

Modern Approaches for PFAS LC-MS/MS Analysis in Aqueous and Solid Matrices

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Overview

PFAS Background

Case Studies

1. SPE + LC-MS/MS
2. Large Volume Direct Inject + LC-MS/MS
3. QuEChERS from Food + UHPLC-MS/MS
4. Online SPE + LC-MS/MS
5. QuEChERS from Sediment + LC-MS/MS

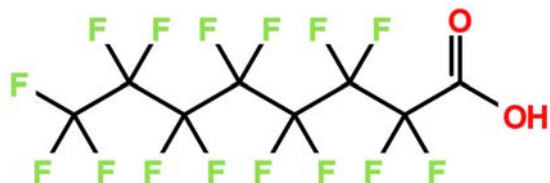
Conclusion

Resources

Acknowledgements



PFAS



- PFAS – Per and Polyfluorinated Alkyl Substance
- Repellent properties popular for consumer products like surface treatment protection, paper protection, and performance chemicals (i.e. firefighting foams)
- Trace levels found throughout global water sources
- Chemically stable, low reactivity, and resistant to degradation in aqueous environments - Important for use in a wide range of products and industrial applications
- Human exposure to PFAS residues linked to adverse health effects



**EPA Guidelines for PFOA and
PFOS In Drinking Water
=
More Testing**

Typical Challenges

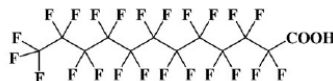
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Chemical structures of three ligands are shown:

- Acidic:** 2,4-dichlorophenyl 2-oxoacetate. SMILES: Clc1cc(Cl)ccc1OC(=O)C(=O)O
- Neutral:** N-methyl-1-(naphthalen-1-yl)carbamate. SMILES: CN(C)C(=O)Oc1cccc2ccccc12
- Basic:** 1-(2,4-dichlorophenyl)-3-(2-methylvinyl)pyrrolidine. SMILES: CC=COCc1cc(Cl)ccc1ClN1CCCC1

A collage of various fresh ingredients including carrots, salmon, shredded cheese, mushrooms, broccoli, green beans, chicken thighs, olive oil, and eggs.

System Contamination

CC(F)(F)C(F)(F)C(=O)OF(C(F)(F)F)(C(F)(F)F)C(F)(F)FC(F)(F)FC(F)(F)FS(=O)(=O)c1ccc(S(N)(=O)=O)cc1

PFOS linear isomer



PFOS branched isomer

The Strata®-X Product Line

Solid Phase Extraction

In SPE, a support particle is modified with different functional groups

- Silica or polymeric particles
- Wide range of functional groups (RP, IEX, NP)

Target analytes bind to the media

- Matrix interferences are washed away using different washing protocols

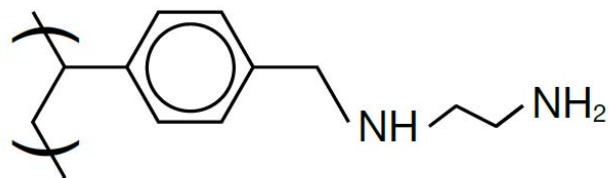
The key distinction is that you optimize your method to target & recover your analytes

- More selective than QuEChERS



Strata[®]-X-AW and XL-AW Weak Anion-Exchange

3 Mechanisms of Retention



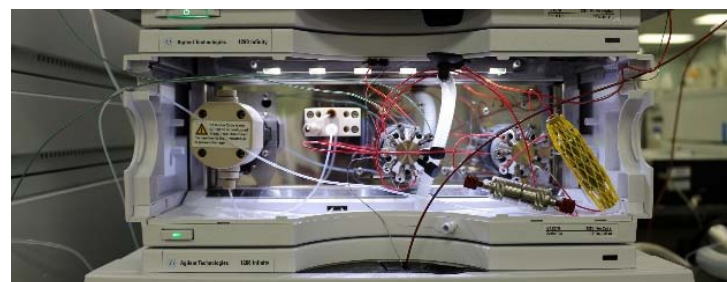
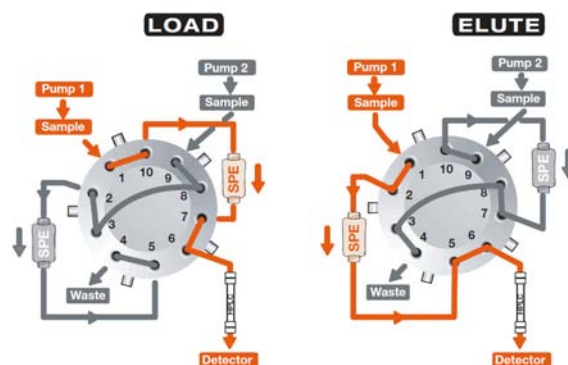
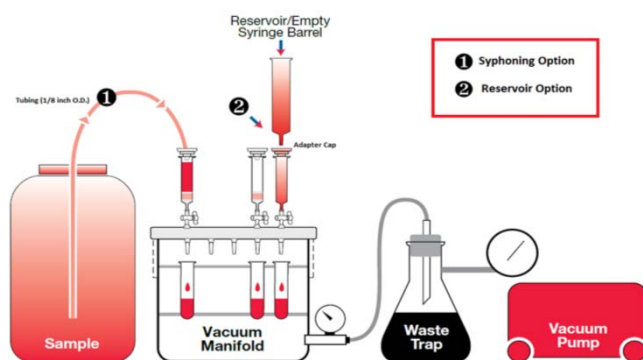
- Weak Anion-Exchange
- π - π Bonding
- Hydrophobic Interaction



Select Your Particle and Pore Size

	Strata -X-AW, 33 μ m, 85 Å	Strata-XL-AW, 100 μ m, 300 Å
High Concentration Samples	✓	
Small Target Analytes (< 10 kDa)	✓	
Large Target Analytes (> 10 kDa)		✓
Large Volume Samples		✓
Viscous Samples		✓

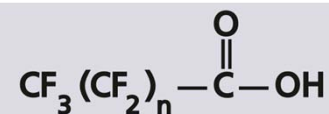
Large Volume Processing and Online SPE



Solid Phase Extraction

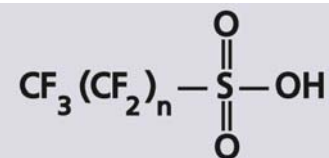
EPA method 537

1. Reversed phase retention (Strata[®]-X)

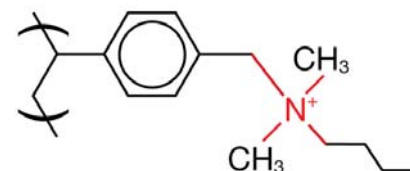


Mixed Mode Anion-Exchange

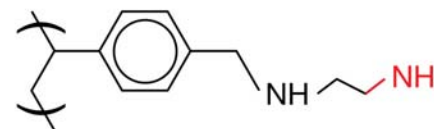
1. Anion-Exchange + Strata-X



Strata-X-A Strong Anion-Exchange



Strata-X-AW Weak Anion-Exchange



Solid Phase Extraction

Strata[®]-X-AW Weak Anion-Exchange

Condition: 3 x 5mL methanol

Condition: 2 x 5mL water

Load: 125 mL sample

Wash: 2 x 5 mL water

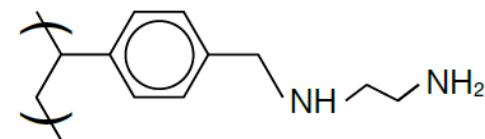
Bottle rinse: 5 mL 0.5% NH₄OH in methanol

Elute: 3 x 5 mL 0.5% NH₄OH in methanol

Evaporate to dryness

Reconstitute with 50/50 methanol/water

3 Mechanisms of Retention



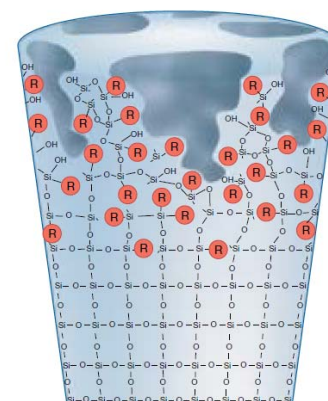
- Weak Anion-Exchange
- π - π Bonding
- Hydrophobic Interaction



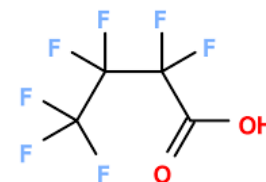
Gemini® TWIN™ Technology

Gemini columns are rugged reversed phase HPLC columns that offer extended lifetime at extreme pH conditions and excellent stability for reproducible, high efficiency separations.

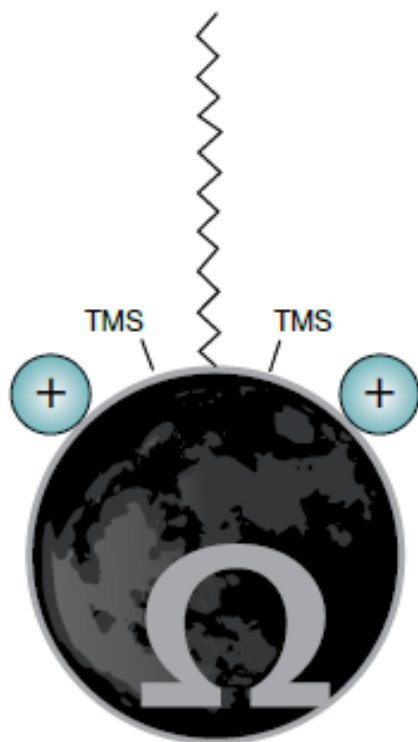
- Take full advantage of high and low pH conditions (pH 1-12) to manipulate selectivity
- Expect longer column lifetime
- For analytical and preparative separations of basic and acidic compounds



