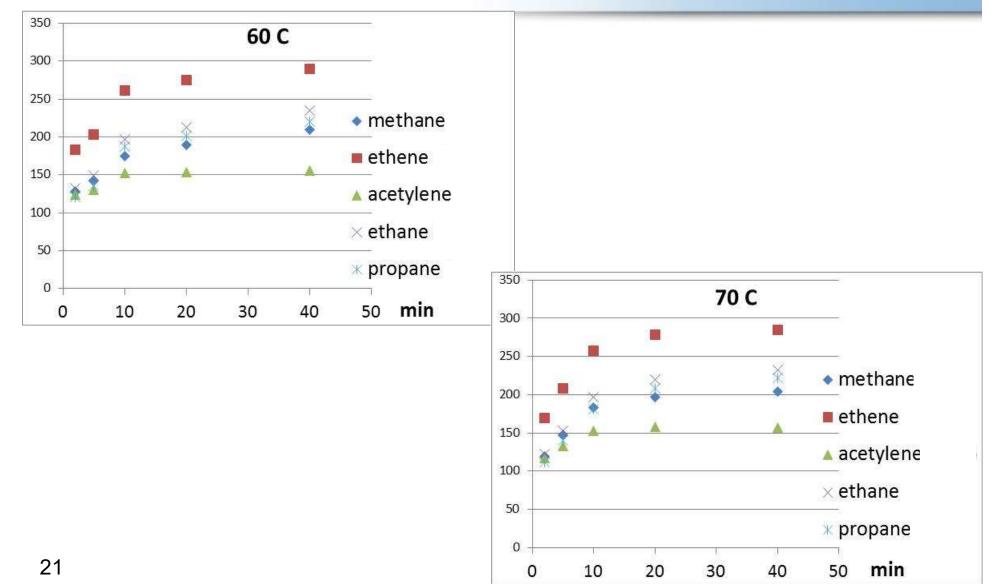


Headspace Equilibrium Optimization





Headspace Equilibrium Optimization





Headspace Equilibrium Optimization

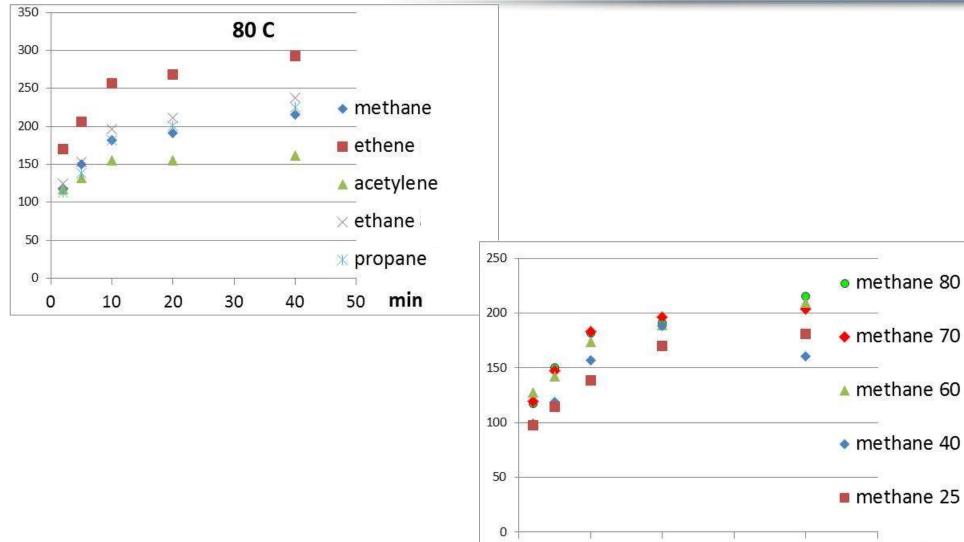
20

10

30

40

50 min



0



Headspace Equilibrium Optimization

- Stir bars
 - Standard mini (12 mm)
 - ~ Spinfin (12.7 & 19 mm)
 - ~ Starburst (9.5 and 19 mm)
- Short Equilibration
 - ~ 60°C
 - ~ 3 min
 - ~ Maximum stir speed