PFBA retention and good peak shape



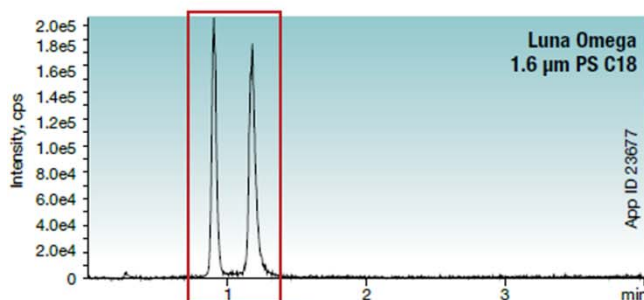
LUNA[®] Omega PS C18



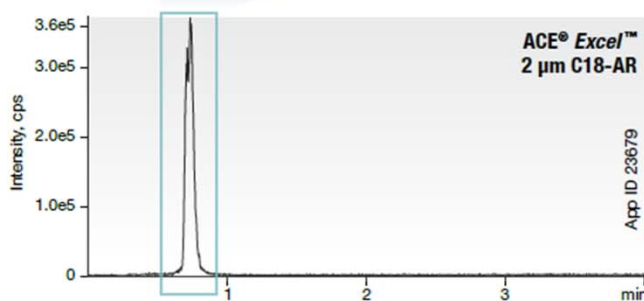
- LUNA Omega 1.6 & 5 μm Silica Based
- **3-step bonding process:**
 - Deposition of positive charge on surface
 - C18 bonding
 - TMS endcapping
- **A C18 with:**
 - Enhanced selectivity for polar acids
 - Stability in 100% aqueous mobile phases
 - Improved peak shape and loadability for bases

Enhanced Selectivity for Acids

MMA and Succinic Acid



Greater Retention and Resolution



Conditions for all columns:

Columns: Luna Omega 1.6 µm PS C18
ACE Excel 2 µm C18-AR

Dimension: 50 x 2.1 mm

Mobile Phase: A: Water with 0.1% Formic Acid
B: Acetonitrile with 0.1% Formic Acid

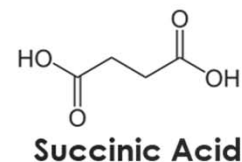
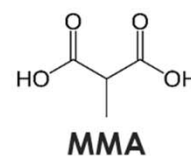
Gradient:	Time (min)	% B
	0	0
	5	50
	5.1	0
	7	0

Flow Rate: 0.5 mL/min

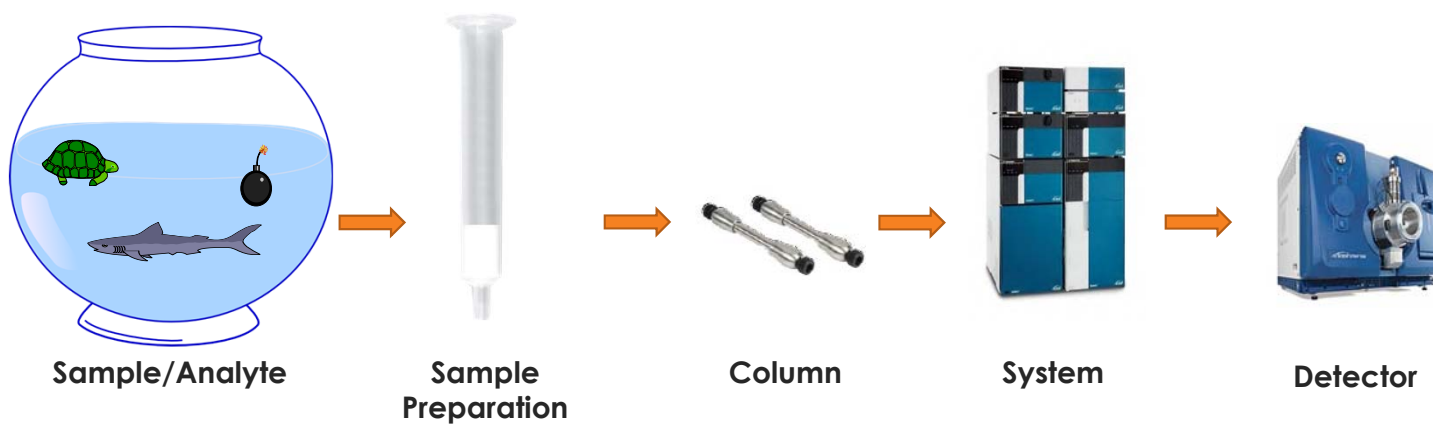
Temperature: 22°C

Detection: MS/MS (SCIEX API 4000™)

Sample: 1. Succinic acid
2. MMA



So Let's Put These Into Action!



PFAS using SPE and LC/MS/MS

Case Study 1

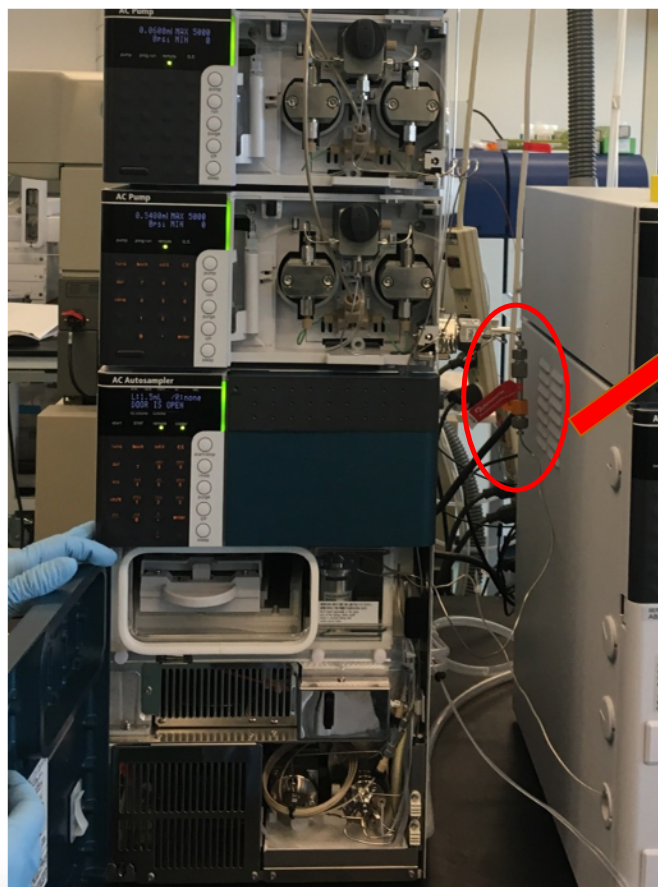


Reducing LC Contamination

- Reducing background LC contamination is critical to obtaining low detection limits
- Important to use high-purity solvents and modifiers (e.g. ammonium acetate) for mobile phases; test each solvent bottle to verify purity
- **PFAS may be present in plastic tubing, Teflon® filters**
 - Replace fluorinated tubing with PEEK
 - Remove PTFE solvent filter frits
 - Replace graphite-filled PTFE pump seals with polyethylene seals
- **However, PFAS contamination may still be present ...**



LC Contamination: Delay Column

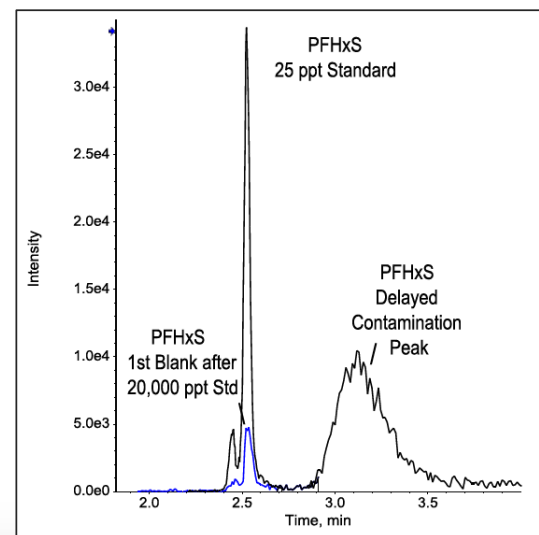


explore
LUNA[®]



LC Contamination: Reduction Strategies

- Use of “delay” or “trap” column placed between the pumps and autosampler, *upstream* of the analytical column
- PFAS leaching from the LC will be retained on the delay column and separated from the analytical peak
- NOT helpful for contamination from method blanks, contaminated standards



SPE Method Overview

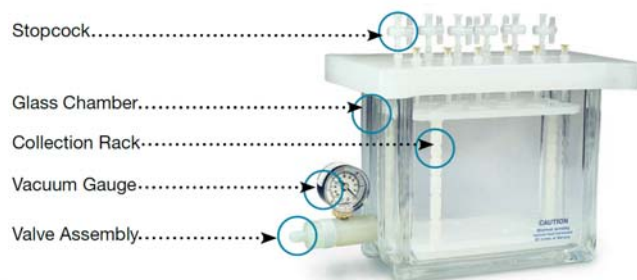
Advantage: Sample Cleanup and Concentration, compatible with DOD QSM 5.1

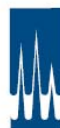
Overview:

Calibration and mass-labelled surrogate standards (i.e. isotope dilution stds) purchased from Wellington Laboratories (Guelph, ON); *wide coverage for surrogates*

Surrogate standards (25 ng) spiked into 250 mL water samples

PFAS compounds extracted and concentrated on weak anion-exchange SPE column (e.g. Phenomenex Strata[®] X-AW)





WELLINGTON Reporter

PRODUCT UPDATES FROM WELLINGTON LABORATORIES

November 17, 2016

Catalogue Number	Product (methanol)	Qty	Conc
PFAC-24PAR	Native PFAS Precision and Recovery Standard Solution (24 components)	1.2 ml	2.0 µg/ml ea
MPFAC-24ES	Mass-Labelled PFAS Extraction Standard Solution (19 components)	1.2 ml	1.0 µg/ml ea

These new solution/mixtures complement our existing line of mixed PFAS reference standards.

Catalogue Number	Product (methanol)	Qty	Conc
PFC-MXA	Native PFCA Solution/Mixture (C ₄ -C ₁₄)	1.2 ml	2.0 µg/ml ea
PFS-MXA	Native PFSA Solution/Mixture (C ₄ , C ₆ -C ₈ , C ₁₀)	1.2 ml	2.0 µg/ml ea
PFAC-MXA	Native PFCA/PFSA Solution/Mixture (10)	1.2 ml	5.0 µg/ml ea
PFAC-MXB	Native PFCA/PFSA Solution/Mixture (17)	1.2 ml	2.0 µg/ml ea
PFAC-MXC	Native PFCA/PFSA Solution/Mixture (21)	1.2 ml	2.0 µg/ml ea
MPFAC-MXA	Mass-Labelled PFCA/PFSA Solution/Mixture (9)	1.2 ml	2.0 µg/ml ea
MPFAC-C-ES	Mass-Labelled PFCA/PFSA Extraction Standard (13)	1.2 ml	2.0 µg/ml ea
MPFAC-C-IS	Mass-Labelled PFCA/PFSA Injection Standard (4)	1.2 ml	2.0 µg/ml ea
EPA-537IS	U.S. EPA Method 537 Internal Standard PDS (3)	1.2 ml	variable
EPA-537SS	U.S. EPA Method 537 Surrogate PDS (3)	1.2 ml	variable



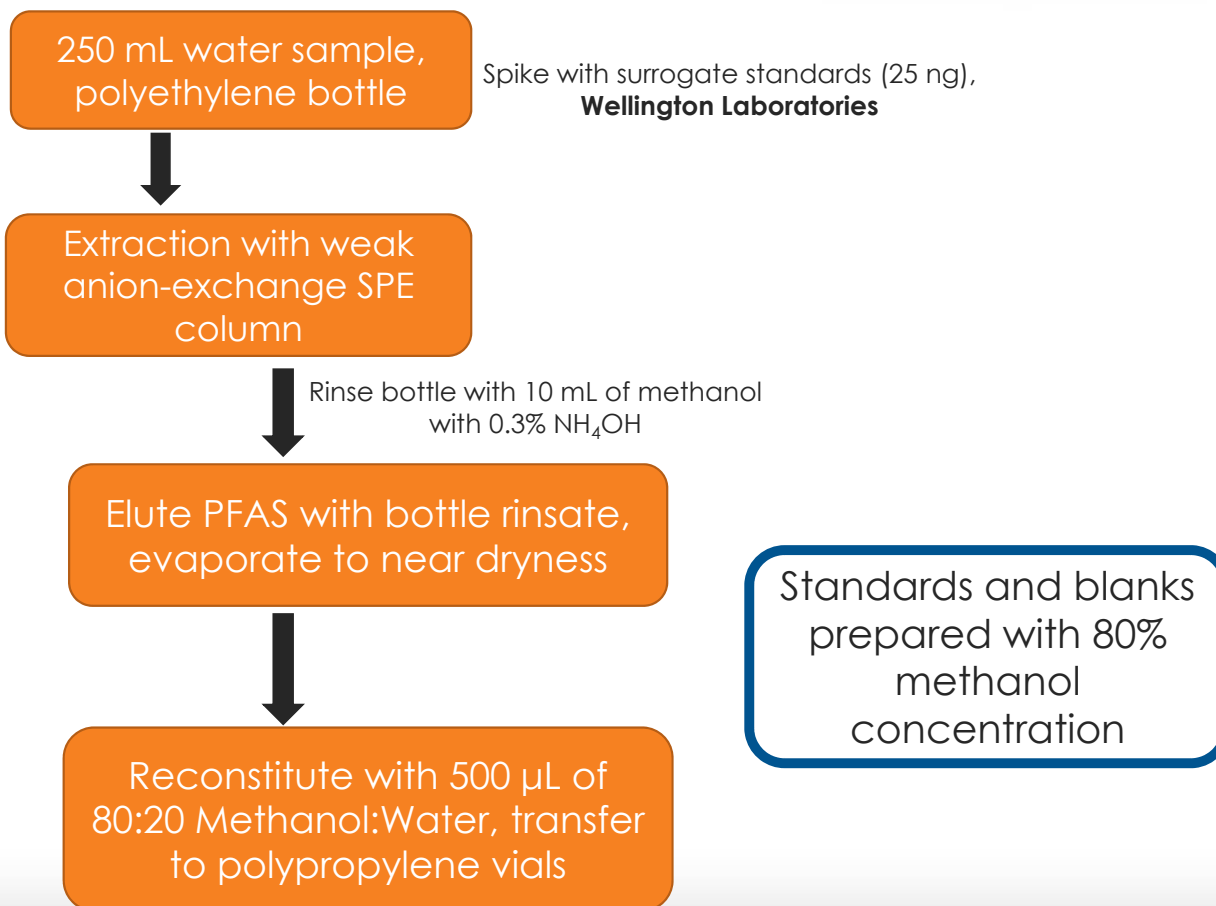
ISO 9001

Please contact your local distributor or info@well-labs.com for pricing and delivery.

Visit our website (www.well-labs.com) for a complete listing of our new products.

WELLINGTON LABORATORIES

SPE Method: Flowchart



LC Conditions For: SPE Method

Column: Gemini[®] 3 μ m C18
Dimensions: 50 x 2 mm
Part No.: 00B-4439-B0
Mobile Phase: A: 20 mM Ammonium Acetate in Water
B: Methanol

Gradient:	Time (min)	% B
	0.00	10
	0.10	55
	4.50	99
	4.95	99
	5.00	10
	6.50	00

Injection: 10 μ L
Flow Rate: 0.6 mL/min
Temperature: 40 °C
Detection: SCIEX Triple Quad™ 5500 with a Turbo V™ source

 Gemini[®]



Mass Spectrometer & Source Gas Conditions

- SCIEX Triple Quad™5500 system with Turbo V™ source
- ESI probe in negative polarity mode
- Source parameters optimized using Compound Optimization (FIA) function in Analyst® software

Parameter	Value
Curtain Gas (CUR)	35 psi
IonSpray Voltage (IS)	-4500 V
Temperature (TEM)	600 °C
Nebulizer Gas (GS1)	50 psi
Heater Gas (GS2)	50 psi



Mass Spectrometer Parameters: Compound Specific

Compound	Q1	Q3	DP	CE
PFCAs				
PFBA	212.9	169	-25	-12
PFPeA	262.9	219	-20	-12
PFHxA	313	269	-25	-12
PFHpA	363	319	-25	-12
PFOA	413	369	-25	-14
PFNA	463	419	-25	-14
PFDA	513	469	-25	-16
PFuDA	563	519	-25	-18
PFDoA	613	569	-25	-18
PFTriDA	663	619	-25	-20
PFTeDA	713	669	-25	-22
PFHxDA	813	769	-25	-24
PFODA	913	869	-25	-26

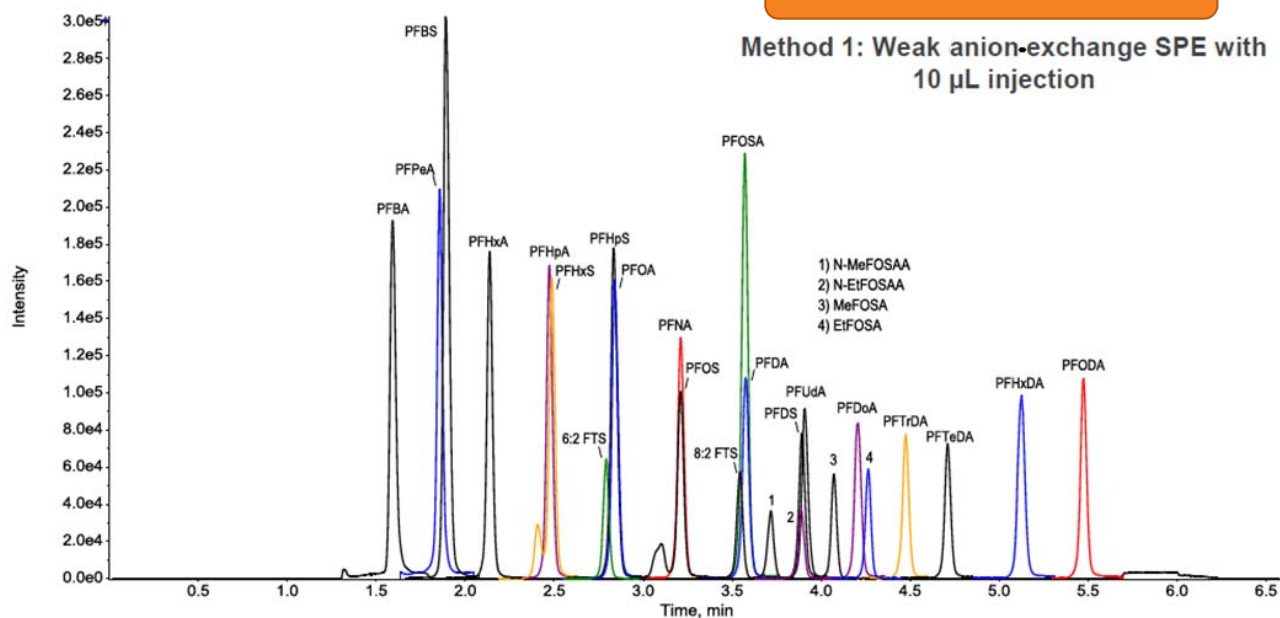
Compound	Q1	Q3	DP	CE
PFASs				
PFBS	298.9	80	-55	-58
PFHxS	399	80	-60	-74
PFHpS	449	80	-65	-88
PFOS	499	80	-65	-108
PFDS	599	80	-85	-118
Other PFASs				
6:2 FTS	427	407	-50	-32
8:2 FTS	527	507	-50	-40
PFOSA	498	78	-60	-85
MeFOSA	512	169	-75	-37
EtFOSA	526	169	-75	-37
N-MeFOSAA	570	419	-40	-36
N-EtFOSSA	584	419	-50	-36

- De-clustering Potential (DP) and Collision Energy (CE) optimized for each compound
- One MRM transition monitored each analyte and internal standard
- *Scheduled MRM™* algorithm used to maximize dwell times and optimize cycle time

Chromatogram: 10 pg on-column injection

6.5 min run time!