Selecting a Stir Bar

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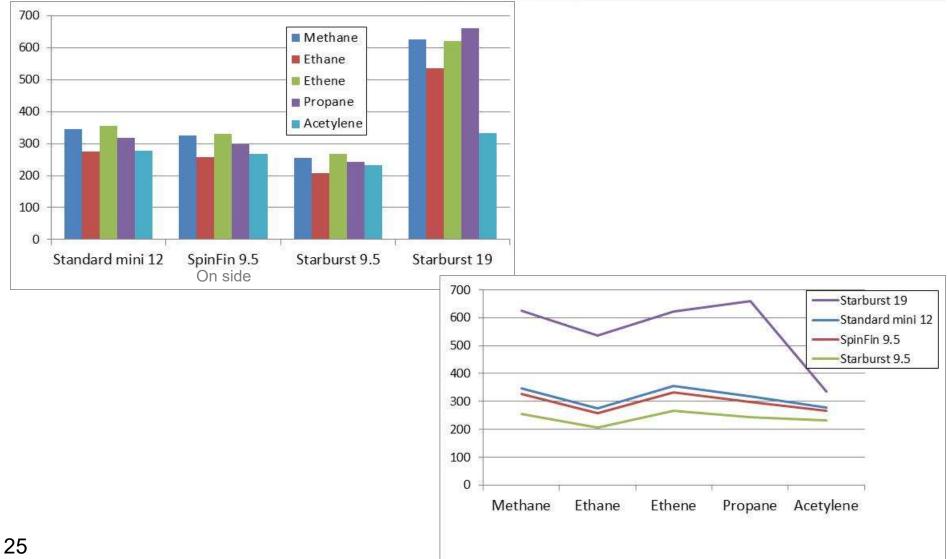


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Selecting a Stir Bar

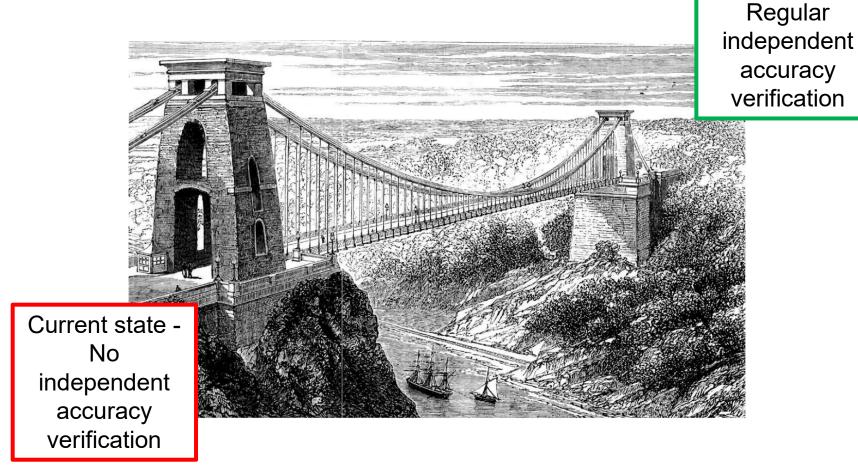
THE LEADER IN ENVIRONMENTAL TESTING





Future state -

• External Proficiency Test samples



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Summary

- Multi-lab validated method
 - ~ On track ASTM D19.09 WK4367
- Water based calibration standards and QC samples
 - ~ Open opportunity for commercial provider
 - Lab based saturated stock water solutions
- Surrogate monitor sample prep & analysis quality
- Automated headspace preparation
 - ~ 60°C, 10 min, high speed stir with 19 mm starburst
- External Proficiency Test samples
 - ~ Open opportunity for commercial provider



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