Method 1: Weak anion-exchange SPE with
10 μ L injection



Overlaid chromatogram of 1 μ g/L standard

Method Performance: Calibration Range, Sensitivity and Accuracy

Compound	Calibration Range (ng/L)	Linear Correlation (r^2)	S:N of 25 ng/L standard	Accuracy of 25 ng/L standard
PFCAs				
PFBA	25-20,000	0.997	108	104%
PFPeA	25-20,000	0.998	88	103%
PFHxA	25-20,000	0.998	104	93%
PFHpA	25-20,000	0.999	116	101%
PFOA	25-20,000	0.999	117	106%
PFNA	25-20,000	0.990	91	109%
PFDA	25-20,000	0.998	103	105%
PFUdA	25-20,000	0.995	84	101%
PFDaA	25-20,000	0.998	60	101%
PFTTrDA	25-20,000	0.998	32	104%
PFTeDA	25-20,000	0.994	15	107%
PFHxDA	25-20,000	0.999	21	103%
PFODA	25-20,000	0.999	33	102%

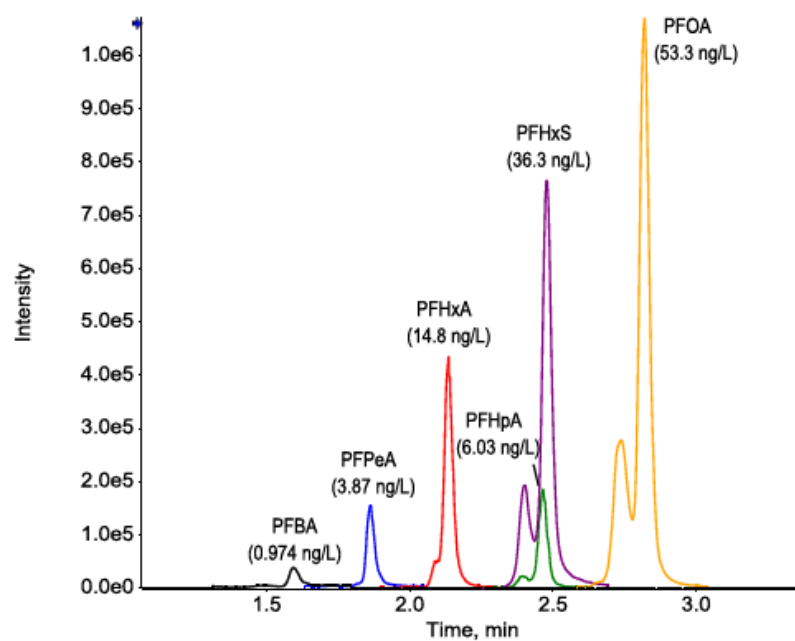
- 3 orders of linear dynamic range
- Excellent signal-to-noise and accuracy of lowest calibrator

Method Performance: Calibration Range, Sensitivity and Accuracy

Compound	Calibration Range (ng/L)	Linear Correlation (r^2)	S:N of 25 ng/L standard	Accuracy of 25 ng/L standard
PFSA s				
PFBS	25-20,000	0.995	31	92%
PFHxS	25-20,000	0.999	604	103%
PFHpS	25-20,000	0.997	103	105%
PFOS	25-20,000	0.995	312	105%
PFDS	25-20,000	0.998	88	102%
Other PFAS s				
6:2 FTS	25-20,000	0.991	100	98%
8:2 FTS	25-20,000	0.992	113	97%
PFOSA	25-20,000	0.997	118	104%
MeFOA	25-20,000	0.996	96	103%
EtFOA	25-20,000	0.994	90	101%
N-MeFOA	25-20,000	0.996	109	100%
N-EtFOA	25-20,000	0.994	61	103%

500x concentration factor:
25 ng/L injection std -> 0.05 ng/L in sample; method flexibility

SPE Method Application: Real-World Water Sample



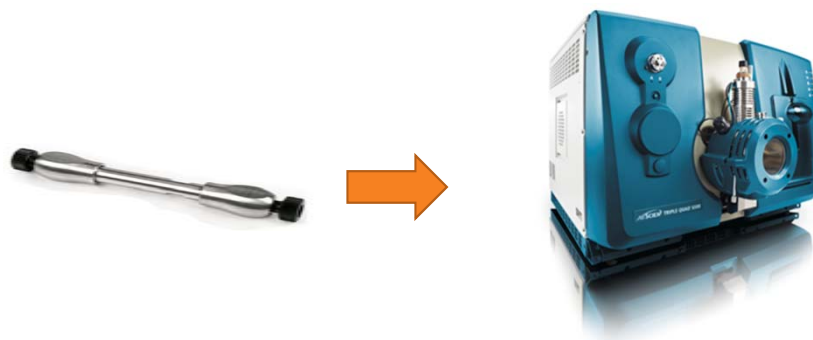
Overlaid chromatogram of PFAS in water sample

* Chromatographic separation of isomers

SPE and LC/MS/MS - Method Summary

- **Sample cleanup and concentration, compatible with DOD QSM 5.1**
- **Short 6.5 min LC method; separation of PFAS compounds and isomers *with HLPC***
 - Good peak shape for lower chain-length PFAS
- **Use of delay column to separate contamination from LC**
- **Method performance:**
 - 3 orders of linear dynamic range, +/- 10% accuracy and excellent S/N for lowest calibrator (25 ng/L), $r^2 > 0.990$
- **Applied to real-world water sample; PFAS detected at 0.97- 53.3 ng/L**
- **Can detect levels well below new EPA drinking water guidelines for PFOA and PFOS**

PFAS using HPLC and Direct Inject LC/MS/MS Case Study 2



Craig Butt, Ph.D.
Product Application Specialist



Large Volume Injection Method Overview

- **Advantage:** Minimal sample preparation, reduced contamination potential
- **Overview:**
 - Water samples diluted with methanol + surrogate standards
 - Direct injection of 950 µL onto analytical column
 - Longer and larger diameter column used to improve retention; resulting in longer runtime (17.5 min)

Large Volume Injection Method: Flowchart

1 mL water sample combined with
0.65 mL methanol + surrogate standards



950 μ L injection on PAL
HTC-xt autosampler

Method Overview

- Method not optimized for PFHxDA (C16) and PFODA (C18)
- Presence of 5 g/L Trizma is not compatible with large volume injection method due to ionization suppression (but is compatible with SPE method)
- Mass Spectrometer:
 - SCIEX Triple Quad™ 5500 system with Turbo V™ source; ESI negative mode
 - Identical source gas and compound-specific parameters as the SPE method



LC Conditions For: Large Volume Injection

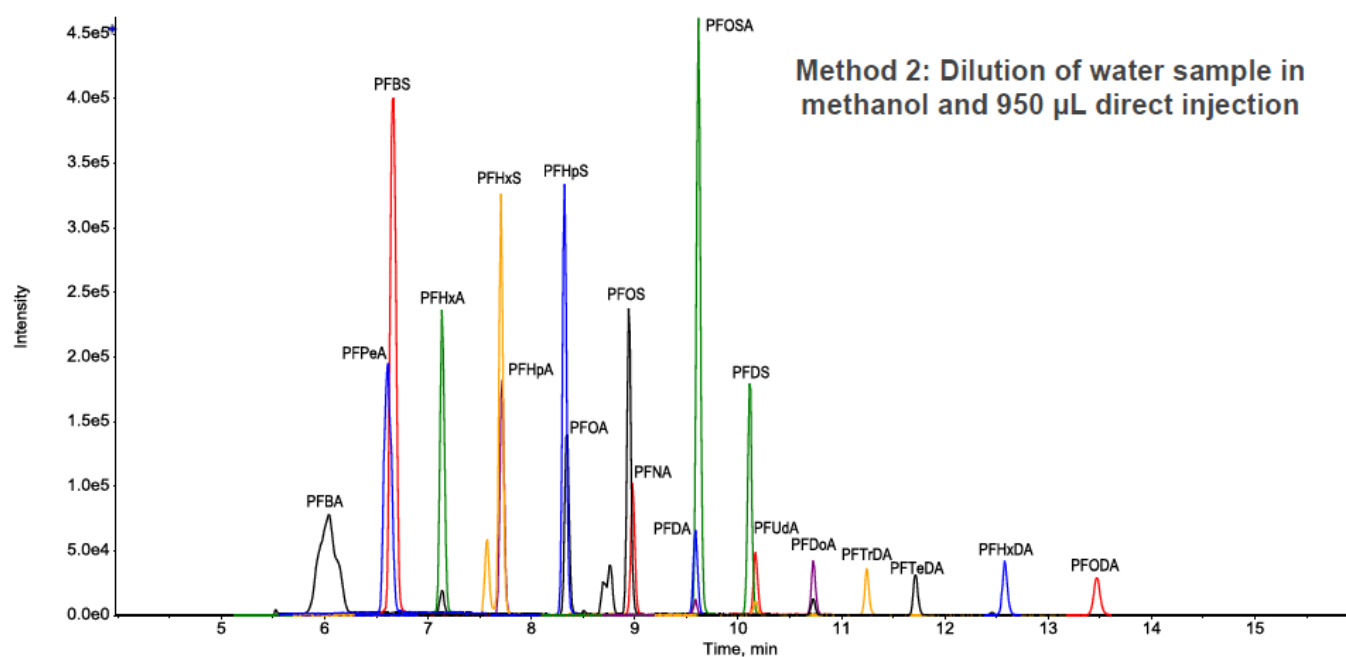
Column: Gemini® 3 µm C18
Dimensions: 100 x 3.0 mm
Part No.: 00D-4439-Y0
Mobile Phase: A: 20 mM Ammonium Acetate in Water
B: Methanol

Gradient:	Time (min)	% B
	0	10
	1.5	65
	8	95
	8.1	99
	12	99
	12.5	10

Injection: 950 µL
Flow Rate: 0.6 mL/min
Temperature: 40 °C
Detection: SCIEX Triple Quad™ 5500 with a Turbo V™ source



Chromatogram: Large Volume Injection Method

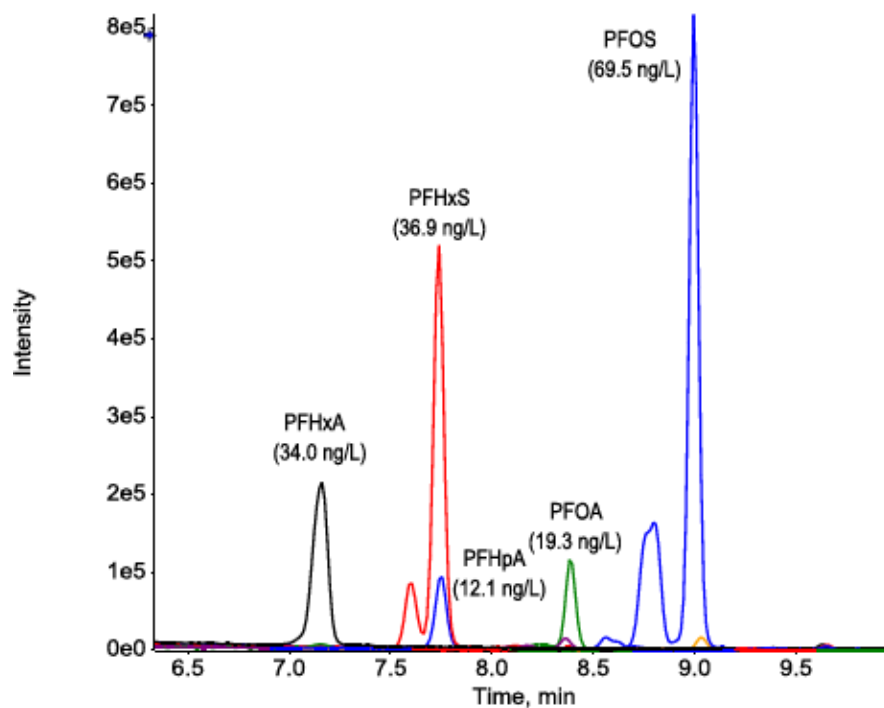


Overlaid chromatogram of 10 ng/L spike into groundwater

Method Performance: Calibration Range, Sensitivity and Accuracy

Compound	Calibration Range (ng/L)	Linear Correlation (r ²)	S:N of 1 ng/L standard	Accuracy of 1 ng/L standard
PFCAs				
PFBA	1-200	0.997	328	97%
PFPeA	1-200	0.999	137	101%
PFHxA	1-200	0.999	284	101%
PFHpA	1-200	0.993	267	96%
PFOA	1-200	0.999	113	99%
PFNA	1-200	0.999	137	101%
PFDA	1-200	0.997	176	96%
PFUdA	1-200	0.998	168	99%
PFDoA	1-200	0.994	127	94%
PFTTrDA	1-200	0.995	125	95%
PFTeDA	1-200	0.998	56	98%
PFSAs				
PFBS	2-200	0.994	1178	100%
PFHxS	1-200	0.998	229	96%
PFHpS	1-200	0.999	327	99%
PFOS	1-200	0.999	251	99%
PFDS	1-200	0.999	516	98%
PFOSA	1-100	0.997	1012	96%

Large Volume Injection Method Application



*Chromatographic separation of isomers

Overlaid chromatogram of PFAS in groundwater sample as analyzed by large volume injection

Large Volume Injection Method Summary

- **Minimal sample preparation; dilute with methanol + surrogate standard and inject (“dilute & shoot”)**
- **Method performance:**
 - 2 orders of linear dynamic range, +/- 10% accuracy and excellent S/N for lowest calibrator (25 ng/L), $r^2 > 0.990$
- **Excellent method robustness; precision (% CV) was <5% for 9 replicates (20 ng/L) over 1 week analysis**
- **Applied to real-world groundwater sample; PFAS detected at 12.1-69.5 ng/L**
- **Can detect levels below EPA drinking water guidelines (70 ng/L for PFOS and PFOA)**

PFAS using SPE or Direct Inject by UHPLC-MS/MS Case Study 3



LC-MS/MS

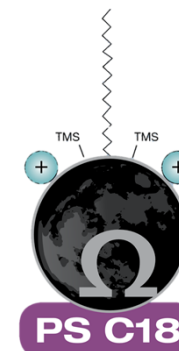
Agustin Pierri, Ph.D.
Technical Director

UHPLC Conditions For: Large Volume Injection and SPE

Column: Luna Omega 1.6 μ m PS C18
Dimensions: 50 x 2.1 mm
Part No.: 00B-4752-AN
Mobile Phase: A: 5 mM Ammonium Acetate in Water
B: Acetonitrile
Gradient:

Time (min)	B (%)
0.0	40
0.5	40
3.0	90
3.1	100
4.0	100

Injection: 1 μ L
Flow Rate: 0.55 mL/min
Temperature: 40 °C



Direct Injection Chromatogram

23 PFAS Analytes

6:2 FTS

EtFOSE

PFDS

PFNA

PFTeDA

8:2 FTS

MeFOSE

PFHpA

PFOA

PFTTrDA

EtFOSA

PFBA

PFHpS

PFOS

PFUdA

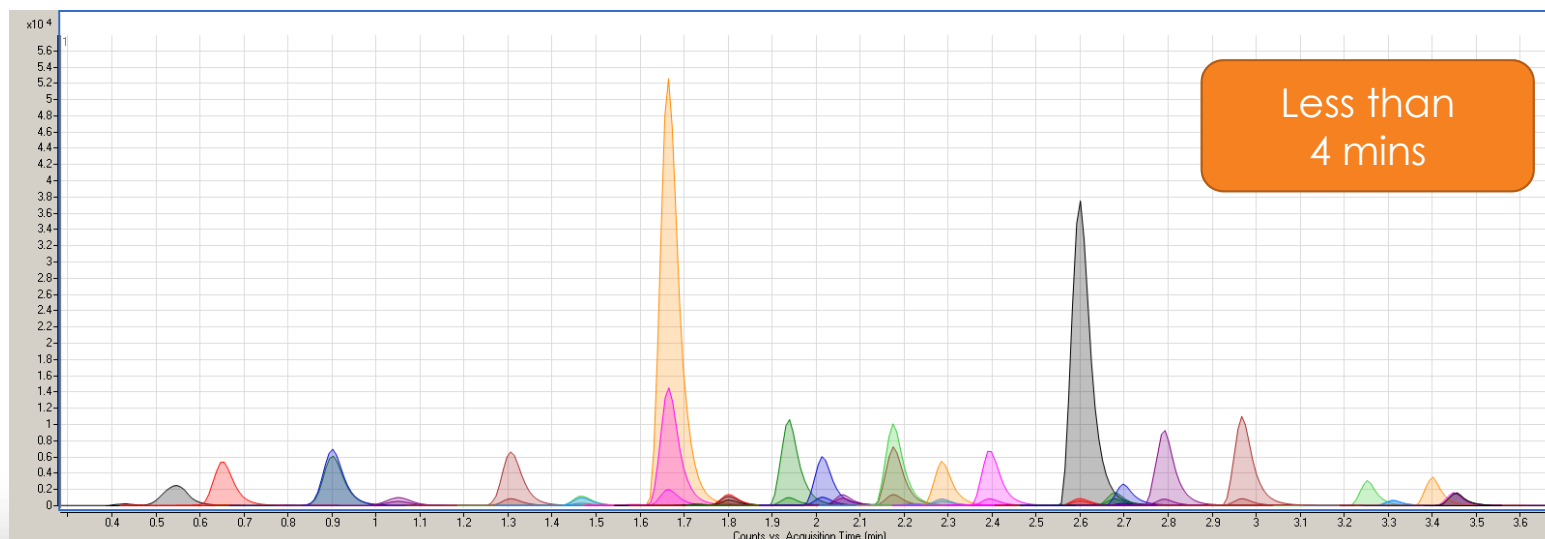
MeFOSA

PFDA

PFHxS

PFPeA

PFDoA



PFAS in Butter, Egg, and Milk QuEChERS, SPE and LC-MS/MS Case Study 3

Two-step extraction procedure- QuEChERS and Solid Phased Extraction (SPE)-



Dr. Agustin Pierri

W L
WECK LABORATORIES, INC.

QuEChERS Protocol

- Add 1.0 gram homogenized sample to 50 mL tube
- Add 10 mL H₂O and 10 mL MeCN
- Add 4 g MgSO₄ and 1 g NaCl
- Vortex 3 minutes, centrifuge 5 minutes
- Transfer 1 mL aliquot to tube with 150 mg MgSO₄ and 50 mg PSA
- Transfer Aliquot for LC-MS/MS analysis
- Analysis was done using on Luna Omega PS C18 50 x 2.1mm



QuEChERS and SPE Protocol

QuEChERS

- Add 1.0 grams homogenized sample to 50 mL tube
- Add 10 mL H₂O and 10 mL MeCN
- Add 4 g MgSO₄ and 1 g NaCl
- Vortex 3 minutes, centrifuge 5 minutes
- Transfer 1 mL aliquot to tube with 150 mg MgSO₄ and 50 mg PSA
- Transfer 500 µL, dilute to ~15 mL with H₂O for SPE

SPE

- Condition Strata-XAW 200 mg/3 mL SPE cartridge with 0.3% NH₄OH/MeCN
- Load diluted QuEChERS extract, wash with 5 mL H₂O
- Elute with 4mL 0.3% NH₄OH/MeCN
- Transfer 200 µL to conical vial for analysis
- Evaporate to 500 µL, transfer to conical vial for analysis
- Analysis was done using Luna Omega PS C18 50 x 2.1mm



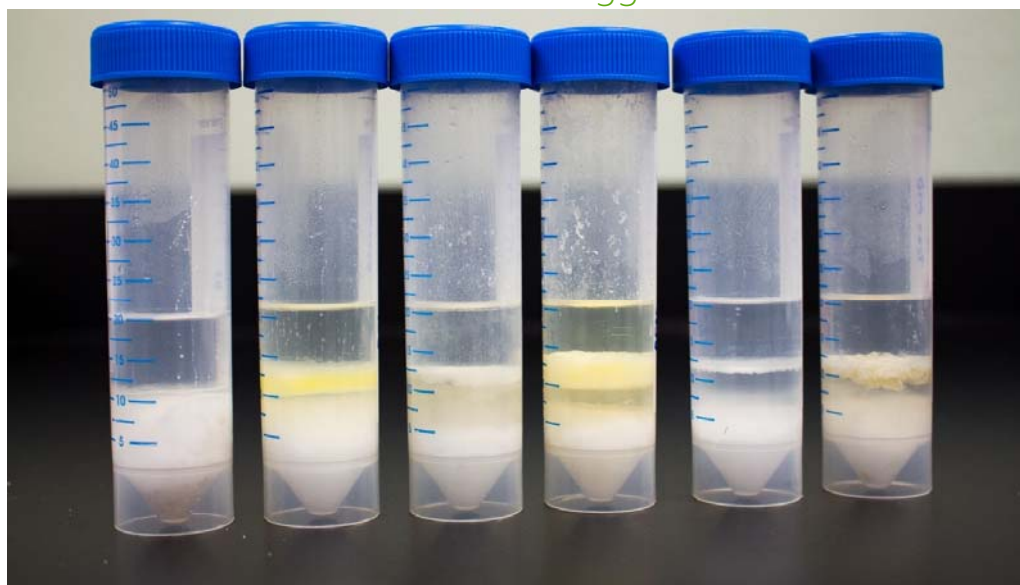
roQ QuEChERS Original Non-buffered



LC-MS/MS

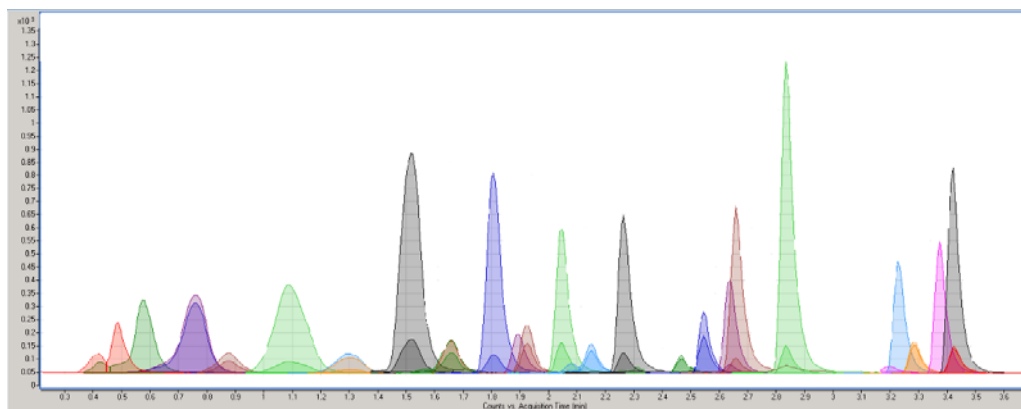
Extracts

Blank Butter Cheese Egg Milk Fish



PFAS in Butter using SPE and LC-MS/MS

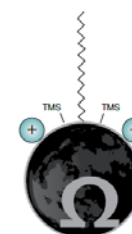
LUNA OMEGA PS C18



Column: Luna Omega 1.6µm PS C18
Dimensions: 100 x 2.1 mm
Part No.: 00D-4752-AN
Mobile Phase: A: 5 mM Ammonium Acetate in Water
 B: Acetonitrile
Gradient:

Time (min)	B (%)
0.0	40
0.5	40
3.0	90
3.1	100
4.0	100

Injection: 2 µL
Flow Rate: 0.55 mL/min
Temperature: 40 °C
Detection: MS/MS



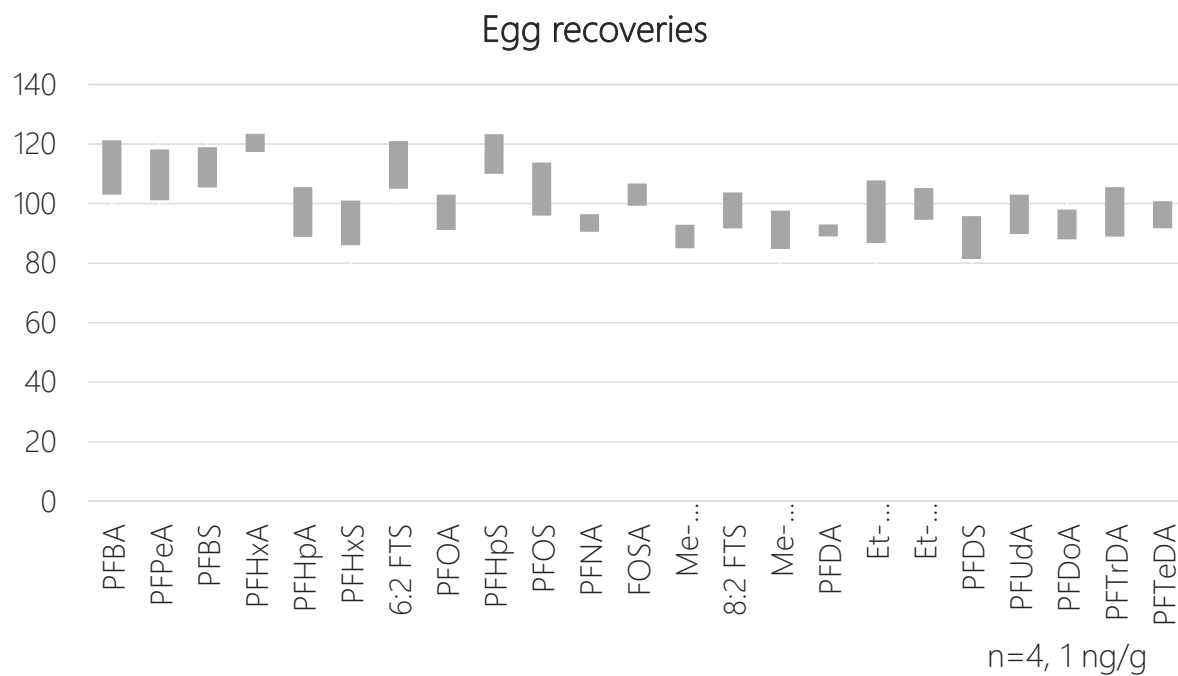
Analytes

- | | | |
|------------|-----------|------------|
| 1. 6:2 FTS | 11. PFDoA | 21. PFTeDA |
| 2. 8:2 FTS | 12. PFDS | 22. PFTrDA |
| 3. EtFOSA | 13. PFHpA | 23. PFUdA |
| 4. EtFOSE | 14. PFHpS | |
| 5. FOSA | 15. PFHxA | |
| 6. MeFOSA | 16. PFHxS | |
| 7. MeFOSE | 17. PFNA | |
| 8. PFBA | 18. PFOA | |
| 9. PFBS | 19. PFOS | |
| 10. PFDA | 20. PFPeA | |

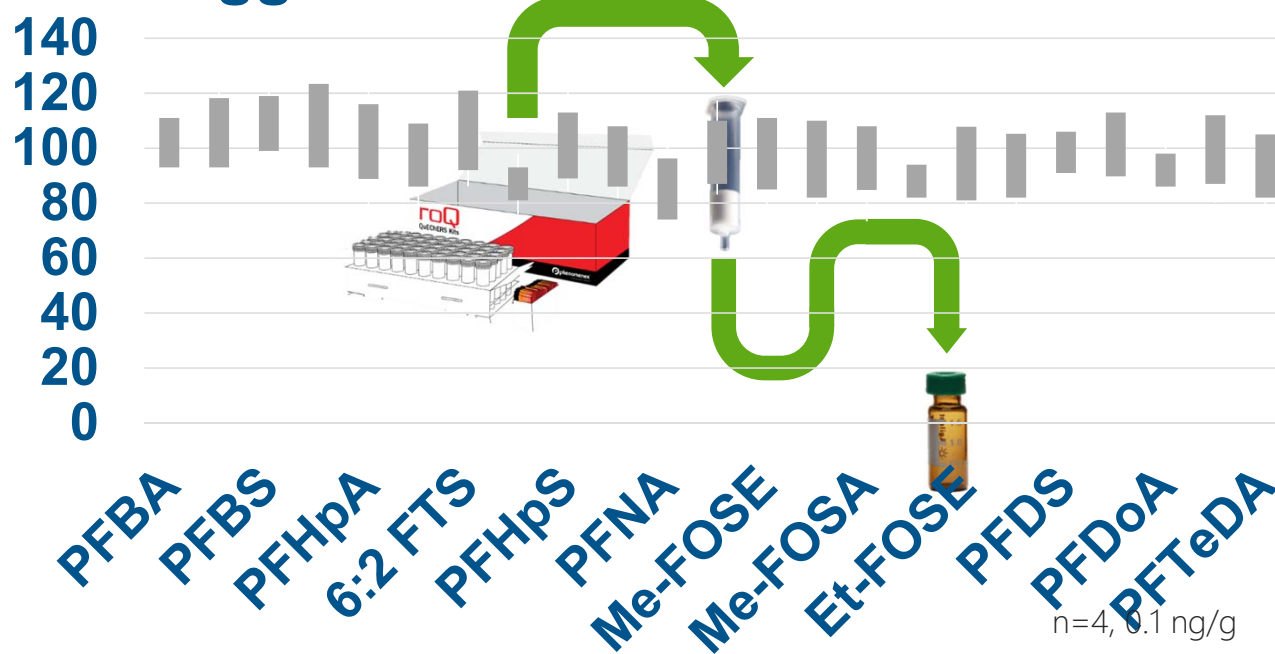
We would like to provide special thanks to Agustin Pierre from Weck Laboratories for contributing this application.



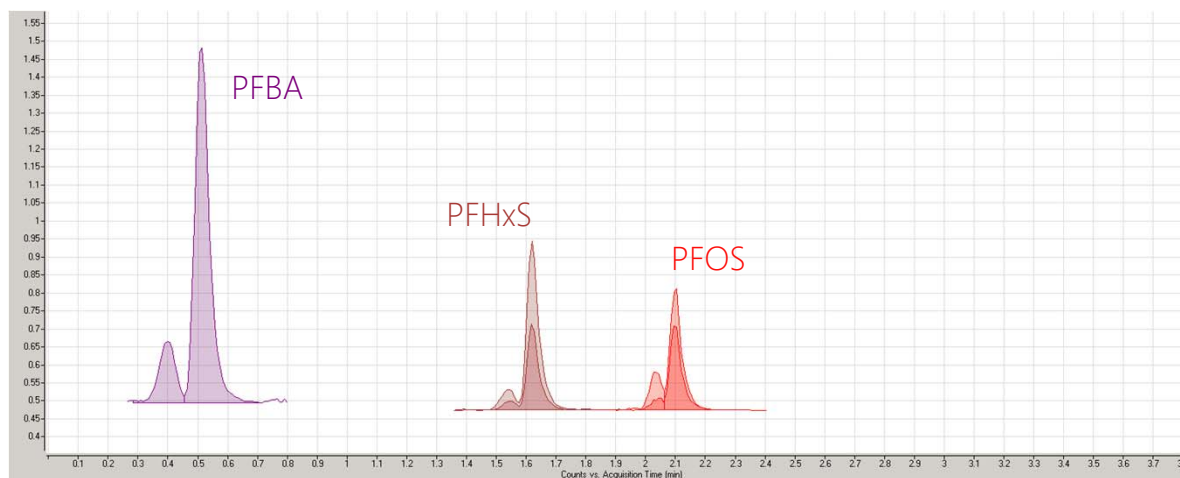
QuEChERS Recoveries, Egg



Egg recoveries: QuEChERS + SPE



Branched vs. linear



PFAS using Online SPE by HPLC-MS/MS Case Study 4



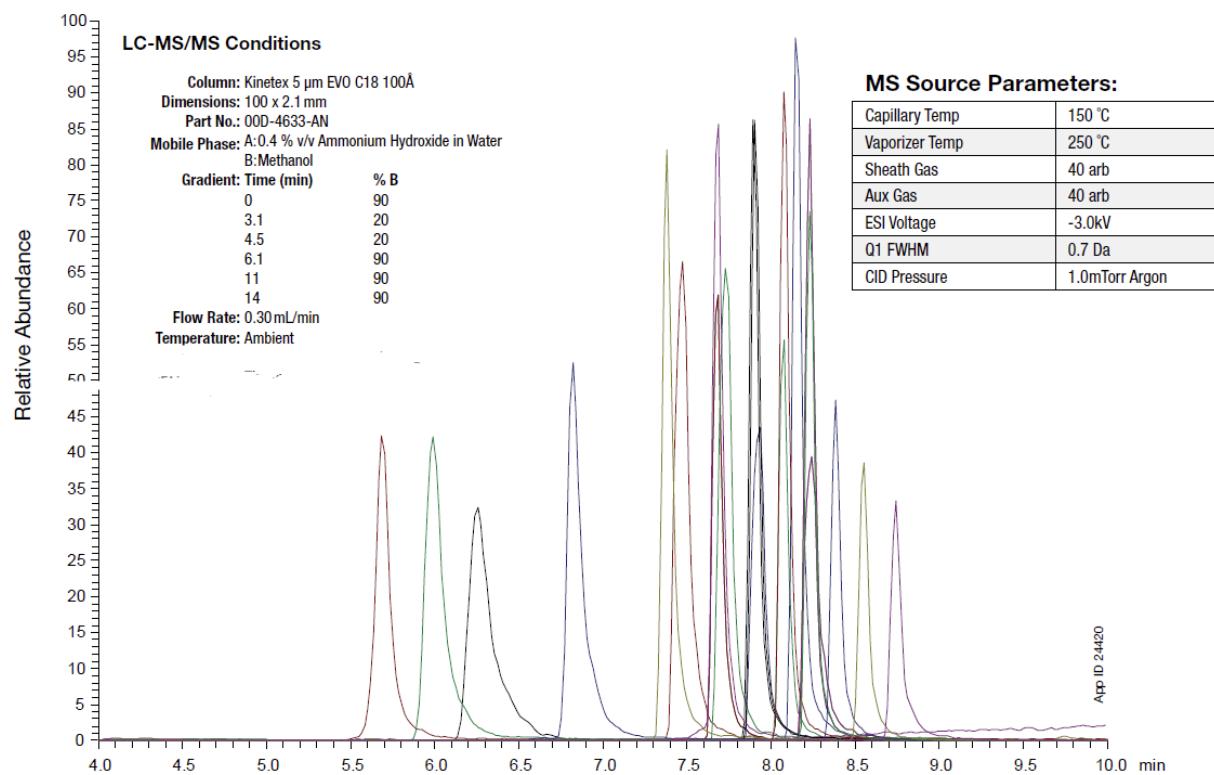
LC-MS/MS

David Schiessel



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The Standard of Excellence for Over 100 Years

Evaluation of Online SPE Sorbents for PFAS



Sample Prep Procedure

1. Samples are collected in polypropylene bottles and preserved with 0.5 g/L Trizma®.
2. A 10 mL aliquot is spiked with surrogates at a concentration of 50 ng/L.
3. If necessary, filter using a 10 mL syringe fitted to a 1.2 µm glass fiber syringe filter.
4. The filtered sample is spiked with internal standard at 50 ng/L.
5. The filtered sample is loaded and analyzed using a 5.0 mL injection volume.
6. The online SPE is completely automated; it includes a sample wash step (2.1 to 4.1 min) to wash Trizma preservative from the media.

LC Gradient (pump 1)

Time (min)	Water (%)	MeOH (%)	0.4% NH ₃ (%)
0.00	0	90	10
3.10	20	20	60
4.50	20	20	60
6.10	0	90	10
11.00	0	90	10
14.00	0	90	10

Note: to decrease PFOA contributed by the eluent system, MeOH is kept at 90 % while loading the online SPE with sample and subsequently brought down to 20 % 1 min prior to online SPE elution.

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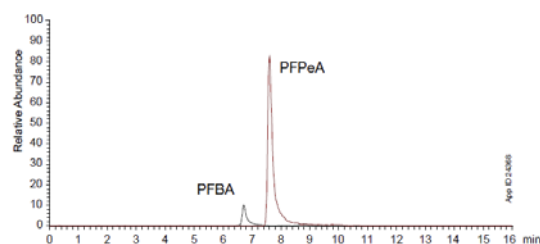
Online SPE Program (pump 2)

Time (min)	Water (%)	MeOH (%)	AcN (%)	Flow (mL/min)	Step
0.00	100	0	0	2.5	Load
2.00	100	0	0	2.5	Load
2.10	100	0	0	2.5	Wash
4.10	100	0	0	2.5	Wash
4.11	30	70	0	0	Idle
9.00	30	70	0	0	Idle
9.01	0	0	100	2.0	Wash
9.49	0	0	100	2.0	Wash
9.50	2.0	98	0	3.0	Wash
11.50	2.0	98	0	3.0	Wash
11.51	100	0	0	3.0	Equil
14.00	100	0	0	3.0	Equil



Evaluation of Online SPE Sorbents for PFAS

Online SPE using C18-E sorbent and 2.0 mM ammonium acetate mobile phase modifier on a Luna Omega C18 50 mm column.



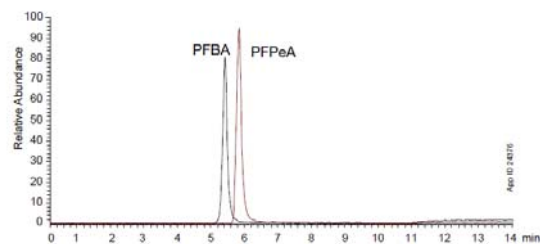
LC-MS/MS Conditions
 Column: Luna[®] Omega 1.6 μ m C18 100Å
 Dimensions: 50 x 2.1 mm
 Part No.: 008-4742-AN
 Mobile Phase: A: 2 mM Ammonium Acetate in Water
 B: Methanol

Gradient Time (min)	% B
0	40
4.1	40
6.1	90
13	90
13.01	40
16	40

 Injection: 5.0 μ L
 Flow Rate: 0.20 mL/min
 Temperature: 40 °C



Online SPE using Strata-X-AW sorbent and 0.4-0.8% ammonia mobile phase modifier on a Kinetex C18 EVO column (final conditions).



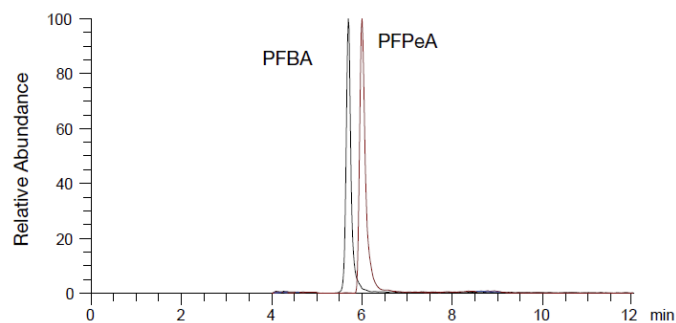
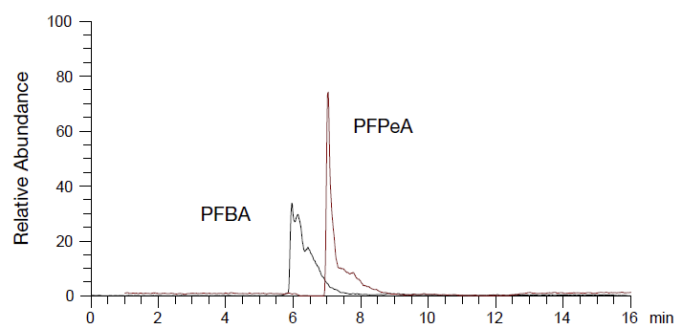
LC-MS/MS Conditions
 Column: Kinetex[®] 5 μ m EVO C18 100Å
 Dimensions: 100 x 2.1 mm
 Part No.: 000-4633-AN
 Mobile Phase: A: 0.4 % v/v Ammonium Hydroxide in Water
 B: Methanol

Gradient Time (min)	% B
0	90
3.1	20
4.5	20
6.1	90
11	90
14	90

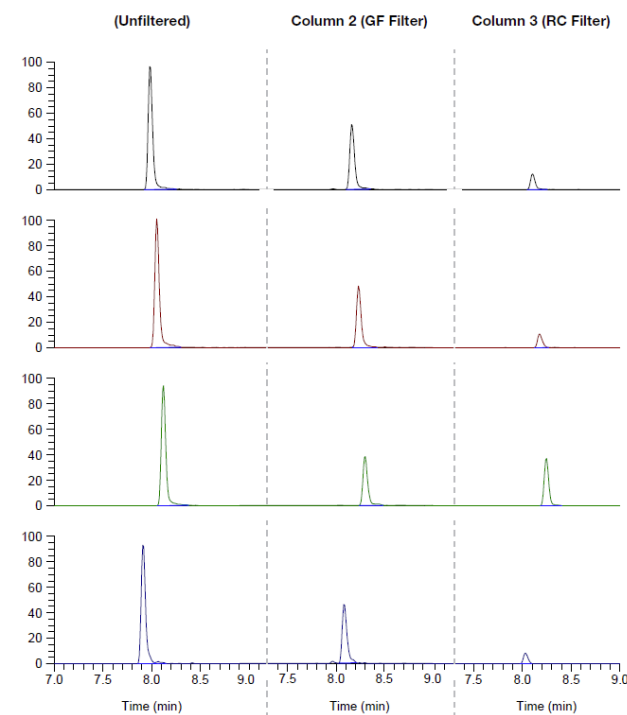
 Injection: 5.0 μ L
 Flow Rate: 0.30 mL/min
 Temperature: Ambient

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Elution strength of 0.04% NH_3 (top) and 0.24% NH_3 (bottom) illustrating more efficient elution of analytes (PFBA and PFPeA) with increased base concentration in the mobile phase.



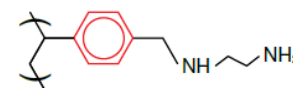
Effect of filtering on recovery of long chain PFCs.



Evaluation of Online SPE Sorbents for PFAS

Options	Column	Strata SPE Sorbent	Sample pH	SPE Conditioning pH	Eluent*	PFBA / PFPeA %	Shape
1	Kinetex EVO C18 5 µm 100 x 2.1mm	X-AW	Trizma (pH=7)	neutral	0.24-0.04 % NH ₃	100	excellent
2	Kinetex EVO C18 5 µm 50 x 2.1mm	X-AW	neutral	neutral	0.04 % NH ₃	106	very poor
3	Kinetex EVO C18 5 µm 50 x 2.1mm	X-AW	neutral	neutral	0.24-0.04 % NH ₃	76	OK
4	Kinetex EVO C18 5 µm 50 x 2.1mm	X-AW	acidic	neutral	0.02 % Formic Acid	13	OK
5	Luna Omega C18 1.6 µm 50 x 2.1mm	C18	neutral	neutral	2 mM NH ₄ OAc	<1	—
6	Luna Omega C18 1.6 µm 50 x 2.1mm	C18	acidic (pH=2)	acidic (pH=2)	0.02 % Formic Acid	22	very poor
7	Luna Omega C18 1.6 µm 50 x 2.1mm	C18	acidic (pH=2)	acidic (pH=2)	2 mM NH ₄ OAc	11	OK
8	Luna Omega C18 1.6 µm 50 x 2.1mm	C18	neutral	acidic (pH=2)	2 mM NH ₄ OAc	11	OK
9	Luna Omega C18 1.6 µm 50 x 2.1mm	X	neutral	neutral	2 mM NH ₄ OAc	5.9	poor
10	Luna Omega C18 1.6 µm 50 x 2.1mm	X	acidic	neutral	2 mM NH ₄ OAc	5.1	poor

* Note: All eluents used a gradient of increasing methanol for elution.



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PFAS in Sediment using QuEChERS

Case Study 5



Syljohn Estil and Eric Nelson

PFAS in Sediment using QuEChERS

Sample Preparation

QuEChERS Extraction Protocol

Step 1

Extraction

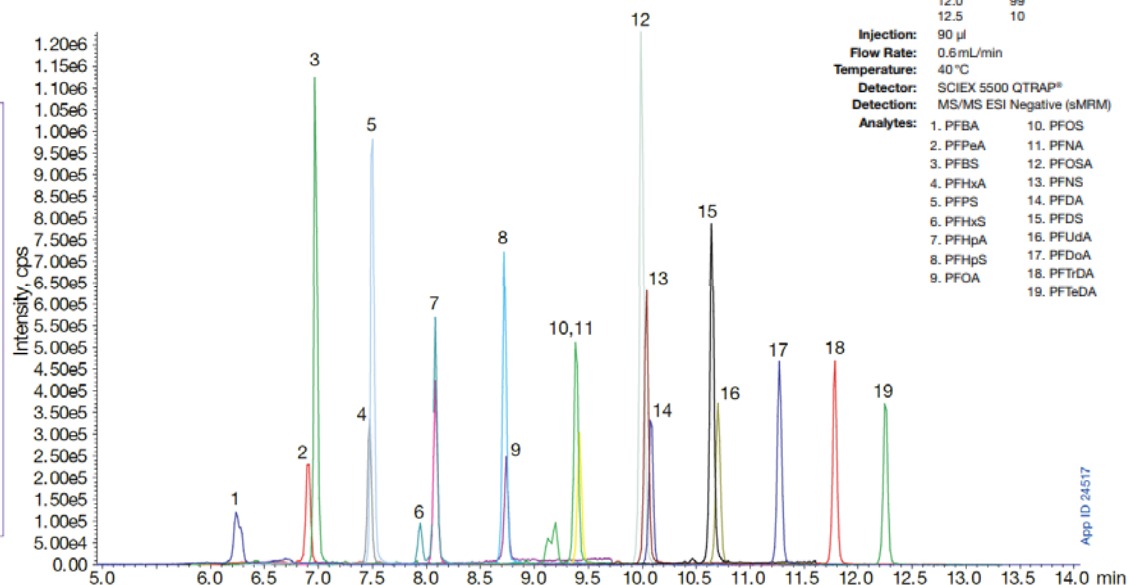
1. Weigh 2.0 g of dried sediment into a polypropylene container and spike with isotopically-labeled internal standards
2. Add 10mL deionized water and vortex. Add 10mL acidified acetonitrile (1% acetic acid) to the slurry and vortex
3. Add the extraction salts (1.5 g Sodium Acetate and 2 g MgSO_4) to the sample and vortex for 1 minute
4. Centrifuge the samples for 5 minutes at 4000 rpm
5. Place the samples in a rack and freeze at -20° for 30-60 minutes. This freezing step allows for easier extraction of the supernatant

Step 2

Clean-up

1. Transfer 8-9mL of the acetonitrile supernatant into a roQ QuEChERS PSA/C18 dSPE clean-up tube (Part no. KS0-8926) and vortex for one minute
2. Centrifuge the dSPE tubes for 10 minutes at 3000 rpm
3. Place an aliquot of the extract in a HPLC vial and dilute 1:1 with deionized water. The sample is now ready for analysis

Extracted ion chromatogram of sediments spiked with 1.0 ng/g of the target analytes



HPLC-MS/MS Conditions

Column:	Gemini [®] 3 μm C18
Dimensions:	100 x 3 mm
Part No.:	00D-4439-Y0
Inline Filter:	Phenomenex Krudkatcher™ Ultra
Delay Column:	Luna [®] 5 μm C18 (2) 30 x 2.0 mm
Part No.:	00A-4252-B0
Mobile Phase:	A: 20 mM Ammonium acetate in water B: Methanol
Gradient:	Time (min) % B
	0.0 10
	1.5 65
	8.0 95
	8.1 99
	12.0 99
	12.5 10
Injection:	90 μL
Flow Rate:	0.6 mL/min
Temperature:	40 $^\circ\text{C}$
Detector:	SCIEX 5500 QTRAP [®]
Detection:	MS/MS ESI Negative (sMRM)
Analyses:	1. PFBA 10. PFOS 2. PFPeA 11. PFNA 3. PFBS 12. PFOSA 4. PFHxA 13. PFNS 5. PFPS 14. PFDA 6. PFHxS 15. PFDS 7. PFHpA 16. PFUDA 8. PFHpS 17. PFDoA 9. PFOA 18. PFTDA 19. PFTeDA

Conclusion

Solid Phase Extraction or Large-Volume Injection are both suitable sample preparation techniques

Delay column to distinguish system related PFAS interferences

HPLC and UHPLC column chemistries suitable for the chromatographic range of polar acids through non-polar acids, esters, amides, and sulfonamides with selectivity of branched vs. linear isomers

SCIEX Triple Quad™ 5500 with Turbo V™ source

Detection limits at low ppt levels with the ability to detect below drinking water guidelines

Environmental Support and Resources

SPE Method Development Tool

1,000s of Applications

On-Site Lab Demos

Environmental Edge Newsletter

Technical Notes

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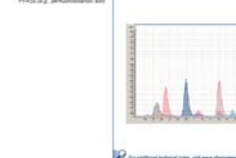


Quantitation of PFASs in Water Samples using LC/MS/MS:
Large-Volume Direct Injection and Solid Phase Extraction

Simon Roberts¹, KC Hyland¹, Craig Bull¹, Scott Kasper², Eric Redman², and Christopher Barker²
¹SCIEX (Redwood City, CA), ²SCIEX (Pharmingen, MA), ³Pharmingen (Torrance, CA), ⁴TestAmerica Laboratories (Torrance, CA)

Overview:
This application note presents two methods for the quantitation of per- and polyfluorinated alkyl substances (PFASs) in water samples. The first method involves sample preparation and injection into the column using a 10-minute HPLC gradient. The second method involves sample preparation and injection into the column using a 10-minute HPLC gradient. Both methods involve sample preparation and injection into the column using a 10-minute HPLC gradient.

Introduction:
PFASs are a group of chemicals that have been found in a wide variety of products and environments. They are known for their resistance to degradation and their ability to bioaccumulate. This application note presents two methods for the quantitation of PFASs in water samples. The first method involves sample preparation and injection into the column using a 10-minute HPLC gradient. The second method involves sample preparation and injection into the column using a 10-minute HPLC gradient.



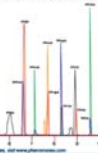
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APPLICATION
Rapid Analysis of 23 Per- and Poly-Fluorinated Alkyl Substances (PFASs) by
UHPLC-MS/MS using Luna[®] Omega 1.6µm PS C18

Simon Roberts¹, KC Hyland¹, Craig Bull¹, Scott Kasper², Eric Redman², and Christopher Barker²
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...breaking with tradition™

APPLICATION
PFAS Analysis in Water Samples using LC/MS/MS Large-Volume Direct Injection

Simon Roberts¹, KC Hyland¹, Craig Bull¹, Scott Kasper², Eric Redman², and Christopher Barker²
¹SCIEX (Redwood City, CA), ²SCIEX (Pharmingen, MA), ³Pharmingen (Torrance, CA), ⁴TestAmerica Laboratories (Torrance, CA)

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Thank You Acknowledgements

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